

*Microbial and Biochemical Mechanisms Underlying Greenhouse Gas
Mitigation in Agriculture*

by

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B.S., Jeonbuk National University, 2017

M.S., Korea Advanced Institute of Science and Technology (KAIST), 2019

AN ABSTRACT OF A DISSERTATION

submitted in partial fulfillment of the requirements for the degree

DOCTOR OF PHILOSOPHY

Department of Civil Engineering
Carl R. Ice College of Engineering

KANSAS STATE UNIVERSITY
Manhattan, Kansas

2026

Abstract

Greenhouse gases (GHGs) such as carbon dioxide (CO_2), nitrous oxide (N_2O), and methane (CH_4) play a crucial role in regulating Earth's climate by trapping heat in the atmosphere. While carbon dioxide is the most abundant anthropogenic GHG, N_2O and CH_4 contribute disproportionately to global warming due to their much higher global warming potential, approximately 298 and 25 times greater than CO_2 , respectively. N_2O and CH_4 are particularly concerned due to their persistence in the atmosphere. Natural processes and human activities both contribute to the production of N_2O and CH_4 . Agricultural activities are the predominant source, responsible for over two-thirds of global anthropogenic N_2O emissions and one-third of total CH_4 emissions. In these agricultural systems, microbial processes are fundamental to the production and mitigation of both N_2O and CH_4 . To elucidate microbial mechanisms regulating GHG fluxes in agricultural systems, this dissertation presents two complementary studies: one focusing on mitigating N_2O emissions during forage conservation, and the other examining the role of biological nitrification inhibition (BNI) in controlling CH_4 oxidation. Building on the recognition that anaerobic forage conservation may serve as an unrecognized source of N_2O , I conducted an extensive investigation using three major silage crops in the U.S.—maize, alfalfa, and sorghum. The total N_2O emission from these silage processes was quantified and emission potential was calculated with USDA National Agriculture Statistics Services annual silage production report, revealing that silage ranks as the third largest source of anthropogenic N_2O emissions in the U.S. Through chemical and molecular analyses, denitrifying bacteria were identified as the primary contributors to these emissions. Subsequently, a chemical combination was developed using chlorate—a specific inhibitor of nitrate reductase—mitigating N_2O emissions from silage up to 99%, offering a practical solution to significantly reduce GHG emissions in agriculture.

The second part of this dissertation examines a novel phenomenon with significant environmental implications. Plant-microbe interactions in the rhizosphere strongly influence nitrogen cycling and GHG emissions. BNI through which plants release natural compounds that suppress nitrification, offers a sustainable alternative to synthetic inhibitors. However, the broader impacts of BNI on other microbial processes remain unclear. In particular, CH_4 oxidation by methanotrophs—the only biological sink for atmospheric CH_4 —may be affected because the

key enzymes involved in nitrification and methanotrophy share similar catalytic mechanisms. This effect could be significant in paddy environments like wetlands and rice fields, which are major contributors to global GHG emissions. To investigate this interaction, a multi-scale experimental approach was undertaken. Soil column experiments with BNI-producing plants demonstrated that BNI compounds reduced CH₄ oxidation activity and altered methanotrophic community structures. Pure culture experiments confirmed the direct inhibitory effects of specific BNI compounds on methanotrophs. These findings provide the first multi-lines of evidence that BNI can adversely affect methanotrophy, suggesting that while BNI reduces nitrification and associated N₂O emissions, it may unintentionally suppress CH₄ oxidation, potentially increasing CH₄ emissions. This novel discovery underscores the need for a balanced assessment of BNI's environmental impacts. Understanding BNI's dual effects is essential for developing sustainable agricultural practices that mitigate GHG emissions without unintended consequences. Collectively, these two studies advance our understanding of how nitrogen and carbon management practices influence GHG emissions in agricultural systems, providing valuable insights for developing more sustainable and climate-resilient agricultural strategies.

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Chapter 1 - Background and literature review

1.1. Introduction

Although global climate change is widely recognized as a long-term threat to humanity and the environment, atmospheric concentrations of GHGs continue to rise, primarily due to ongoing human activities (IPCC Core Writing Team, 2014; Montzka et al., 2011; D. Reay et al., 2007). GHGs regulate the Earth's energy balance by absorbing and re-emitting heat from the surface, effectively forming an insulating layer that elevates surface temperatures. GHGs regulate the Earth's energy balance by absorbing and re-emitting heat from the surface, effectively forming an insulating layer that elevates surface temperatures (IPCC Core Writing Team, 2014; D. Reay et al., 2007; U.S. EPA, 2018). While CO₂ is the dominant driver of anthropogenic warming, other gases—particularly N₂O and CH₄—also contribute substantially due to their respective warming potentials, 298 and 25 times that of CO₂, respectively. Among various sectors, agriculture is a major source of these gases, accounting for over two-thirds of global N₂O and about 40% of CH₄ emissions (Davidson & Kanter, 2014a; D. S. Reay et al., 2012a; USEPA, 2019). Most agricultural N₂O emissions originate from nitrogen fertilizer use, which stimulates microbial processes such as nitrification and denitrification—key components of the respective cycle that convert ammonia to nitrate and nitrate to nitrogen gas, respectively, with N₂O produced as an intermediate (Hu et al., 2015a; Klimasmith & Kent, 2022a).

In addition to soil-based emissions, recent studies have identified forage conservation—particularly ensiling and storage—as an emerging but overlooked source of N₂O. These emissions result from microbial transformations of nitrogen compounds under anaerobic and partially aerobic conditions typical of silage fermentation. However, N₂O released during forage conservation is not currently included in the U.S. Environmental Protection Agency's GHG

inventory, suggesting a potential underestimation of agricultural emissions. Quantifying these emissions and elucidating their microbial drivers are critical steps toward developing mitigation strategies and improving national GHG inventories. Beyond mitigation during storage, reducing N₂O production at its microbial source—through strategies such as biological nitrification inhibition (BNI) – offers a complementary approach to minimizing nitrogen losses and enhancing agricultural sustainability.

Similarly, CH₄ emissions in agriculture are primarily due to enteric fermentation in ruminant livestock and the anaerobic decomposition of organic matter in rice paddies, wetland and manure storage facilities. In these oxygen-poor environments, methanogenic archaea produce CH₄ as a metabolic product. Conversely, CH₄ oxidation by methanotrophs—bacteria that consume CH₄ in aerobic conditions—can mitigate CH₄ emissions by converting it to CO₂, thereby playing a vital role in the carbon cycle (Colin Murrell & Jetten, 2009; Conrad et al., n.d.; Wahlen M, 1993). Additionally, the implementation of BNIs presents a promising strategy for mitigating N₂O emissions by suppressing the activity of nitrifying microbes in the soil (O’Sullivan et al., 2016b; Subbarao et al., 2007a, 2009a). However, there is emerging evidence that BNIs may also affect methanotrophic communities (Holmes et al., 2006; Lieberman & Rosenzweig, 2004; Sayavedra-Soto et al., 2011). Since methanotrophs are essential for oxidizing CH₄ and reducing its emission into the atmosphere, understanding the potential impacts of BNIs on these organisms is crucial. A comprehensive understanding of the microbial processes driving N₂O and CH₄ emissions in agricultural systems is essential for developing effective mitigation strategies. By exploring how forage conservation affects N₂O emissions and investigating the potential influence of BNIs on methanotrophs, we can identify agricultural

practices that reduce GHG emissions while maintaining or enhancing productivity. Such insights are vital for achieving global climate objectives and promoting sustainable agriculture.

1.2. Forage conservation

Forages, which are plants or parts of plants consumed by herbivorous animals, are essential for livestock nutrition and farming efficiency. Forage conservation is critical for feeding livestock during periods of limited pasture growth or inadequate pasture conditions, ensuring a consistent food supply throughout the year (Finch et al., 2014a; Owens, 2003a). Historically, throughout the world, forage conservation is a crucial element for productive and efficient livestock farming. Forages consist of highly digestible cell content—proteins, fats, and soluble carbohydrates—and less digestible cell walls composed of cellulose, hemicellulose, and lignin (Barnes et al., 2003). The primary methods of forage conservation are haymaking and silage production. Hay is produced by drying forage to a moisture content below 20% (typically 12–20% w/w) to inhibit microbial growth, and it is stored under aerobic conditions (Barnes et al., 2003). Silage, on the other hand, is prepared with higher moisture content (50~70%, w/w) and stored under strict anaerobic conditions to promote fermentation. In silage, indigenous or exogenous lactic acid bacteria (LAB) convert soluble carbohydrates to organic acids, mainly lactic acid, which act as a natural preservative to inhibit undesirable microorganisms (Müller et al., 2008a; Soundharrajan et al., 2017a). Therefore, commercial inoculants containing LAB and other anaerobic bacteria (such as *Lactobacillus buchneri*) are added in practice (Muck et al., 2018a). Silage production is a significant segment of the global livestock industry. In 2022, the United States alone harvested approximately 162.3 million metric tons of crops for silage (USDA, 2023). The global market for silage inoculants was valued at \$503 million in 2021 and is projected to grow to \$630 million by 2028, reflecting an annual growth rate of 3.8%. Despite its high production volume and

widespread use, the forage conservation process has been relatively understudied as a potential source of GHGs (Grossi et al., 2019). However, several researchers have reported the detection of N₂O emissions from silages. For instance, N₂O production was observed from corn silages stored in open bunker silos, where ambient N₂O concentrations were orders of magnitude higher than rural background levels (Maw et al., 2002). A few other studies reported N₂O production from silage (Bueno et al., 2020a; Krommweh et al., 2020a), but a comprehensive quantitative assessment has not been performed.

1.3. Nitrogen cycle and N₂O emissions

N₂O is the third-largest contributor of GHG emissions to the atmosphere after CO₂ and CH₄ (USEPA, 2023). N₂O is also the dominant ozone-depleting substance in the stratosphere (Ravishankara et al., 2009a). N₂O is produced by both natural and anthropogenic sources, but the emissions from human activities have increased 20% since the pre-industrial era, and its concentration has increased 15% since 1750 (USEPA, 2023). Approximately 40% of N₂O emissions come from human activities, and of those, the majority are from agricultural practices. In the US, about 75% of all anthropogenic N₂O emissions are attributed to agriculture (USEPA, 2023). The mechanism of N₂O production is not completely understood, but microbial production of N₂O is generally considered to be the main driver of N₂O emissions (Hu et al., 2015; Klimasmit & Kent, 2022b; Kuypers et al., 2018a; a. J. Thomson et al., 2012; Tian, Xu, Canadell, Thompson, Yao, et al., 2020b). Although nearly all microorganisms involved in biogeochemical nitrogen cycling can potentially produce N₂O, there are several key microbial pathways contributing the most to N₂O production including heterotrophic denitrification, ammonia oxidation, and nitrifier denitrification (Butterbach-Bahl et al., 2013; Kuypers et al., 2018a). Heterotrophic denitrification is a multistep respiratory process that reduces oxidized

mineral forms of nitrogen, i.e., NO_3^- and nitrite (NO_2^-) to the gaseous nitric oxide (NO), N_2O , and dinitrogen (N_2). The reaction typically occurs under anaerobic conditions, although a novel group of aerobic denitrifying bacteria has recently been discovered (J. Yang et al., 2020a). On the other hand, ammonia oxidation and nitrifier denitrification occur under aerobic conditions. Nitrification is a two-step process where ammonia (NH_3) is oxidized to NO_2^- by ammonia oxidizing bacteria (AOB) and archaea (AOA), and subsequently to NO_3^- by NO_2^- oxidizing bacteria. N_2O is known to be produced indirectly via chemical decomposition of ammonia oxidation intermediates or end-product (hydroxylamine, nitroxyl hydride, or NO_2^-) (S. Liu et al., 2017a). In addition, some AOB can directly produce N_2O via nitrifier denitrification where NH_3 is oxidized to NO_2^- followed by the reduction of NO_2^- to NO and N_2O (Kozłowski et al., 2016a; Wrage, Velthof, van Beusichem, et al., 2001; Wrage-Mönnig et al., 2018a).

1.4. BNI and its inhibition mechanisms

Globally, about 50-70% of the nitrogen fertilizer applied to cropping systems is lost to the environment mainly due to microbial nitrification (Cassman et al., 2002; Ladha et al., 2005). Nitrification, a key microbial process in the global nitrogen-cycle, transforming immobile ammonium (NH_4^+) to highly mobile NO_3^- , is known to enhance losses of fertilizer nitrogen by leaching and denitrification (Coskun et al., 2017). Nitrification has been known to be a two-step process where NH_4^+ is first oxidized to NO_2^- by AOB and/or AOA, and subsequently to NO_3^- by NO_2^- -oxidizing bacteria, although recent studies demonstrated that *Nitrospira* spp. can carry out both reactions (Daims et al., 2015; van Kessel et al., 2015). Further, nitrification is the primary process of N_2O emissions in soils (R. Liu et al., 2016), which is the third most important GHG next to CO_2 and CH_4 .

For these reasons, there have been attempts to mitigate nitrification in agriculture. Since the mid-1950s, studies on nitrification inhibitors have been conducted, and hundreds of synthetic nitrification inhibitors (SNIs) have been identified. However, these artificially synthesized nitrification inhibitors are not widely applied in practice, because (1) their biological stability is limited leading to shorter effective time of nitrification inhibition, (2) some of these artificially synthesized nitrification inhibitors have toxicity and adversely affect beneficial soil microorganisms, resulting in their limited usability, and (3) artificially synthesized nitrification inhibitors are in general expensive (Coskun et al., 2017; Subbarao, Ito, et al., 2006). On the other hand, despite many early studies, it has recently become apparent that certain plant species are capable of suppressing microbial nitrification by releasing inhibitory phytochemicals from roots, which is termed BNI (Subbarao et al., 2009a; Sun et al., 2016; Tesfamariam et al., 2014). This phenomenon has emerged as a promising new area of research not only because of its potential to improve NUE in cropping systems (Coskun et al., 2017; Subbarao et al., 2013), but also its implications for agricultural soil management, i.e., reduced N₂O emissions (Byrnes et al., 2017; Peters et al., 2013; Subbarao et al., 2017). For example, N₂O emission was also suppressed by 90% in the field of high-BNI *Brachiaria* pasture compared with soybean (*Glycine max*), which lacks BNI capacity (Subbarao et al., 2009a). Since the first report of BNI in *Brachiaria humidicola* (*B.humidicola*), BNIs have been discovered in several crops, including sorghum (*Sorghum biocolor* L.) and wheat (*Triticum aestivum* L.) (O'Sullivan et al., 2016a; Zakir et al., 2008a). More recently, one of the major cereal crop rices (*Oryza sativa* L.) exudate was reported as crops which can produce detectable BNI activity (Pariasca Tanaka et al., 2010).

The first step of nitrification, the oxidation NH₃ to NO₂⁻ via hydroxylamine (NH₂OH), is catalyzed by two enzymes: ammonia monooxygenase (AMO) and hydroxylamine

oxidoreductase (HAO). Many studies measure BNI activity using nitrification inhibition assay (Subbarao, Ishikawa, et al., 2006) which measures NO_2^- concentration to identify the activity of several BNI exuded from roots with *Nitrosomonas* spp. in pure culture. The mode of inhibitory action of the BNIs varies for each species. For example, recent research has demonstrated that methyl 3-(4-hydroxyphenyl) propionate (MHPP) in sorghum and 1,9 decanediol (1,9-D) in rice specifically block the AMO step in bacterial ammonia oxidation (Coskun et al., 2017; Pariasca Tanaka et al., 2010; Zakir et al., 2008a). Also, Brachialactone, which is the exudate of *B. humidicola*, inhibited both AMO and HAO but had a greater inhibitory effect on AMO than the HAO (Subbarao et al., 2007b).

1.4. Methanotrophs and possibility of potential inhibition from BNI

Methanotrophs are specialized microbes capable of deriving all their carbon and energy from CH_4 , the second most abundant GHG after CO_2 (Core Writing Team, 2014; Denman et al.,). The most widely studied methanotrophs belong to the classes Gammaproteobacteria and Alphaproteobacteria, and they have historically been divided into type I and type II methanotrophs based on physiological characteristics (Dedysh & Dunfield, 2011; Knief, 2015a). All aerobic methanotrophs initiate CH_4 oxidation using methane monooxygenase (MMO), which converts CH_4 to methanol. This reaction is followed by further oxidation of methanol to formaldehyde by methanol dehydrogenases (Keltjens et al., 2014). The MMO is catalyzed by two forms, cytoplasmic or soluble form of methane monooxygenase (sMMO) and membrane-bound or particulate methane monooxygenase (pMMO; Semrau et al., 2013, 2020). All pMMO-containing methanotrophs use copper as the catalytic metal in the first step of CH_4 oxidation to methanol, and? copper suppresses the expression of the sMMO (Knapp et al., 2007; Trotsenko & Murrell, 2008) which require iron as they belong to a diverse enzymatic family of di-iron

carboxylate enzymes (Crombie & Murrell, 2014; Nichol et al., 2019). Most extant methanotrophs, including those that have the ability to synthesize sMMO, also express pMMO and such expression and activity increases with increasing copper availability (Choi et al., 2003; Lontoh & Semrau, 1998). These two distinct MMOs have different kinetics for CH₄ oxidation, with sMMO having a much higher turnover, but lower affinity for CH₄ than pMMO (Lee et al., 2006). As a result, the type of MMO expressed can have a significant impact on the overall CH₄ consumption rate (Semrau et al., 2013). Interestingly, pMMO, which catalyzes the initial oxidation of CH₄ to methanol, is evolutionarily related to AMO found in AOB. Both pMMO and AMO are membrane-associated enzymes and share striking similarities in their genetic encoding (*pmoCAB* and *amoCAB*, respectively), structural configuration ($\alpha_3\beta_3\gamma_3$), metal requirements (copper in pMMO and possibly iron in AMO), substrate ranges, and inhibition patterns (Arp et al., 2007; Hakemian & Rosenzweig, 2007). These similarities suggest that certain BNIs, known to inhibit AMO, may also inhibit MMO, particularly pMMO (Holmes et al., 1995). Such inhibition could reduce the efficiency of CH₄ oxidation in methanotrophs, potentially increasing CH₄ release into the atmosphere. The evolutionary relationship between AMO and pMMO underscores the complex interactions between CH₄ and N₂ cycling and highlights the potential impacts of BNI compounds on atmospheric CH₄ dynamics.

Chapter 2 - Hypotheses and research objectives

The goal of the proposed research is to provide a comprehensive understanding of the impact of nitrogen and carbon management practices on GHG emissions in agricultural systems, thereby offering valuable insights for developing strategies that promote more sustainable and climate-resilient agriculture.

2.1. The role of microbial diversity in controlling greenhouse gas emissions from conserved forages

2.1.1 Hypotheses

2.1.1.1 Evaluation and optimization of lysis methods for microbial DNA extraction from epiphytic phyllosphere samples

This study hypothesizes that DNA extraction methods employing enzymatic lysis will yield higher total DNA concentrations in silage-derived phyllosphere samples compared to methods based primarily on mechanical disruption, due to more extensive lysis of both microbial and plant-derived cells. However, mechanical lysis approaches, such as bead-beating, are expected to provide greater specificity toward microbial cells, resulting in higher bacterial 16S rRNA gene copy numbers and reduced co-extraction of plant DNA and inhibitory compounds.

It is further hypothesized that optimizing the balance between enzymatic and mechanical disruption will improve DNA quality, as reflected by DNA integrity and downstream sequencing performance. Additionally, differences in extraction methodology are expected to influence the observed microbial community composition and diversity, with mechanically biased methods yielding more reproducible and biologically representative profiles of epiphytic microbial communities in silage samples.

2.1.1.2 Evaluation of N₂O emissions from conserved forages and validation of the role of denitrification in N₂O production

This study hypothesized that conserved forages, including maize, alfalfa, and sorghum, will exhibit distinct N₂O emission patterns driven by differences in their carbon-to-nitrogen ratios and associated microbial communities. Denitrification and nitrification are expected to serve as the dominant pathway for N₂O production during forage conservation, as evidenced by the upregulation of denitrification-associated genes such as *narG*, *nirK/nirS*, and *nosZ*, along with measurable increases in N₂O accumulation during the ensiling process. Additionally, N₂O production is hypothesized to correlate positively with nitrate availability and negatively with carbon availability and pH, linking emission magnitude to the nutritional and physicochemical parameters of conserved forages.

2.1.1.3 Optimization of N₂O emission mitigation methods for conserved forages

This study hypothesized that the controlled supplementation of chlorate will selectively inhibit nitrate reductase activity, thereby reducing N₂O accumulation without negatively impacting silage fermentation quality. The co-application of readily available carbon sources, such as acetate or glucose, is anticipated to stimulate complete denitrification to N₂, enhancing the inhibitory efficiency of chlorate and resulting in a synergistic mitigation of N₂O emissions. Consequently, an optimized combination of chlorate concentration and carbon amendment is expected to achieve a sustained reduction in N₂O flux while preserving the nutritional quality of the silage, presenting a practical and cost-effective GHG mitigation strategy.

2.1.2 Objectives

2.1.2.1 Evaluation and optimization of lysis methods for microbial DNA extraction from epiphytic phyllosphere samples

1. Compare the efficiency of two commercially available DNA extraction kits that employ different lysis methodologies.
2. Assess the impact of various cell lysis techniques under different extraction conditions on DNA yield, purity, DIN, and the bacterial 16S rRNA gene copy number per ng of template genomic DNA (gDNA).
3. Investigate the effects of distinct DNA extraction methods on microbial community composition, diversity, and reproducibility, utilizing 16S rRNA gene amplicon sequencing.
4. Optimize the DNA extraction procedure for conserved forage based on the obtained results.

2.1.2.2 Evaluation of N₂O emissions from conserved forages and validation of the role of denitrification in N₂O production

1. Monitor simulated silages from three major crops in the United States (maize, alfalfa, and sorghum) over a 4-week period.
2. Assess the influence of various treatments on N₂O production in conserved forages, including dynamics of gene and transcript abundance.
3. Evaluate N₂O production and its correlation with physicochemical parameters.
4. Examine the environmental implications of N₂O emissions originating from conserved forages

2.1.2.3 Optimization of N₂O emission mitigation methods for conserved forages

1. Optimize chlorate concentrations to minimize N₂O emissions from forage conservation while inhibiting the denitrification process.

2. Investigate the addition of supplemental carbon sources to enhance the denitrification process in conserved forages, thereby mitigating N₂O emissions.

2.2 Impact of BNI on methanotrophs

2.2.1 Hypothesis

2.2.1.1 Examination of the influence of BNI compounds on pMMO-expressing *Methylosinus trichosporium* OB3b

This study posits that BNI compounds—namely MHPP, 1,9-D, and linoleic acid (LA)—will significantly inhibit the activity of pMMO in *M. trichosporium* OB3b. The inhibitory effects are expected to manifest as measurable alterations in kinetic parameters, including maximum methane oxidation rate ($V_{\max}$) and apparent substrate affinity (K_m).

2.2.1.2 Impact of BNI compounds on different conditions of *M. trichosporium* OB3b and various types of methanotrophs

It is anticipated that the inhibition patterns induced by BNI compounds will differ between the soluble methane monooxygenase (sMMO)-expressing and pMMO-expressing forms of *M. trichosporium* OB3b, given the structural and cofactor variations between these enzyme systems. Additionally, the inhibitory responses are hypothesized to vary among methanotroph types, with Type I methanotrophs (e.g., *Methylobacterium album* BG8) showing different susceptibility patterns than Type II (*M. trichosporium* OB3b).

2.2.1.3 Molecular gene abundance and RNA sequencing analysis

I hypothesized that BNI compounds will trigger distinct transcriptomic responses in methanotrophic cultures, reflecting both enzyme-specific inhibition and broader stress responses. Also, RNA-seq analysis will reveal compound-specific transcriptional signatures, providing mechanistic insights into how BNIs modulate methanotroph metabolism at the molecular level.

2.2.2 Objective

2.2.2.1 Examination of the influence of BNI compounds on *M.trichosporium* OB3b

1. Evaluate the inhibitory effects of BNI compounds MHPP, 1,9-D, and LA on the kinetic parameters of *M. trichosporium* OB3b.

2.2.2.2 Impact of BNI compounds on different conditions of *M. trichosporium* OB3b and various types of methanotrophs

1. Assess the inhibition rates of BNI compounds on the kinetic values for sMMO-expressing *M. trichosporium* OB3b.
2. Evaluate the inhibition rates of BNI compounds on the kinetic values for Type I methanotrophs (*M. album* BG8)

2.2.2.3 Plants exudates affect the methanotrophs-enriched soil

1. Assess the inhibition of CH₄ oxidation in the presence of growing plants which exudate BNI compounds.

2.2.2.4 Molecular gene abundance and RNA sequencing

1. Elucidate the transcriptomic responses of pure cultures of methanotrophs upon exposure to BNI compounds.

Chapter 3 - The role of microbial diversity in controlling GHGs emissions from conserved forages

N₂O is the third most important GHG, following CO₂ and CH₄. Per mole, N₂O has a warming potential 298 times greater than CO₂ and remains in the atmosphere for an extended period, estimated at 100–150 years (USEPA, 2023). Recent studies have also emphasized N₂O as the primary ozone-depleting substance in the stratosphere (Fang et al., 2019a; Ravishankara et al., 2009a). While natural processes and human activities contribute to N₂O production, agricultural activities are the predominant source, responsible for over two-thirds of global anthropogenic N₂O emissions (Davidson & Kanter, 2014b; D. S. Reay et al., 2012b). Despite extensive research on agricultural N₂O sources such as soils and manure management, forage conservation remains a relatively overlooked contributor. Forage conservation plays a vital role in livestock farming, ensuring a reliable feed source during times when fresh pasture is limited due to seasonal changes or inadequate growing conditions (Finch et al., 2014; Owens, 2003). Silage, a common forage conservation method, involves the anaerobic fermentation of plant material, which preserves nutrients through the activity of lactic acid bacteria (LAB) that convert soluble carbohydrates to lactic acid. This process creates an acidic environment that inhibits spoilage organisms and enhances forage stability (Müller et al., 2008a; Soundharrajan et al., 2017a). However, the same anaerobic and microbially active conditions that stabilize silage can also facilitate nitrogen-transforming processes that generate N₂O. In particular, microbial denitrification is considered a primary pathway contributing to N₂O emissions under these conditions, as part of the broader biogeochemical nitrogen cycle (Butterbach-Bahl et al., 2013a; Kuypers et al., 2018a). These processes suggest that conserved forages may represent a

previously underappreciated source of N₂O emissions within agricultural systems. To elucidate the mechanisms driving N₂O production in these systems, it is essential to characterize the structure and function of the associated microbial communities. Within conserved forages, particularly in the phyllosphere—the aerial region of plants—microbial communities are highly diverse and metabolically dynamic. This environment hosts both epiphytic (surface-associated) and endophytic (internal) microorganisms that actively participate in nitrogen cycling and influence plant–microbe interactions (Vorholt et al., 2012). However, accurate characterization of these microbial communities remains a critical challenge. DNA extraction from plant-associated samples is often hindered by co-extraction of host plant DNA, low microbial biomass, and the presence of inhibitory compounds such as polyphenols and polysaccharides. These constraints can introduce substantial bias in downstream molecular analyses, leading to misrepresentation of microbial community composition and function. As a result, methodological limitations in DNA extraction represent a key bottleneck in linking microbial ecology to observed N₂O emissions.

Addressing this limitation requires systematic evaluation and optimization of DNA extraction strategies tailored to phyllosphere samples. Therefore, this chapter first examines the environmental significance of N₂O emissions from forage conservation systems and the microbial processes underlying their production. It then focuses on the methodological challenges associated with DNA extraction from plant-associated matrices, including a comparative evaluation of lysis techniques and their impact on microbial community profiling. In addition, this study investigates mitigation strategies aimed at reducing N₂O emissions during forage fermentation, including the application of chlorate to inhibit denitrification pathways and the supplementation of external carbon sources to manipulate microbial activity. Together, these

approaches aim to both improve mechanistic understanding and identify practical solutions for reducing greenhouse gas emissions from forage-based agricultural systems.

Understanding these microbial communities, especially their role in N₂O emissions, requires high-quality DNA extraction for accurate molecular analyses. This chapter will provide a comprehensive background on the environmental impact of forage conservation, with a focus on microbial processes contributing to N₂O emissions. Additionally, it will explore the challenges of DNA extraction from plant-associated samples, reviewing various lysis techniques and evaluating their effectiveness in capturing accurate microbial profiles for molecular analyses. Also, we will optimize reducing GHG emission from forage fermentation using chlorate and extra carbon sources. This understanding is crucial for advancing sustainable forage management practices and mitigating GHG emissions in agriculture.

3.1. Evaluation and optimization of lysis method for microbial DNA extraction from epiphytic phyllosphere samples

3.1.1 Introduction

The phyllosphere is the aerial region of the plant, which supports diverse groups of microorganisms that can live on the surface (epiphytic) and interior (endophytic) of plant tissues (Vorholt, 2012). The epiphytic phyllosphere is an important microbial habitat and plays a crucial role in the global carbon and nitrogen cycles (Bringel and Cou'ee, 2015; Delmotte et al., 2009) as well as in plant health and growth (Wang et al., 2015). Epiphytic microorganisms are also vital in many industrial applications, such as *Pseudozyma* in bioplastic production (Sato et al., 2017; Watanabe et al., 2014) and lactic acid bacteria in the anaerobic forage conservation process (e.g., silage) (Holzer et al., 2003). Understanding the abundance, diversity, and functions of the epiphytic microbiome often depends on culture-independent molecular techniques

(Lucaciu et al., 2019). Molecular analysis of the epiphytic microbiome necessitates applying additional pretreatment steps such as washing (Kadivar and Stapleton, 2003; Saito et al., 2007; Wang et al., 2019) and sonication (Cui et al., 2021; Yang and Crowley, 2000) to isolate and purify microbial cells. However, a cell pellet sometimes unavoidably contains residual plant debris, and DNA and PCR inhibitors released from the plant cells often interfere with downstream molecular analysis of the microbial communities (Dent et al., 2004; Ikeda et al., 2009; Saito et al., 2007). Particularly, chloroplast 16S rRNA genes may ultimately hinder the detection of the bacterial 16S rRNA gene in PCR amplification (Dent et al., 2004). It is critical to develop a simple and reproducible DNA extraction method optimized for each sample type to explore the microbial community in environmental samples using a molecular approach. Typical DNA extraction protocols from an epiphytic phyllosphere sample comprise three major steps: pretreatment, cell lysis, and subsequent DNA purification. This study focuses on the second step, cell lysis, to minimize contamination from plant-derived biomolecules. Notably, cell lysis relies on mechanical or enzymatic disruption in most commercial DNA isolation kits. Mechanical disruption is typically based on bead beating to break the cell membrane. The advantage of mechanical disruption is efficacy and concurrent sample homogenization; however, applications relying on intact chromosomes could be limited owing to DNA shearing (Bag et al., 2016; Corcoll et al., 2017; Pakpour et al., 2013; Salonen et al., 2010). Enzymatic disruption is more suitable for obtaining genomic DNA (gDNA) but may suffer from extraction bias owing to the limited spatial access to all target organisms and different sensitivity for different cell types (Robe et al., 2003; Saito et al., 2007). Using the Google Scholar database (accessed on 9/13/2022), we selected the first 100 recent articles published after 2020, searched using the keywords “phyllosphere” and “sequencing” and classified them based on the lysis method. A

total of 70 studies employed commercial kits with mechanical disruption, whereas 30 used commercial kits with enzymatic disruption (data available upon request), indicating that both methods are widely used in different research areas. However, to the best of our knowledge, there is no study examining the effect of cell lysis methods and extraction conditions on DNA quantity and quality and subsequent molecular analysis of the phyllosphere microbiome. Factors affecting DNA extraction efficiency by each lysis method include bead-beating time, speed, and temperature for mechanical disruption (Li et al., 2007) and temperature and contact time for enzymatic disruption (Qamar et al., 2017). For this study, we selected two commercially available DNA extraction kits employing different lysis methods and evaluated the effect of cell lysis methods under different extraction conditions on DNA extraction yield, purity, DNA integrity number (DIN), and bacterial 16S rRNA gene copy number per ng template gDNA as a measure of preferential lysis of bacterial cells over plant cells. Furthermore, the effect of DNA extraction methods on microbial community composition, diversity, and reproducibility was examined using 16S rRNA gene amplicon sequencing.

3.1.2 Materials and methods

3.1.2.1 Sample and pretreatment

An epiphytic phyllosphere sample was collected from a 30-day-old alfalfa silage. Silage is a form of forage conservation in an anaerobic environment (Collins et al., 2017). Silage is typically based on fermentation and, thus, is a microorganism-rich phyllosphere niche. As a conventional pretreatment method for epiphytic phyllosphere samples to isolate and purify microbial cells minimizing plant material, each sample (5 g) was suspended in 45 mL of sterile 0.85% NaCl solution for 2 h at 20 °C, filtered through two cheesecloth layers, and centrifuged at 12,000 ×g at 4 °C for 15 min (Li et al., 2020; Saito et al., 2007; Wang et al., 2019). The

supernatant was discarded, the pellet was stored at -80°C until subsequent DNA extraction, and the solid residue was also collected and stored at -80°C .

3.1.2.2 DNA extraction

Mechanical and enzymatic disruptions are two cell lysis methods commonly used for DNA extraction from phyllosphere samples. This study selected the DNeasy PowerSoil Pro kit and DNeasy Blood and Tissue kits from the same manufacturer (Qiagen, Hilden, Germany) to represent each lysis method. DNA extraction using the PowerSoil kit was conducted according to the manufacturer's instructions with the following modifications: (1) three different bead-beating times, i.e., 5, 10, and 15 min at (2) two temperatures, i.e., 20°C and 30°C . The protocol recommended by the manufacturer was bead-beating for 10 min at 20°C . Mechanical cell disruption was performed using a Bead Mill 24 homogenizer (Fisher Scientific, Waltham, MA, USA) in a temperature-controlled room. DNA extraction using the Blood and Tissue kit was performed with the following modifications: (1) four incubation times with proteinase K (5, 10, 20, and 60 min) at (2) three temperatures (20°C , 40°C , and 56°C). The manufacturer's protocol for prokaryotes was incubation for 10 min at 56°C , and 30 min was recommended for gram-positive bacteria. A longer incubation time (60 min) was employed to evaluate the effect of lower incubation temperatures. The enzyme-sample mixture was incubated in a water bath set at different temperatures. The total DNA yield was obtained by repeated elutions until the DNA concentration was lower than $5\text{ ng }\mu\text{L}^{-1}$ (**Table 1**). All extractions were performed in triplicate. One gram of the solid residue on the cheesecloth was also subjected to DNA extraction using the DNeasy PowerSoil Pro kit to ensure the efficacy of the pretreatment process.

3.1.2.3 Determination of DNA yield, purity, and integrity

DNA yield and purity were determined using the Epoch 2 Microplate Spectrophotometer and TAKE3™ microvolume plate (Agilent Technologies, Santa Clara, CA, USA). An absorption ratio at 260/280 nm ($OD_{260/280}$) was used as an indicator of protein contamination, and the ratio of 260/230 nm ($OD_{260/230}$) was used for a secondary purity check. DIN was measured using the Agilent 2200 TapeStation system (Agilent Technologies, Santa Clara, CA, USA). Notably, DIN determines the level of DNA degradation by assessing the distribution of electrophoretic signals across the size range using a proprietary algorithm. The higher the DIN value, the better the integrity of the gDNA sample.

3.1.2.4 Relative abundance of microbial DNA in the DNA extract

Quantification of PCR (qPCR) targeting bacterial 16S rRNA gene was performed using the CFX Opus 96 Real-Time PCR System (Bio-Rad, Hercules, CA, USA) in 20- μ L reaction mixtures containing 10 μ L of SsoAdvanced™ Universal SYBR Green Supermix (Bio-Rad, Hercules, CA, USA), 300 nM of each primer, and 2 μ L of template DNA. A primer pair of 1055F and 1392R was used for amplification (Park et al., 2017). Standard curves for quantification were prepared in triplicate using double-stranded synthetic DNA fragments (gBlocks®, Integrated DNA Technologies, Coralville, IA, USA) as control templates. The primer sequences are presented in **Table 2**. The qPCR program was initiated at 95 °C for 10 min and continued at 40 cycles of 95 °C for 15 s and 59 °C for 1 min.

Table 1. DNA yield (μg) of each elution for the samples prepared by the enzymatic disruption under different conditions. The numbers in the parentheses indicate the standard errors of the estimates.

Elution	Incubation time	20°C			Avg	40°C			Avg	56°C			Avg
	min	1	2	3		1	2	3		1	2	3	
1st	5	12.5	11.7	13.7	12.6 (0.8)	7.8	6.6	7.2	7.2 (0.5)	10.9	11.0	12.0	11.3 (0.5)
	10	13.6	15.5	11.5	13.5 (1.6)	6.0	6.6	5.8	6.1 (0.3)	10.3	14.3	12.0	12.2 (1.6)
	20	14.2	6.9	7.6	9.6 (3.3)	5.0	6.3	6.1	5.8 (0.6)	12.7	16.2	12.8	13.9 (1.6)
	60	5.5	5.7	6.9	6.0 (0.6)	4.8	3.9	4.5	4.4 (0.4)	9.6	10.3	9.5	9.8 (0.3)
2nd	5	22.5	17.2	20.8	20.1 (2.2)	14.7	14.5	11.9	13.7 (1.3)	13.5	14.6	18.0	15.4 (1.9)
	10	12.1	19.9	19.9	17.3 (3.7)	13.7	14.1	8.9	12.2 (2.4)	14.4	18.5	14.3	15.7 (2.0)
	20	22.8	14.1	13.4	16.8 (4.3)	8.3	5.7	14.6	9.5 (3.8)	16.0	15.9	22.4	18.1 (3.0)
	60	14.1	12.5	13.1	13.2 (0.7)	7.9	8.9	7.7	8.2 (0.6)	16.4	18.4	17.8	17.6 (0.8)
3rd	5	13.4	21.6	16.9	17.3 (3.4)	9.6	8.9	9.5	9.4 (0.3)	12.1	12.4	10.6	11.7 (0.8)
	10	18.7	13.3	16.6	16.2 (2.2)	6.3	8.6	11.5	8.8 (2.1)	9.2	16.1	11.5	12.3 (2.9)
	20	12.1	9.9	12.3	11.4 (1.1)	14.2	12.6	11.0	12.6 (1.3)	15.4	15.3	10.5	13.7 (2.3)
	60	12.6	12.9	7.5	11.0 (2.5)	15.4	8.5	8.5	10.8 (3.3)	8.8	10.1	10.8	9.9 (0.8)
4th	5	7.9	11.7	11.1	10.2 (1.6)	11.0	9.8	10.4	10.4 (0.5)	8.4	7.9	8.0	8.1 (0.2)
	10	12.0	8.6	10.9	10.5 (1.4)	10.1	6.4	13.7	10.1 (3.0)	9.4	7.6	7.2	8.0 (1.0)
	20	9.6	6.5	8.6	8.2 (1.3)	7.1	9.7	7.4	8.1 (1.2)	5.4	6.6	6.7	6.3 (0.6)
	60	8.8	8.7	11.4	9.7 (1.3)	6.5	7.4	6.2	6.7 (0.5)	5.3	6.0	5.3	5.5 (0.3)
5th	5	6.9	8.4	7.2	7.5 (0.6)	12.7	12.8	17.8	14.4 (2.4)	5.1	5.7	5.2	5.4 (0.3)
	10	6.4	7.2	10.0	7.9 (1.5)	13.9	15.9	13.5	14.4 (1.0)	4.6	6.9	6.5	6.0 (1.0)
	20	14.3	18.7	12.4	15.1 (2.7)	16.0	18.5	14.1	16.2 (1.8)	6.7	6.3	7.2	6.7 (0.4)
	60	12.4	13.2	14.1	13.3 (0.7)	13.0	11.0	11.2	11.7 (0.9)	5.1	5.9	5.7	5.6 (0.3)
6th	5	4.1	5.6	5.9	5.2 (0.8)	2.9	3.0	3.9	3.3 (0.5)	3.5	3.4	3.8	3.6 (0.2)
	10	4.8	5.7	8.3	6.3 (1.5)	3.8	3.9	3.5	3.7 (0.2)	4.8	4.4	4.1	4.5 (0.3)
	20	2.3	3.8	1.8	2.6 (0.8)	3.2	2.8	2.9	2.9 (0.2)	3.1	3.7	4.0	3.6 (0.4)
	60	1.8	2.8	2.2	2.3 (0.4)	2.6	2.9	2.3	2.6 (0.2)	3.7	3.0	3.8	3.5 (0.4)
7th	5	4.1	5.6	5.9	5.2 (0.8)	2.9	3.0	3.9	3.3 (0.5)	3.9	5.4	4.7	4.7 (0.6)
	10	4.8	5.7	8.3	6.3 (1.5)	3.8	3.9	3.5	3.7 (0.2)	4.6	4.9	6.2	5.2 (0.7)
	20	2.3	3.8	1.8	2.6 (0.8)	3.2	2.8	2.9	2.9 (0.2)	4.5	4.7	5.0	4.7 (0.2)
	60	1.8	2.8	2.2	2.3 (0.4)	2.6	2.9	2.3	2.6 (0.2)	3.7	4.9	4.3	4.3 (0.5)
8th	5	5.7	6.2	4.8	5.5 (0.6)	3.3	3.7	4.0	3.7 (0.3)	4.0	4.6	4.5	4.4 (0.3)
	10	4.3	4.5	9.1	6.0 (2.2)	2.4	2.6	3.7	2.9 (0.6)	3.8	4.4	3.3	3.9 (0.5)

	20	3.2	3.4	2.6	3.1 (0.4)	3.4	3.6	2.9	3.3 (0.3)	3.8	4.4	4.1	4.1 (0.2)
	60	2.1	2.6	1.6	2.1 (0.4)	2.9	1.8	1.9	2.2 (0.5)	4.2	3.5	3.2	3.6 (0.4)
9th	5	2.9	2.9	3.3	3.0 (0.2)	2.9	3.4	3.2	3.2 (0.2)	2.5	2.8	2.8	2.7 (0.1)
	10	3.1	2.8	2.5	2.8 (0.2)	2.1	2.1	2.4	2.2 (0.1)	1.9	3.5	2.1	2.5 (0.7)
	20	2.2	2.1	1.9	2.1 (0.1)	1.9	2.4	2.2	2.2 (0.2)	2.2	2.5	2.5	2.4 (0.2)
	60	1.6	1.7	2.1	1.8 (0.2)	3.1	1.4	1.5	2.0 (0.8)	1.6	1.9	1.7	1.7 (0.1)
10th	5	2.6	2.3	2.2	2.4 (0.2)	2.8	2.9	2.2	2.6 (0.3)	1.7	1.7	2.2	1.9 (0.3)
	10	2.6	2.0	1.4	2.0 (0.5)	1.9	2.3	1.6	1.9 (0.3)	1.6	2.9	1.6	2.0 (0.6)
	20	1.7	2.1	1.7	1.8 (0.2)	2.2	1.4	1.4	1.7 (0.4)	1.6	1.8	2.2	1.9 (0.2)
	60	1.5	1.6	0.9	1.4 (0.3)	1.4	1.3	2.1	1.6 (0.3)	1.3	1.7	1.6	1.5 (0.2)
11th	5	2.7	2.1	2.1	2.3 (0.3)	2.0	3.0	1.6	2.2 (0.6)	1.5	1.8	1.7	1.6 (0.1)
	10	2.8	2.0	0.8	1.9 (0.9)	2.2	1.8	1.1	1.7 (0.4)	1.2	2.1	0.8	1.4 (0.6)
	20	1.7	1.8	1.5	1.6 (0.1)	1.7	1.2	1.4	1.4 (0.2)	1.3	1.6	1.7	1.5 (0.2)
	60	1.1	1.2	0.9	1.1 (0.1)	1.5	0.9	1.1	1.2 (0.2)	1.1	1.7	0.9	1.2 (0.3)
12th	5	1.5	1.6	1.6	1.6 (0.0)	2.0	2.6	1.6	2.1 (0.4)	1.2	1.3	1.4	1.3 (0.1)
	10	1.8	2.0	2.1	2.0 (0.1)	1.8	2.0	1.0	1.6 (0.4)	1.4	2.3	0.7	1.5 (0.7)
	20	1.2	1.6	2.9	1.9 (0.7)	1.2	1.2	1.4	1.3 (0.1)	1.1	1.3	1.5	1.3 (0.2)
	60	2.0	1.3	1.2	1.5 (0.4)	1.7	0.9	1.0	1.2 (0.4)	0.8	0.9	1.0	0.9 (0.1)
13th	5	1.2	1.3	1.3	1.3 (0.0)	1.0	2.4	1.4	1.6 (0.6)	1.0	1.1	1.9	1.3 (0.4)
	10	1.9	1.2	1.1	1.4 (0.4)	1.1	1.5	0.8	1.1 (0.3)	0.9	2.2	0.4	1.2 (0.7)
	20	0.8	2.9	0.8	1.5 (1.0)	1.4	1.4	0.8	1.2 (0.3)	0.9	1.1	1.2	1.1 (0.1)
	60	1.0	1.0	1.4	1.1 (0.2)	0.8	0.9	1.1	0.9 (0.1)	0.7	0.8	0.8	0.7 (0.1)
14th	5	1.1	1.2	1.3	1.2 (0.1)	0.2	1.7	1.6	1.2 (0.7)	0.9	0.9	1.1	1.0 (0.1)
	10	1.4	1.0	0.8	1.1 (0.2)	1.6	1.0	1.6	1.4 (0.3)	0.7	1.5	0.4	0.9 (0.5)
	20	1.0	2.2	0.8	1.3 (0.6)	1.2	0.7	0.9	0.9 (0.2)	0.7	0.8	0.9	0.8 (0.1)
	60	0.9	0.7	0.5	0.7 (0.1)	1.3	0.7	0.5	0.8 (0.4)	0.6	0.7	0.5	0.6 (0.1)
15th	5	0.7	0.8	1.0	0.8 (0.1)	1.0	2.0	1.5	1.5 (0.4)	0.0	0.0	0.0	0.0 (0.0)
	10	1.9	0.9	0.9	1.2 (0.5)	1.8	0.9	1.0	1.2 (0.4)	0.0	0.0	0.0	0.0 (0.0)
	20	0.5	1.0	1.8	1.1 (0.5)	0.0	0.0	0.0	0.0 (0.0)	0.0	0.0	0.0	0.0 (0.0)
	60	0.5	0.5	0.4	0.5 (0.0)	0.0	0.0	0.0	0.0 (0.0)	0.0	0.0	0.0	0.0 (0.0)
16th	5	0.9	0.7	0.9	0.8 (0.1)	1.6	1.2	1.0	1.3 (0.2)	0.0	0.0	0.0	0.0 (0.0)
	10	0.4	0.7	0.1	0.4 (0.2)	1.1	1.0	0.5	0.9 (0.3)	0.0	0.0	0.0	0.0 (0.0)
	20	0.0	0.0	0.0	0.0 (0.0)	0.0	0.0	0.0	0.0 (0.0)	0.0	0.0	0.0	0.0 (0.0)
	60	0.0	0.0	0.0	0.0 (0.0)	0.0	0.0	0.0	0.0 (0.0)	0.0	0.0	0.0	0.0 (0.0)

Table 2. Summary of primers information

Target gene	Primer	Nucleotide sequence (5'-3')	Reference
Universal 16S rRNA	1055F	ATGGCTGTCAGCT	(Ferris et al., 1996)
	1392R	ACGGGCGGTGTGTAC	
V3 and V4 region	Forward	CCTACGRRBGCASCAGKVRVGAAT	(Teng et al., 2018)
	Reverse	GGACTACNVGGGTWTCTAATCC	
V4 and V5 region	Forward	GTGYCAGCMGCCGCGGTAA	(Teng et al., 2018)
	Reverse	CTTGTGCGGKCCCCCGYCAATTC	

where, N=G,A,T,orC;Y=C or T;M=A or C;K=G or T;S=G or C;R=G or A;B=C,T or G;W=A or T;V=A, C or G

3.1.2.5 Amplicon sequencing and bioinformatic analysis

Bacterial 16S rRNA amplicon sequencing was conducted at GENEWIZ, Inc. (South Plainfield, NJ, USA). DNA samples were quantified using a Qubit 2.0 Fluorometer (Invitrogen, Carlsbad, CA, USA), and 30–50 ng DNA was used to generate amplicons using a MetaVx™ Library Preparation kit (GENEWIZ, Inc., South Plainfield, NJ, USA). The V3/V4/V5 hypervariable regions were analyzed using two amplicons—one each for the V3/V4 and V4/V5 regions. **Table 2** presents the primer sequences targeting each region. The amplicons were sequenced on the same run using the Illumina MiSeq™ with a 2×250 bp paired-end configuration. The Quantitative Insights Into Microbial Ecology (QIIME) 1.1.9, an open-source bioinformatics pipeline, was used for data analysis. Quality filtering on sequence reads was performed to remove low-quality or ambiguous reads. Beta diversity was calculated using weighted and unweighted Unifrac, and principal coordinate analysis (PCoA) was performed with R v.4.1.1 software. The sequencing data have been deposited at NCBI under accession numbers SRR22404852, SRR22404853, SRR22404854, SRR22404855, and SRR22404856.

3.1.2.6 Statistical analysis

One-way ANOVA analysis was employed with the Tukey Honestly Significant Difference test to assess differences between the extraction methods for DNA yield, purity, DIN, and normalized microbial 16S rRNA gene copy number. P-values were adjusted for multiple testing using the false discovery rate method. All statistical analyses were performed using the R v.4.1.1 software.

3.1.3 Results and discussion

3.1.3.1 DNA yield, purity, and integrity

The effect of the cell lysis methods on microbial DNA extraction from the phyllosphere sample was examined using two commercially available kits with modifications. The Blood and Tissue kit was used for enzymatic disruption, and the PowerSoil kit was used for mechanical disruption. As shown in **Table 3**, enzymatic disruption exhibited one to two orders of magnitude higher DNA yields (66.5–97.1 µg) than mechanical disruption (0.54–5.24 µg) regardless of the extraction conditions. However, such a high yield by enzymatic disruption was possible only after repeated elutions up to 16 times (**Table 1**). In contrast, mechanical disruption required a single elution (data not presented). In multiple elutions, similar amounts of DNA (6–20 µg) were retrieved for the first three to four elutions, and then the DNA concentrations gradually decreased. At the first elution, the DNA yield was 10.39 ± 2.83 µg on average, only two times higher than the sample prepared via mechanical disruption at 20 °C (4.73 ± 0.41 µg). The OD_{260/280} of the samples prepared by the enzymatic disruption ranged from 1.46 to 1.52 (**Table 1**), which is considered “appreciably lower” (<1.6) (Corcoll et al., 2017). Severe contamination by plant cell materials was observed with enzymatic disruption even in the mildest lysis conditions (i.e., incubation with proteinase for 5 min at 20 °C). Given that the recommended protocol by the manufacturer of the Blood and Tissue kit for the prokaryote is much more intense (i.e., incubation for 10 min at 56 °C for bacteria in general and 30 min at 56 °C for gram-positive

bacteria), extra care should be taken when enzymatic disruption is used for microbial DNA extraction from an environmental sample containing plant materials. We performed an additional DNA purification step to improve the OD_{260/280} using a commercial kit, but the purity was not noticeably improved (data not shown). The OD_{260/280} for mechanical disruption was higher than 1.8, not exceeding 2.0 (**Table 3**), which is generally accepted as “pure” for DNA (Carrigg et al., 2007; Jorgez et al., 2006). For enzymatic disruption, as the lysis intensity increased, i.e., the incubation temperature or incubation time increased at fixed temperatures, the DNA yield decreased, but there were no significant differences in the OD_{260/280}. The highest DNA yield was observed in the mildest lysis condition (incubation for 5 min at 20 °C), indicating that further optimization of this method may result in a higher yield. The temperature significantly affected the DNA yield ($P < 0.001$) for mechanical disruption. DNA yields at 20 °C ($4.73 \pm 0.54 \mu\text{g}$) were 10 times higher than those at 30 °C ($0.57 \pm 0.03 \mu\text{g}$) regardless of the bead-beating time. At the same temperature, variations in the bead-beating time had no impact on the DNA yield and the OD_{260/280}. There were no significant differences in the OD_{260/280} regardless of the extraction conditions. The OD_{260/230} values were low (0.7 – 0.86) irrespective of the extraction methods, typically indicating the presence of polysaccharides from plant materials (Varma et al., 2007). Higher DINs indicate larger DNA fragments, and low DINs indicate fragmented DNA. Sequencing prefers higher DINs. Unexpectedly, lower DINs (< 2), i.e., more severe DNA shearing, were observed for enzymatic disruption regardless of the extraction conditions, whereas mechanical disruption samples showed significantly higher DINs (> 5 , $P < 0.001$). In general, mechanical disruption is the most popular and commercially available method with high throughput and high efficiency (Husseini et al., 2022), but it can result in extensive shearing of DNA. Enzymatic disruption is typically considered the most common alternative to mechanical

disruption to extract longer DNA fragments. However, in this study, the DINs were consistently lower in the samples prepared with enzymatic disruption than those with mechanical disruption. Further optimization with enzymatic disruption, such as shorter incubation time and lower temperature, may improve DNA integrity. For mechanical disruption, as the bead-beating time increased, DIN decreased significantly at 20 °C ($P < 0.01$) and only numerically at 30 °C. This suggests that temperature is the controlling factor in minimizing DNA shearing for the mechanical lysis method.

3.1.3.2 Quantification of bacterial 16S rRNA gene

The total copy numbers of the bacterial 16S rRNA gene in the solid residue samples were two orders of magnitude smaller than the pellets, suggesting that the pretreatment method used in this study was suitable for collecting bacterial biomass. The bacterial 16S rRNA gene copy numbers normalized by the template DNA concentration were used to assess the microbial DNA extraction efficiency from the epiphytic phyllosphere sample, where a lower value indicates high plant DNA content. Mechanical disruption exhibited 2–12 times higher bacterial 16S rRNA gene copies per ng template gDNA ($2.7 \pm 0.1 \times 10^7 - 6.5 \pm 0.1 \times 10^7$) than enzymatic disruption ($5.2 \pm 0.0 \times 10^6 - 1.8 \pm 0.0 \times 10^7$) (**Fig. 1**). The copy numbers were consistent for mechanical disruption when extraction was performed at the same temperature irrespective of the bead-beating time, suggesting that the extraction temperature is a controlling factor for the preferential lysis of bacterial rather than plant cells (**Fig. 1b**). At 30 °C, two times higher copy numbers ($6.5 \pm 0.1 \times 10^7$) were detected than at 20 °C ($3.4 \pm 0.0 \times 10^7$). For enzymatic disruption, there were significant variations in the gene copy numbers depending on extraction conditions (**Fig. 1a**). At 56 °C, no significant differences were observed in the copy number as the incubation time varied. At 20 °C and 40 °C incubation temperatures, the 16S rRNA gene copy numbers increased

Table 3. Effects of cell lysis methods and extraction conditions on DNA yield (μg), purity (absorbance ratios of 260/280 nm ($\text{OD}_{260/280}$) and 260/230 nm ($\text{OD}_{260/230}$)), and DNA integrity (DIN). The numbers in the parentheses indicate the standard errors of

Blood & tissue kit													
Incubation	Temperature	20 °C				40 °C				56 °C			
	Time (min)	5	10	20	60	5	10	20	60	5	10	20	60
DNA total yield		97.1 (4.6)	96.6 (5.4)	80.9 (7.3)	67.9 (1.0)	80.9 (1.8)	74.1 (0.5)	70.0 (0.2)	57.0 (6.2)	74.3 (3.2)	77.3 (10.4)	80.1 (3.3)	66.5 (2.7)
DIN		1.33 (0.29)	1.40 (0.1)	1.20 (0.16)	1.00 (0.00)	1.2 (0.28)	1.13 (0.12)	1.00 (0.00)	1.05 (0.05)	1.00 (0.00)	1.27 (0.05)	1.53 (0.12)	1.13 (0.09)
OD _{260/230}		0.73	0.73	0.74	0.73	0.73	0.73	0.73	0.72	0.70	0.70	0.71	0.71
OD _{260/280}		1.49	1.46	1.52	1.51	1.49	1.50	1.52	1.50	1.50	1.47	1.49	1.50

Power Soil kit							
Incubation	Temperature	20 °C			30 °C		
	Time (min)	5	10	15	5	10	15
DNA total yield		4.2 (0.1)	4.8 (0.2)	5.2 (0.1)	0.5 (0.0)	0.6 (0.0)	0.6 (0.0)
DIN		6.23 (0.12)	5.83 (0.17)	5.37 (0.05)	6.33 (0.12)	6.20 (0.08)	6.07 (0.09)
OD _{260/230}		0.81	0.86	0.78	0.80	0.82	0.79
OD _{260/280}		1.85	1.84	1.85	1.80	1.98	1.85

as the incubation time increased. The results suggest that mechanical disruption is more suitable for the preferential lysis of bacterial cells than enzymatic disruption.

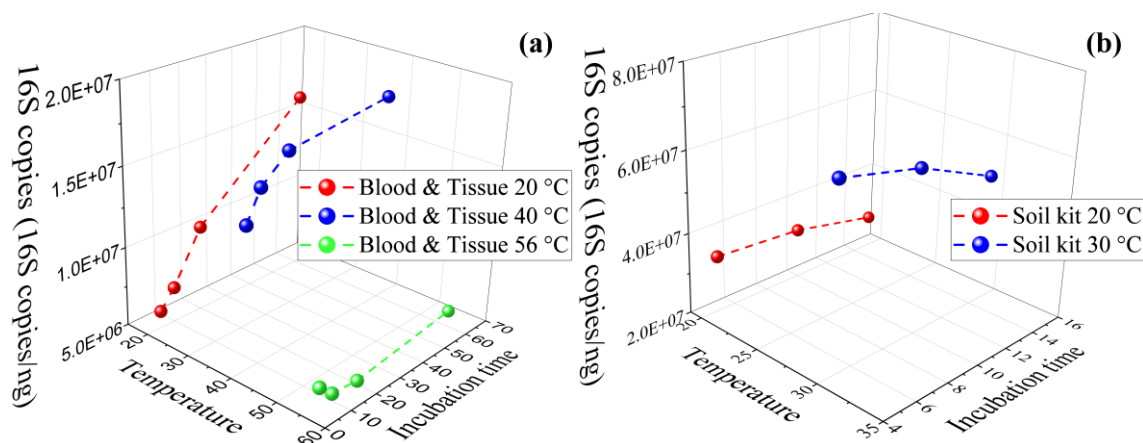


Figure 1. The copy numbers of bacterial 16S rRNA gene normalized by the template DNA amount in the samples prepared by (a) enzymatic and (b) mechanical lysis methods, respectively, under different extraction conditions.

3.1.3.3 Composition and diversity of the bacterial community

Bacterial 16S rRNA amplicon sequencing was performed on selected DNA extracts in duplicate to examine the effects of different DNA extraction methods on the patterns of soil bacterial community structure and diversity. The sequences per sample ranged from 15,094 to 445,904, clustered at a 97% sequence identity threshold. A total of 2767 OTUs were identified at both phylum and genus levels. Depending on the extraction conditions, distinctive variations were observed in the bacterial community compositions between the samples prepared via enzymatic disruption. In contrast, the samples prepared via mechanical disruption exhibited consistent community compositions irrespective of the extraction conditions (**Fig. 2**). At the phylum level, there was a high abundance of unclassified phyla (72%) in the samples prepared via enzymatic disruption for 5 min at 20 °C (BT 5m20c) but not in the samples prepared using the same lysis method for 60 min at 40 °C (BT 60m40c, <5%) (**Fig. 2a**). In the samples prepared via

mechanical disruption, the compositions were consistent. In all the samples, firmicutes was the predominant phylum, followed by proteobacteria, actinobacteria, and cyanobacteria. At the genus level, BT 5m20c exhibited the highest abundance of unclassified genera (up to 88%), whereas BT 60m40c displayed the lowest abundance of unclassified genera (20%). A positive correlation between small-sized (sheared) DNA and an abundance of unclassified sequences was previously reported (De Liphay et al., 2004; Desneux and Pourcher, 2014; Teng et al., 2018). However, in this study, DIN did not correspond to the abundance of unclassified sequences (**Table 3**). It can be speculated that the proportion of plant DNA may have been significant in the extracts prepared via enzymatic disruption, and the DINs did not represent the DIN of microbial DNA. Simultaneously, microbial DNA may have been released earlier than plant DNA and may have been exposed to the proteinase longer, resulting in more small-sized DNA in BT 60m40c than in BT 5m20c. The samples prepared using mechanical disruption also showed consistent community compositions at the genus level regardless of the extraction conditions. In all samples, *Pediococcus* was a dominant genus, followed by *Weissella*, *Lactobacillus*, and *Lactococcus* (**Fig. 2b**). Previous studies have shown that differences in the bacterial cell wall and membrane structures cause DNA extraction to be more or less effective from some organisms. Such a discrepancy can introduce bias in the estimates of relative abundances of microbes in samples; thus, mechanical lysis methods are considered less biased than enzymatic lysis methods (Carrigg et al., 2007; Krsek and Wellington, 1999). Indeed, in this study, the samples prepared using the mechanical lysis method exhibited consistent microbial community compositions irrespective of the extraction conditions. PCoA analysis also confirmed that mechanical disruption allowed consistent analysis of microbial community structure (**Fig. 3**). The first axis of the PCoA explains 59.98% of the variance, whereas the second axis explains 14.02% of the

variance from the samples prepared using the Blood and Tissue and PowerSoil kits under different conditions. All duplicate samples were clustered close to each other, suggesting that both lysis methods are reproducible. Irrespective of the extraction conditions, the samples prepared via mechanical disruption were clustered together, whereas those prepared via enzymatic disruption were located separately in response to the different extraction conditions. This indicates that the mechanical lysis method is less biased than the enzymatic approach.

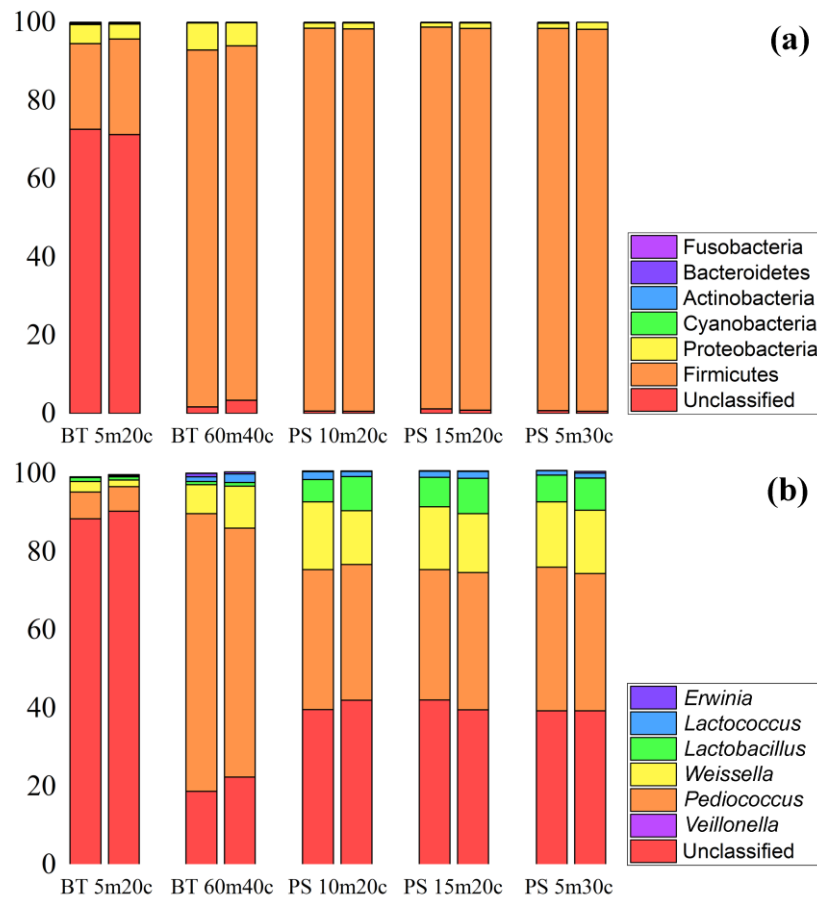


Figure 2. Comparison of relative abundance of 16S rRNA OTUs in the samples prepared by two different lysis methods under different extraction conditions at the phylum (a), the genus (b) level. Each bar represents one replicate sample.

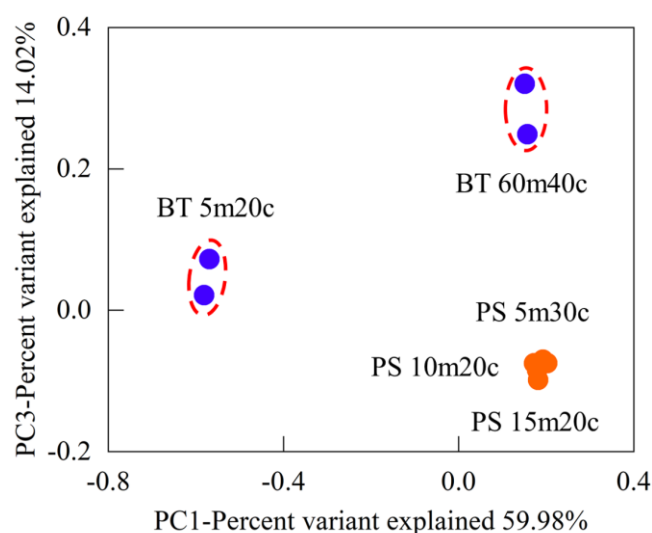


Figure 3. PCoA plot of 16S rRNA OTU abundance in all samples. Each symbol represents different samples prepared by the following extraction conditions: 5 min incubation at 20°C using Blood and Tissue kit (BT 5m20c), 60 min incubation at 40°C using Blood and Tissue kit (BT 60m40c), 10 min bead-beating time at 20°C using PowerSoil kit (PS 10m20c), 15 min bead-beating time at 20°C using PowerSoil kit (PS 15m20c), and 5 min bead-beating time at 30°C using PowerSoil kit (PS 5m30c).

3.1.4 Conclusion

We hypothesized that an optimized cell lysis protocol would allow preferential bacterial cell lysis over plant cells. With enzymatic disruption, the DNA quality was suboptimal for downstream applications, and more importantly, the bias in the microbial community profiles was prominent, albeit with high DNA yield. In the meantime, the samples prepared via mechanical disruption exhibited high DNA quality and consistent microbial compositions irrespective of extraction conditions. Therefore, this study demonstrates that the mechanical disruption method is more suitable for epiphytic phyllosphere samples than the enzymatic disruption method, and disruption at 20 °C for 5 min was found to be optimal for the fermented forage samples used in this study.

3.2 Forage conservation is a neglected N₂O source

3.2.1 Introduction

N₂O is the third most important GHG, following CO₂ and CH₄. On a molar basis, N₂O has a warming potential 300 times greater than CO₂ and remains in the atmosphere for an extended period, estimated at 100–150 years (USEPA, 2019). Recent studies have also emphasized N₂O as the primary ozone-depleting substance in the stratosphere (Fang et al., 2019b; Ravishankara et al., 2009b). Atmospheric N₂O concentrations have increased by over 20% compared to preindustrial levels, with the fastest increase occurring in the last five decades (Hall et al., 2007; Tian, Xu, Canadell, Thompson, Yao, et al., 2020b). While natural processes and human activities contribute to N₂O production, agricultural activities are one of the dominant sources, responsible for over two-thirds of global anthropogenic N₂O emissions (Davidson & Kanter, 2014a; D. S. Reay et al., 2012a; Tian, Xu, Canadell, Thompson, Yao, et al., 2020b). Numerous countries compile GHG emission inventories following the technical guidelines provided by the Intergovernmental Panel on Climate Change. For example, the USEPA annually publishes a report titled “Inventory of U.S. Greenhouse Gas Emissions and Sinks,” which accounts for N₂O emissions from agricultural activities through three sources: agricultural land management, manure management, and the burning of agricultural residues. Forages, the plant materials consumed by herbivores, are conserved to sustain livestock during periods of limited pasture growth or inadequate grazing conditions (Finch et al., 2014b; Owens, 2003b).

Globally significant for productive and efficient livestock production, forage conservation methods mainly involve hay and silage. For long-term storage, hay is dried to below 20% moisture (12–20%, w/w) to curtail microbial growth and stored under aerobic conditions (Owens, 2003b). Conversely, silage is produced at higher moisture levels (40–70%,

w/w) and stored strictly under anaerobic conditions to facilitate fermentation. Indigenous or exogenous lactic acid bacteria (LAB) convert soluble carbohydrates into organic acids, predominantly lactic acid, which acts as a natural preservative that inhibits unwanted microorganisms (Müller et al., 2008b; Soundharrajan et al., 2017b). Commercial silage inoculants containing LAB, such as *Lentilactobacillus buchneri* and other facultative anaerobic bacteria, are used as additives to enhance silage fermentation (Muck et al., 2018b). The fermentation process typically takes 2 to 4 weeks to complete, and most silages can be stored for 6 to 12 months, although the length of storage time depends on the crop and weather conditions (Owens, 2003b). The market for global silage inoculants reached USD 503 million in 2021 and is projected to grow to USD 630 million by 2028 at a compound annual growth rate of 3.8%. Silage is a significant segment of the global livestock industry, with 162.3 million metric tons (MMT) harvested in the United States in 2022 (Grossi et al., 2019b; USDA, 2023b). Despite its high production volume and extensive use, forage conservation has been limitedly studied as a potential source of GHG emissions (Grossi et al., 2019b).

While gas production during forage conservation has been investigated, previous studies have primarily focused on odorous chemicals, such as volatile organic compounds and ammonia (NH₃) (Hafner et al., 2013; Morgan & Pereira, 1962; Pusede & Cohen, 2012). At best, the detection of N₂O has been reported in previous studies (Bueno et al., 2020b; Chuan Wang & Burris, n.d.; Krommweh et al., 2020b; Schmithausen et al., 2022), but a quantitative assessment is still lacking. That is, our study, to our knowledge, is the first to provide comprehensive quantitative estimates on a per-crop basis, which can be scaled to national and sector-level estimates for N₂O emission from forage conservation. The precise mechanism underlying N₂O production remains partially understood, but microbial processes are widely considered the main

contributors to N₂O emissions (Hu et al., 2015c; Klimasmith & Kent, 2022c; Kuypers et al., 2018b; A. J. Thomson et al., 2012; Tian, Xu, Canadell, Thompson, Winiwarter, et al., 2020). While nearly all microorganisms involved in the biogeochemical nitrogen cycle can potentially produce N₂O, specific microbial pathways including heterotrophic denitrification, NH₃ oxidation, and nitrifier denitrification, are pivotal to N₂O production (Butterbach-Bahl et al., 2013b; Kuypers et al., 2018b). Heterotrophic denitrification is a multistep respiration process that involves the reduction of oxidized mineral forms of nitrogen (i.e. NO₃⁻ and NO₂⁻) to gaseous nitric oxide (NO), N₂O, and dinitrogen (N₂). This process typically occurs under anaerobic conditions, although recent discoveries have identified a new group of aerobic denitrifying bacteria (J. Yang et al., 2020b). Conversely, NH₃ oxidation and nitrifier denitrification occur under aerobic conditions. Nitrification involves a two-step process: the oxidation of NH₃ to NO₂⁻ by AOB and AOA, followed by further oxidation to NO₃⁻ by NO₂⁻-oxidizing bacteria. N₂O is indirectly produced through the chemical decomposition of intermediate or end products of NH₃ oxidation (hydroxylamine, nitroxyl hydride, or NO₂⁻) (S. Liu et al., 2017b). Additionally, certain AOB can directly produce N₂O through nitrifier denitrification by oxidizing NH₃ to NO₂⁻ and subsequently reducing it to NO and N₂O (Kozłowski et al., 2016b; Wrage, Velthof, Van Beusichem, et al., 2001; Wrage-Mönnig et al., 2018b). In this study, we conducted a comprehensive quantitative assessment of N₂O emissions during forage conservation, especially in silage form. Simulated silages derived from three major silage crops in the United States—maize, alfalfa, and sorghum—were monitored for N₂O emissions over a 4-week period. Our findings showed significant N₂O release from silages, making forage conservation the third-largest source of N₂O emissions within the agricultural

sector. Further experiments confirmed the significant role of denitrification in N₂O production in conserved forages, as validated by molecular analyses.

3.2.2 Materials and methods

3.2.2.1 Forage sample preparation

Maize (Dyna Gro D53VC55RIB) and sorghum (Dyna Gro Super Sile 20) were harvested from a private farm near Manhattan, KS (39°38' N, 96°88' W) at their optimal maturity stages (i.e. 2/3 milk line and soft dough stages, respectively). The plants were chopped to a theoretical length of 2 cm using a standard forage harvester without inoculation. Two distinct alfalfa varieties, HVX MegaTron (WinField United L.L.C., Arden Hills, MN, USA) and HybriForce 3400 (Dairyland Seed Co.), were grown in experimental plots at the Kansas State University Agronomy Research Farm in Manhattan, Kansas (39°20' N, 96°59' W). Each variety was harvested at two different stages of maturity (mid-bud and early flowering) using a sickle bar mower. The harvested alfalfa was subsequently chopped to a length of 2 cm using a stationary forage chopper.

3.2.2.2 Experimental design

We used a simulated silage model known as mini silos. These mini-silos comprised a 1-L glass jar (Contreras-Govea et al., 2011) connected to a 3-L Tedlar bag (**Fig. 4**). Each crop was ensiled with four treatments: (i) no inoculant (control, I⁻), (ii) crop-specific commercial silage inoculant (brand names omitted for confidentiality) following the manufacturer's instructions (I⁺), (iii) inoculant + chlorate (0.1%, w/w, potassium salt) (I⁺ Ch⁺), and (iv) inoculant + acetate (0.1%, w/w, sodium salt) (I⁺ Ac⁺). Chlorate was added to inhibit denitrifiers (Kučera, 2006; Oremland & Capone, 1988), and acetate was added as a readily available external carbon source to increase the initial C/N ratio (Itokawa et al., 2001; Kampschreur et al., 2009). And for the optimization of the chlorate concentration and finding the economical method, each crop was ensiled with seven

different chlorate concentrations (0, 0.002, 0.005, 0.01, 0.05, 0.1, and 0.5%) to optimize the chlorate concentrations. Also, each crop was ensiled with 0.002% of chlorate and three different extra carbons (acetate, glucose, molasses) to find the best combination.

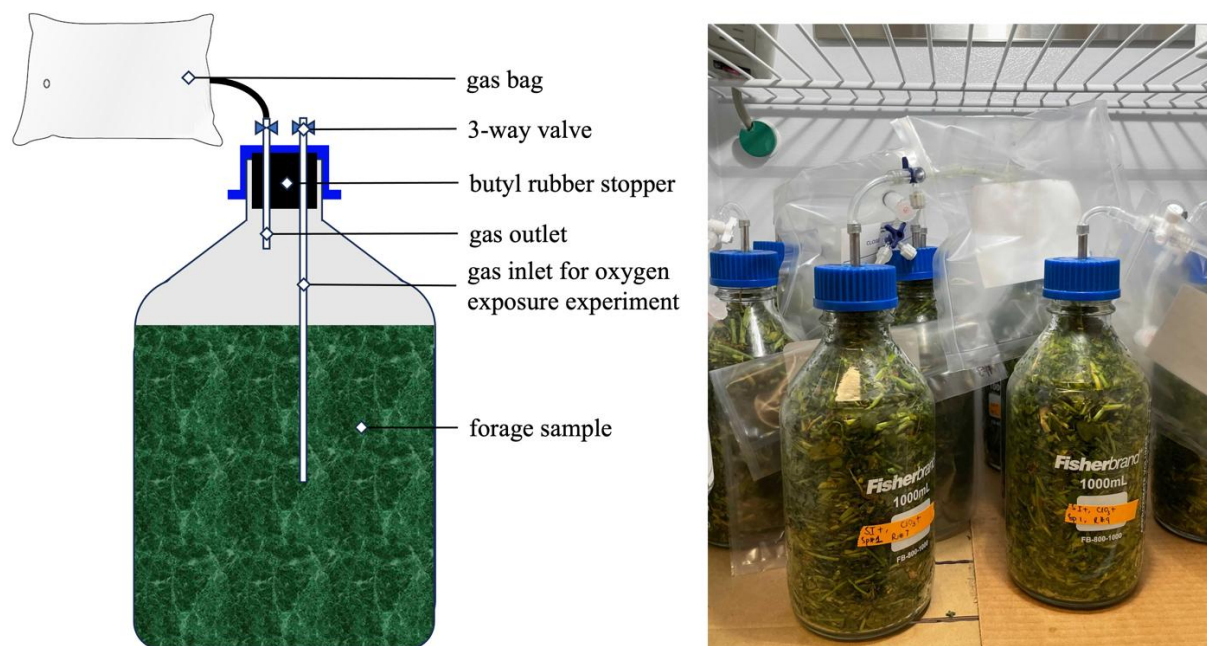


Figure 4. A schematic (left) and photo (right) of mini-silos used in this study. Gas inlet was only used for the oxygen exposure experiment.

Separate inoculant, chlorate, and acetate solutions were prepared and sprayed onto the plants. The initial moisture content was measured using the conventional microwave oven method (Lacerda et al., 2009) and adjusted to 70% (w/w, wet weight basis) by adding deionized water to the treatment solutions. All crops were packed in mini-silos with a bulk density of 650 kg/m^3 (41 lb/ft^3) (Darby & Jofriet, 1993), and each silo contained 650 g of the crops (wet weight), which was equivalent to 195 g_{DM}. The mini-silos were incubated at 30 °C in the dark, and N₂O production was monitored regularly for up to 4 weeks. Gas bags were removed on days 1, 2, 3, 5, and 20 to characterize temporal variations in the volume and composition of gas production and

replaced with new bags at each sampling. Fresh feed samples were collected and stored at -80°C for chemical analysis. Additionally, for molecular analysis, four mini-silos were prepared for each treatment and sacrificed on days 1, 3, 5, and 20. A subset of the samples (5 g) was preserved in 5 mL RNA preservation solution (RNAprotect, QIAGEN, Hilden, Germany) and stored at -80°C . Each treatment comprised three replicates of the mini-silos.

3.2.2.3 Contribution of nitrification to N_2O emissions

To investigate the relative contribution of nitrification to N_2O production, mini-silos were prepared with alfalfa (HVX Megatron) harvested at the early flowering stage, and acetylene (10 Pa) (Jia & Conrad, 2009) was added at the beginning of incubation. In addition, the mini-silos were intermittently exposed to oxygen on days 3, 5, and 10, achieved by supplying 20 mL of air (4 mL of oxygen) through a stainless-steel tube that reached the center of the mini-silos (**Fig. 5**). Oxygen was added to separate bottles on each injection date.

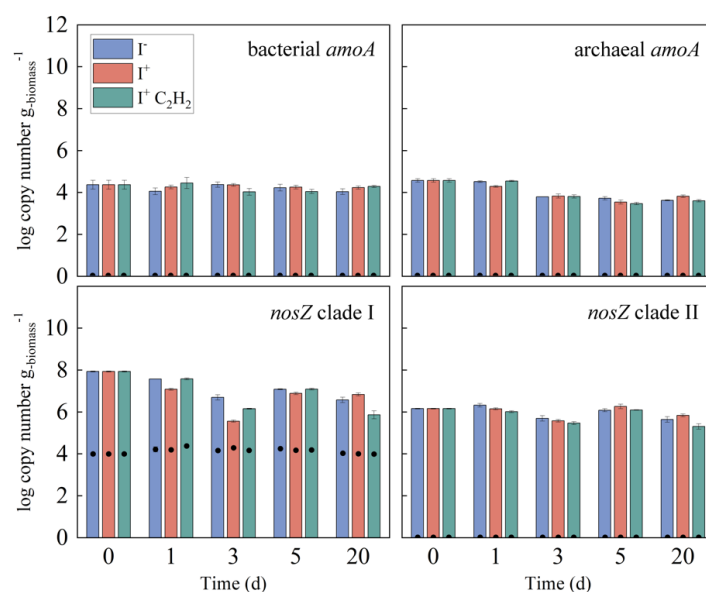


Figure 5. Effect of acetylene (C_2H_2) on the abundance of bacterial/archaeal *amoA* genes and clade I/II *nosZ* genes (bar) and their transcripts (black circle). The error bars represent the standard deviations of the triplicate incubations. Error bars may not be visible if their magnitude is smaller than the symbols.

3.2.2.4 Analytical methods

The total gas production volume was measured using the water displacement method. N₂O was quantified using an Agilent 7890 gas chromatograph with an electron capture detector and an HP-PLOT/Q column (30 m × 0.53 mm × 40 m). Fresh forage samples were sent to Rock River Laboratory, Inc., Watertown, WI, USA for nutritional analysis. The following parameters were examined: crude protein (CP), total amino acid, NH₃-N content, ADF, amylase-treated neutral detergent fiber (aNDF), lignin, starch, and ethanol-soluble carbohydrate (ESC). NO₃⁻-N content (% of DM) was measured at the Kansas State University Soil Testing Laboratory.

3.2.2.5 Nucleic acid extraction and quantitative PCR

Microbial nucleic acid extraction from plant material, particularly from the epiphytic phyllosphere, poses challenges due to plant derived biomolecules such as proteins and nucleic acids (Dent et al., 2004; Ikeda et al., 2009). In this study, we used a microbial DNA extraction method optimized for silage samples, as detailed in our previous study (S. Yang et al., 2023). Briefly, for DNA extraction, five grams of the sample was suspended in 45 mL of sterile 0.85% NaCl solution, shaken on a rotary shaker at 120 rpm for 2 h at room temperature, filtered through two layers of gauze cloth to remove large plant debris, and centrifuged at 12,000 g for 15 min at 4°C. The supernatant was discarded, and the pellet was stored at -80°C until subsequent DNA extraction. The RNA preservation solution containing the sample was shaken on a rotary shaker at 120 rpm for 2 h at room temperature. Four milliliters of the supernatant were collected, pelleted by centrifugation at 12,000 g for 15 min at 4°C, and immediately subjected to RNA extraction. DNA and RNA extractions from the pellets were performed using the DNeasy and RNeasy PowerSoil kits (Qiagen, Hilden, Germany), following the manufacturer's instructions, with slight modifications. Cells were lysed by bead beating at 20°C for 5 min. RNA samples

were subjected to DNase treatment (ezDNaseTM, Invitrogen, CA, USA) and reverse transcribed into complementary DNA (cDNA) using SuperScript^{IV} Reverse Transcriptase (Thermo Fisher Scientific, Waltham, MA, USA). PCR was conducted with universal 16S rRNA gene primers on DNase-treated RNA samples to confirm the absence of DNA contamination. Target genes and transcripts were quantified by qPCR using a CFX Opus 96 Real-Time PCR System (Bio-Rad, Hercules, CA, USA) in 20 μ L reaction mixtures containing 10 μ L of SsoAdvancedTM Universal SYBR Green Supermix (Bio-Rad, Hercules, CA, USA), 300 nM of each primer, and 2 μ L of template DNA. Quantification was performed using standard curves prepared from serial 10-fold dilutions of cloned plasmids or double-stranded synthetic DNA fragments (gBlocks[®], Integrated DNA Technologies, Coralville, IA, USA) in triplicate. Detailed information on primer sequences, standard sequences, and detection limits can be found in **Table 4**.

3.2.2.6 Statistical analysis

Statistical analyses were conducted using the MIXED procedures of SAS/STAT software, Version 9.4 (SAS Institute Inc., Cary, NC, USA). A two-way ANOVA was applied to the N₂O emission data, crop, treatment, and their interaction term as fixed effects. A Bonferroni multiplier adjustment was performed when comparing N₂O emissions between crops in each treatment group or treatments in each crop group (conditional pairwise comparison). Probability values of $P < 0.05$ (two-tailed) were considered statistically significant for all comparisons. The correlation coefficients between N₂O production and nutrient variables were calculated as Pearson correlation coefficients. A Pearson correlation coefficient > 0.8 or < -0.8 with $P < 0.05$ (two-tailed) was deemed indicative of a strong correlation between the two variables.

Table 4. Quantitative PCR primers, standards, and detection limits of each assay.

Target gene	Primer set	Primer sequences (5'-3')	Quantification range
16S rRNA (Amann et al., 1990)	1055F 1392R	ATGGCTGTOGTCAGCT ACGGGCGGTGTGTAC	10 ² -10 ⁹
<i>napA</i> (Henry et al., 2008)	V67m V17m	AAYATGGCVGARATGCACCC GRTRAARCCCATSGTCCA	10 ² -10 ⁹
<i>narG</i> (Bru et al., 2007)	narG-F narG-R	TCGCCSATYCCGGCSATGTC GAGTTGTACCAGTCRGC SGAYTCSG	10 ² -10 ⁹
<i>nirK</i> (Wei et al., 2015)	nirKC1-F nirKC1-R	ATGGCGCCATCATGGTNYTNCC TCGAAGGCCTCGATNARRTTRTG	10 ⁴ -10 ⁹
<i>nirS</i> (Braker et al., 1998)	nirS-1F nirS-6R	CCTAYTGGCCGCCRCART CGTTGAACTTRCCGGT	10 ³ -10 ⁹
<i>qnorB</i> (Braker & Tiedje, 2003)	qnorB-2F qnorB-5R	GGNCAYCARGGNTAYGA ACCCANAGRTGNACNACCCACCA	10 ² -10 ⁹
Clade I <i>nosZ</i> (Henry et al., 2006)	nosZ-1F nosZ-1R	WCSYTGTTCMTGACAGCCAG ATGTCGATCARCTGVKCRTTYTC	10 ² -10 ⁹
Clade II <i>nosZ</i> (Chee-Sanford et al., 2020)	nosZ 1374F nosZ 1577R	CCBYTBCAYACSCARTTYG TGSGASAGCTTGTTTSAGSS	10 ² -10 ⁹
Bacterial - <i>amoA</i> (Rotthauwe et al., 1997)	amoA-1F amoA-2R	GGGGTTTCTACTGGTGGT CCCCTCKGSAAAGCCTTCTTC	10 ² -10 ⁹
Archaeal - <i>amoA</i> (Francis et al., 2005)	arch-amoAF arch-amoAR	STAATGGTCTGGCTTAGACG GCGGCCATCCATCTGTATGT	10 ² -10 ⁹

Abbreviations for degenerate nucleotide positions are as follows: R = A or G; K = G or T; M = A or C; S = C or G; W = A or T; Y = C or T; B = C, G, or T; D = A, G, or T; V = A, C, or G; H = A, C, or T. References for the primers and probes are included in the first column.

3.2.3 Results

3.2.3.1 Chemical properties of silage materials

The freshly chopped, non-inoculated plant materials showed characteristic nutritional properties for each crop (**Fig. 6**). Among them, alfalfa, a leguminous crop, showed notably higher concentrations of proteins and amino acids than maize and sorghum. Conversely, cereal crops like maize and sorghum exhibited a higher starch content than alfalfa. The total protein and amino acid contents in alfalfa decreased as it matured, accompanied by an increase in fiber (acid detergent fiber [ADF] and neutral detergent fiber [aNDF]) and lignin contents. Notably, the

alfalfa variety HybriForce 3400 consistently showed higher protein and amino acid contents than HVX MegaTron, regardless of maturity stage.

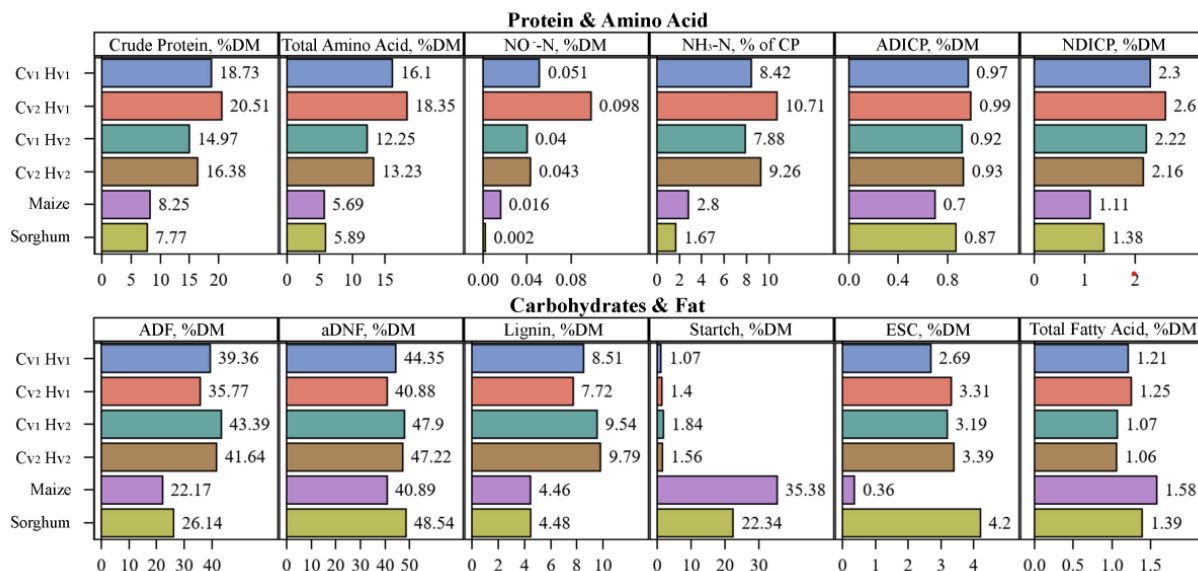


Figure 6. Nutritional parameters of fresh maize, alfalfa (two distinct varieties harvested at two different stages of maturity), and sorghum samples. The labels Cv₁ and Cv₂ denote two alfalfa varieties, HVX MegaTron and HybriForce 3400, respectively. The labels Hv₁ and Hv₂ denote alfalfa samples harvested at the mid-bud and early flowering stages, respectively. DM: dry matter; CP: crude protein; ADICP: acid detergent-insoluble CP; NDICP: neutral detergent-insoluble CP; ADF: acid detergent-treated fiber; aNDF: amylase-treated neutral detergent fiber; ESC: ethanol-soluble carbohydrates. The numerical annotations next to the individual bars are their respective measured values.

3.2.3.2 N₂O emissions from simulated silage

The total amount of N₂O produced in the maize, alfalfa (HybriForce 3400, harvested at mid-bud stage, Cv₂ Hv₁), and sorghum silage over 28 d of incubation was 6.7 (±0.7), 62.3 (±4.0), and 1.8 (±0.1) mL, which corresponded to 18.2 (±1.9), 169.7 (±10.9), and 4.8 (±0.2) g CO₂ equivalent (eq.) per kgDM, respectively (**Fig. 7a**). N₂O emissions began immediately after incubation commenced, with the majority (>90%) being produced within 24 h for maize and sorghum and 5 days for alfalfa. For alfalfa harvested at the same growth stages, no significant difference in N₂O production was observed between the varieties ($P > 0.05$) (**Fig. 7b**). However, incubations with

alfalfa harvested at the later maturity stages (i.e. early flowering stage, Hv2) produced significantly lower N₂O emissions ($P < 0.05$) regardless of the varieties (**Fig. 7b**). The statistical analysis is summarized in **Fig. 8**.

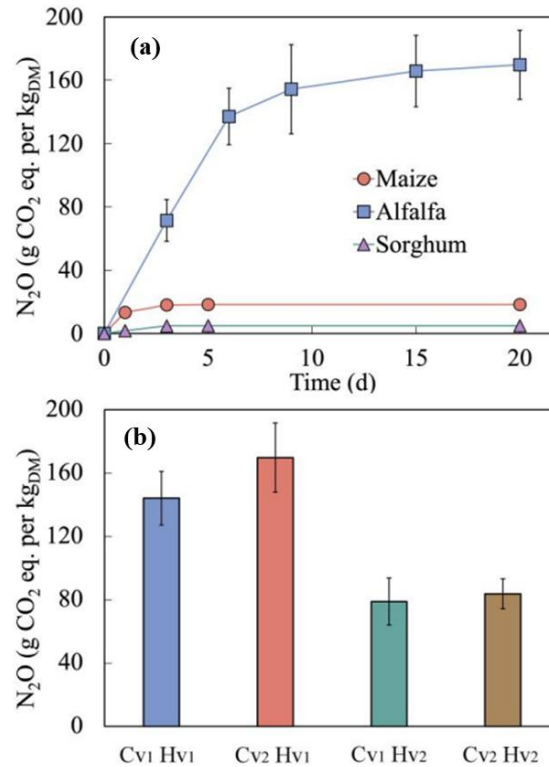


Figure 7. Cumulative N₂O production of (a) maize, alfalfa, and sorghum and (b) two distinct alfalfa varieties harvested at two different stages of maturity. The labels Cv₁ and Cv₂ denote the two alfalfa varieties, HVX MegaTron and HybriForce 3400, respectively. Similarly, the Hv₁ and Hv₂ denote alfalfa samples harvested at the mid-bud and early flowering stages. The error bars represent the standard deviations derived from the triple incubations. Error bars may not be visible if their magnitude is smaller than the symbols.

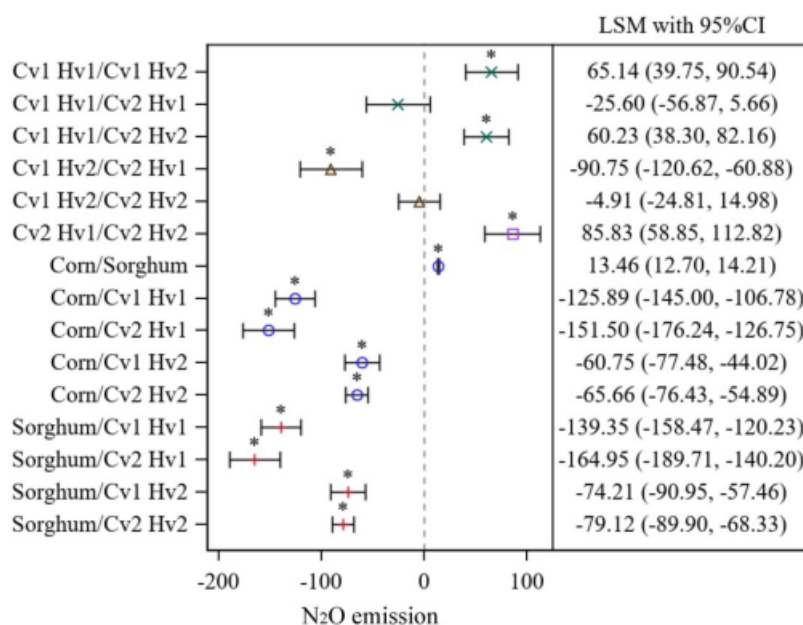


Figure 8. Pairwise comparisons N_2O emissions for crops under the controls (I^-) group. Results are reported as least square means for N_2O emissions with 95% confidence intervals. ‘*’ indicates a significant difference observed between the two crop groups.

3.2.3.3 Effects of different treatments on N_2O production

The effects of various treatments on N_2O production were examined within the same sample group, revealing consistent trends regardless of the crops or alfalfa varieties harvested at different growth stages (**Fig. 9a**). Compared to the controls (I^-), the addition of inoculants (I^+) had no significant effect on N_2O production ($P > 0.05$), except for sorghum, where a significant difference was observed ($P < 0.05$). Notably, applying chlorate treatment and the inoculant ($I^+ Ch^+$) significantly reduced N_2O production by up to 99%. The addition of acetate as an external carbon source alongside the inoculant ($I^+ Ac^+$) resulted in a statistically insignificant but numerically lower N_2O production than I^+ ($P > 0.05$). However, the effect of acetate on N_2O reduction was significant in alfalfa Cv2 Hv1 and sorghum ($P < 0.05$). Conversely, neither acetylene (C_2H_2) addition nor intermittent O_2 exposure at different periods (days 3, 5, and 10) impacted N_2O production (**Fig. 9b**). Lower chlorate concentrations, as low as 0.01% (w/w), still

achieved 92% N₂O reduction (**Fig. 9c**). Residual chlorate could not be quantified due to technical limitations in ion chromatography. The statistical analysis is summarized in **Table 5**.

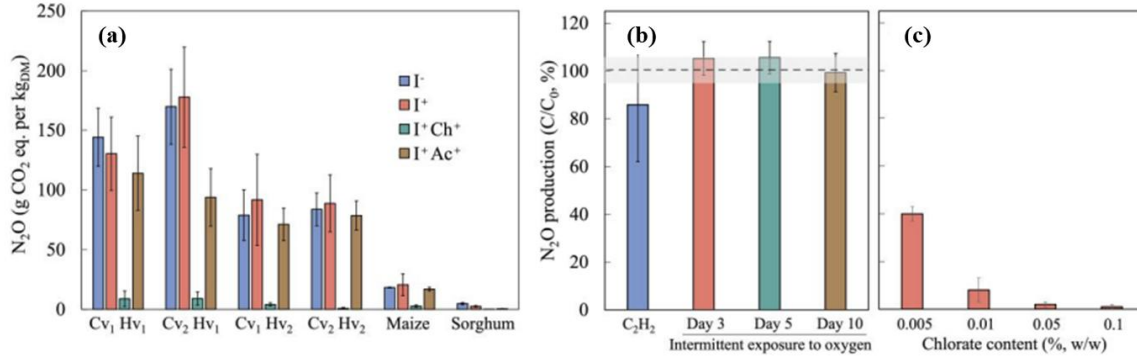


Figure 9. Cumulative N₂O production of (a) maize, alfalfa, and sorghum and (b) two distinct alfalfa varieties harvested at two different stages of maturity. The labels Cv₁ and Cv₂ denote the two alfalfa varieties, HVX MegaTron and HybriForce 3400, respectively. Similarly, the Hv₁ and Hv₂ denote alfalfa samples harvested at the mid-bud and early flowering stages. The error bars represent the standard deviations derived from the triple incubations. Error bars may not be visible if their magnitude is smaller than the symbols.

Table 5. Least square means for N₂O emissions (g-CO₂ eq. per kgDM-forage) with standard error (SE) of the two-way analysis of variance (ANOVA) model under variance heterogeneity.^{1,2,3}

Crops	I ⁻	Treatments		
		I ⁺	I ⁺ Ch ⁺	I ⁺ Ac ⁺
Cv ₁ Hv ₁	144.14 ± 10.90 ^a	130.33 ± 10.90 ^a	8.90 ± 10.90 ^b	114.07 ± 10.90 ^a
Cv ₂ Hv ₁	169.74 ± 10.90 ^a	177.72 ± 10.90 ^a	9.01 ± 10.90 ^c	93.84 ± 10.90 ^b
Cv ₁ Hv ₂	78.99 ± 7.77 ^a	91.82 ± 7.77 ^a	4.15 ± 7.77 ^b	71.23 ± 7.77 ^a
Cv ₂ Hv ₂	83.91 ± 7.77 ^a	88.79 ± 7.77 ^a	0.83 ± 7.77 ^b	78.71 ± 7.77 ^a
Maize	18.24 ± 1.87 ^a	20.57 ± 1.87 ^a	2.70 ± 1.87 ^b	16.96 ± 1.87 ^a
Sorghum	4.79 ± 0.22 ^a	2.46 ± 0.22 ^b	0.07 ± 0.22 ^c	0.52 ± 0.22 ^c

^{a, b, c} Means in the same row with different superscript letters differ significantly ($P < 0.05$).

¹Treatments are no inoculant (I⁻), crop-specific commercial silage inoculant (I⁺), inoculant and chlorate (I⁺Ch⁺), and inoculant and acetate (I⁺Ac⁺). The notations Cv₁ and Cv₂ indicate two alfalfa cultivars, HVX MegaTron and HybriForce 3400, respectively. The notations Hv₁ and Hv₂ indicate samples harvested at two different maturity stages, i.e., mid-bud and early-flowering, respectively.

²Results are reported as least square means ± standard error (SE) for each combination group of crop and treatment.

³P values of two-way analysis of variance (ANOVA) under variance heterogeneity for crop, treatment, and crop by treatment interaction are < 0.01, <0.01, and <0.01, respectively.

3.2.3.4 Analysis of the correlation between N₂O production and fresh matter nutrient parameters

The relationship between various nutrient parameters and N₂O production in the controls (I⁻) was measured using a Pearson correlation coefficient (**Fig. 3**). Parameters related to protein and amino acids, including crude protein ($r = 0.98$, $P = 0.031$), total amino acids ($r = 0.99$, $P = 0.001$), NO₃⁻ - N ($r = 0.94$, $P = 0.017$), NH₃ - N ($r = 0.96$, $P = 0.010$), and NDICP ($r = 0.92$, $P = 0.029$), exhibited strong correlations with N₂O production ($r > 0.8$ and $P < 0.05$). Conversely, parameters related to carbohydrates and fats, including ADF, aNDF, lignin, starch, ethanol-soluble carbohydrates, and total fatty acids, exhibited no significant correlation with N₂O production ($r < 0.4$ or $P > 0.05$).

3.2.3.5 Gene and transcript abundance dynamics

The abundance of the genes and transcripts in incubations with alfalfa harvested at the early flowering stage (Hv₂) under various treatments is summarized in **Fig. 11**. Notably, the abundance of *narG*, the gene encoding membrane-bound NO₃⁻ reductase, exhibited a marked two-order-of-magnitude decrease with the addition of chlorate (I⁺ Ch⁺), in which N₂O production was reduced by up to 99%, compared to I⁺. The abundance of other denitrification genes showed no trend across the various treatments over time. Notably, the abundances of bacterial and archaeal *amoA* gene encoding AMO were lower than those of denitrification genes, and no trend was observed over time across the various treatments. Transcript analysis revealed the expression of *narG* was completely suppressed by the addition of chlorate (I⁺ Ch⁺), similar to the gene abundance. Transcripts of archaeal and bacterial *amoA*, both *nir* genes encoding NO₂⁻ reductase, and *nosZ* gene encoding N₂O reductase clade II were not detected. The expression level of *napA* gene

encoding periplasmic NO_3^- reductase, *norB* gene encoding nitric oxide reductase, and *nosZ* gene encoding N_2O reductase clade I were not affected by the treatments used.

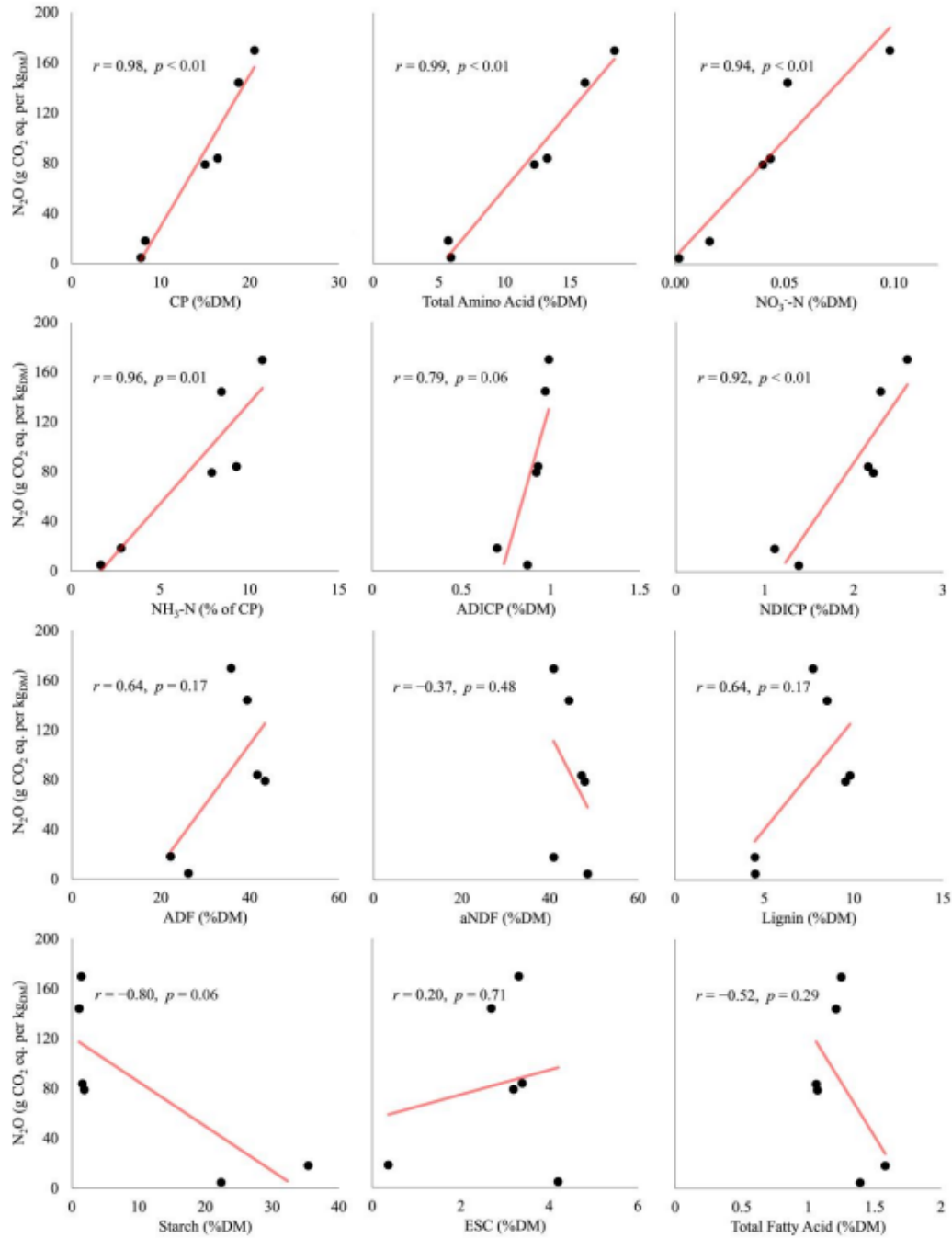


Figure 10. Correlation analysis of N_2O production and fresh matter nutrient parameters. DM, dry matter; CP, crude protein; ADICP, acid detergent insoluble CP; NDICP, neutral detergent insoluble CP; ADF, acid detergent treated fiber; aNDF, amylase-treated neutral detergent fiber; ESC, ethanol-soluble carbohydrates

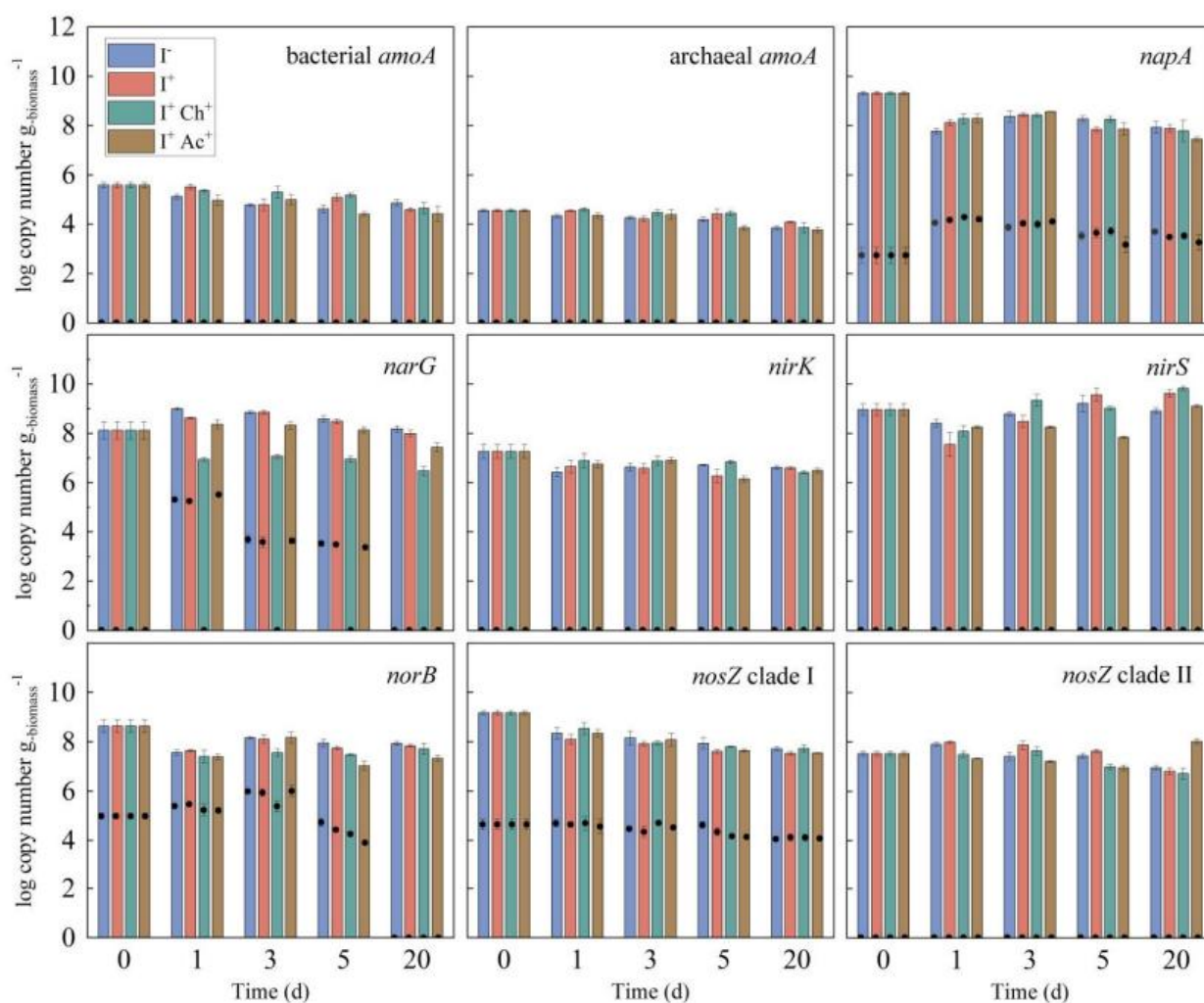


Figure 11. Abundance of functional genes (bar) and their transcripts (black circle) in incubation with alfalfa harvested at the early flowering stage (Hv₂) under various treatments. The error bars represent the standard deviations of the triplicate incubation. Error bars may not be visible if their magnitude is smaller than the symbols.

3.2.3.6 Optimization of chlorate and additional carbon sources combination

The cumulative N₂O produced from alfalfa silage controls was 16.6 ± 2.3 mL, equivalent to 45.2 g CO₂-eq kg⁻¹ dry matter (DM). Chlorate sharply suppressed N₂O in a dose-dependent manner: 0.01% (w/w) achieved a 92% reduction, while 0.005% reduced N₂O by 60% (**Fig. 12**). To evaluate potential co-benefits with added carbon and to retain an interpretable dynamic range, 0.005% chlorate was selected as the working dose for combination tests. At this dose,

supplemental glucose or acetate increased the inhibition from 62.9% (chlorate alone) to 79.8% and 85.0%, respectively, on Day 1 and maintained ~80% inhibition through Day 20. In contrast, molasses did not improve inhibition relative to chlorate alone.

Selected concentration for combination with carbon sources

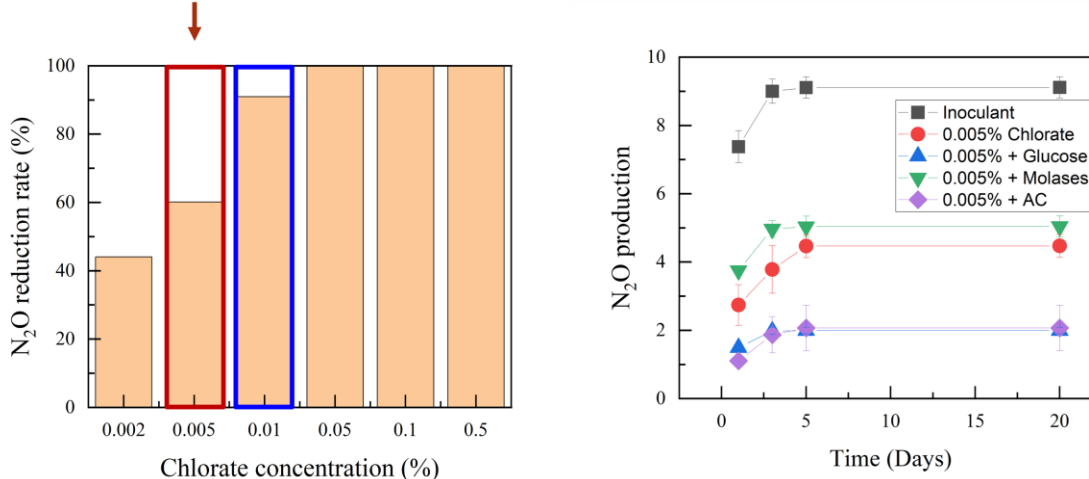


Figure 12. N₂O reduction rate for different chlorate concentrations

3.2.4 Discussion

3.2.4.1 N₂O production mechanism

The majority of N₂O production in this study occurred within the first week of incubation (**Fig. 7a**), which is consistent with previous studies (Krommweh et al., 2020b; Schmithausen et al., 2022). During this initial phase, transient aerobic or microaerobic conditions can be expected due to residual oxygen, followed by anaerobic conditions (Ávila & Carvalho, 2020). Under these conditions, both ammonia oxidizers and denitrifiers are potential contributors to N₂O production (Barnard et al., 2005; Schinner et al., 1996). Ammonia oxidizers are recognized for producing N₂O through direct (i.e. nitrifier denitrification) and indirect (i.e. abiotic processes) mechanisms. Furthermore, ammonia oxidizers can contribute to N₂O production by converting NH₃ to NO₃⁻, subsequently fueling denitrification. If nitrification contributes to N₂O production, we

hypothesize that additional O₂ could stimulate N₂O production since nitrification activity is dependent on residual O₂ levels. However, N₂O production did not increase when O₂ was added at different time points ($P > 0.05$, **Fig. 9b**), indicating that the contribution of nitrification to N₂O production may not be significant. This finding is further supported by the observation that C₂H₂ (10 Pa), a potent inhibitor of bacterial and archaeal NH₃ oxidation (Berg et al., 1982; Jia & Conrad, 2009), did not affect N₂O production ($P > 0.05$, **Fig. 9b**). The absence of bacterial and archaeal *amoA* transcripts also confirmed the noninvolvement of ammonia oxidizers in N₂O production (**Fig. 11**). At high concentrations (1 – 20 kPa), C₂H₂ has been shown to inhibit N₂O reductase activity of denitrifying micro-organisms (Knowles, 1996; Yoshinari et al., 1977). However, such inhibition (i.e. increased N₂O production) was not observed in this study (**Fig. 9b**). The expression of Clade I *nosZ* remained unaffected in the samples supplemented with C₂H₂ (**Fig. 5**), suggesting that C₂H₂ did not inhibit denitrifying bacteria at the concentration used in this study (i.e. 10 Pa). Furthermore, although recent studies have demonstrated that active nitrification can occur at pH levels as low as 3.0 (Norton & Stark, 2011), nitrification is typically limited in acidic conditions due to the unavailability of substrate (NH₃) for AMO, the enzyme catalyzing NH₃ oxidation. In silage, low pH caused by organic acids generated during fermentation may also inhibit nitrification. Due to the chemical similarities between NO₃⁻ and chlorate, chlorate has been used as an inhibitor for dissimilatory NO₃⁻ reduction, the first step in denitrification (Kučera, 2006; Oremland & Capone, 1988). The significant decrease in N₂O production, by up to 99% upon the addition of chlorate (**Fig. 9a**), indicates that denitrifiers are the main contributors to N₂O production. This finding was further confirmed by the diminished abundance of the *narG* genes and transcripts in the chlorate-amended samples (**Fig. 11**). In another study, the same concentration of chlorate (0.1% w/w) was used as a ruminant

supplement to reduce *E. coli* O157:H7 population (Callaway et al., 2002), suggesting that chlorate has the potential to be used as a silage additive to reduce N₂O emissions. Further studies are warranted to assess the potential hazards, such as the ultimate fate of the added chlorate and its impact on animal health if it remains in the silage. Denitrification is a microbial process wherein NO₃⁻ is reduced to N₂ via intermediates including N₂O. Factors such as a low C/N ratio have been reported to lead to N₂O accumulation during denitrification (Itokawa et al., 2001; Kampschreur et al., 2009). When the C/N ratio is low, the amount of available organic carbon is insufficient to fully reduce NO₃⁻ to N₂ (J. Wang et al., 2023; W. Zhou et al., 2020). Moreover, low C/N ratios affect microbial community dynamics, i.e., a low C/N ratio can favor populations less efficient at the N₂O reduction step, further contributing to higher N₂O levels (Xu et al., 2024). Additionally, conditions with a low C/N ratio often lead to residual oxygen or higher NO₂⁻ concentrations, both of which inhibit N₂O reductase, exacerbating N₂O accumulation (W. Zhou et al., 2020). In our study, adding acetate as an external carbon source, effectively increasing the C/N ratio, resulted in a slight reduction in N₂O production (**Fig. 9a** and **Table 5**). This suggests that N₂O production in conserved forage may be influenced, at least in part, by a low C/N ratio. Overall, these findings suggest that denitrification inhibitors, such as chlorate, can be combined with an external carbon source, such as acetate, as an effective additive to mitigate N₂O emissions from the forage conservation process.

3.2.4.2 Chlorate combination with extra carbon sources

Chlorate is a structural analog of NO₃⁻ and selectively inhibits the membrane-bound NO₃⁻ reductase, blocking the first step of respiratory denitrification and preventing downstream N₂O formation. Parallel observation of suppressed *narG* expression aligns with this primary target. At a low operational dose (0.005%), residual N₂O persists—useful for detecting whether added

carbon shifts electron flow toward the nitrous oxide reductase and completion to N_2 . Both extra carbon sources (glucose and acetate) are readily bioavailable, low-complexity substrates that (i) elevate C: N, (ii) provide fast electron supply at low redox potential, and (iii) minimize metabolic overhead for denitrifiers. These traits favor complete denitrification and support *nosZ* activity, explaining the jump from ~63% to ~80–85% inhibition and its persistence through Day 20.

3.2.4.3 N_2O production and its relationship with nutritional parameters

N_2O production was closely correlated with most parameters related to protein and amino acids, including NO_3^- -N (**Fig. 10**), presumably the main source of N_2O production. The NO_3^- -N content in forages varies with the stage of plant maturity (Murphy & Smith, 1967), and both alfalfa varieties harvested at later growth stages, which produced significantly less N_2O ($P < 0.05$) (**Fig. 7b**), contained lower NO_3^- -N levels (**Fig. 6**). Reports have also demonstrated that nitrogen fertilization directly impact NO_3^- -N content (Murphy & Smith, 1967), implying that nitrogen fertilization, especially preceding harvest, may contribute to higher N_2O production at ensiling. Further studies are needed to investigate the effects of nitrogen fertilization schedule on N_2O production. Simple changes in agricultural practice may reduce N_2O emissions. A source of organic carbon is an important component of denitrification, serving as an electron donor. Many studies have shown that external carbon sources, such as methanol, ethanol, and acetate, stimulate denitrification and usually reduce N_2O production (Fu et al., 2022). Consistent with these findings, our study showed that the addition of acetate reduced N_2O production (**Fig. 9a**). However, carbohydrate-related parameters, such as ADF, aNDF, lignin, starch, and ethanol-soluble carbohydrates, did not correlate with N_2O production (**Fig. 10**), which could be due to the recalcitrance of these carbon sources. Similarly, denitrification was promoted by plant-based

carbon substrates, such as rice straw, but there was a significant lag before denitrification became active (B. Zhou et al., 2019).

3.2.4.4 Environmental implications

What gets measured gets managed. The first step in reducing GHG emissions is to measure them. The US EPA publishes an annual report titled “Inventory of US Greenhouse Gas Emissions and Sinks,” which estimates total GHG emissions by source across all sectors of the economy at the national level (1). Notably, agriculture is the largest contributor to N₂O emissions in the United States, accounting for 80% in 2021. The EPA monitors major sources in the agricultural sector, including agricultural land management, manure management, and the field burning of agricultural residues (**Table 1**) (USEPA, 2023;). In our study, 18.2 ($\pm$ 1.9), 169.7 ($\pm$ 10.9), and 4.8 ($\pm$ 0.2) g CO₂ eq. per kg_{DM}-forage of N₂O were produced from maize, alfalfa, and sorghum, respectively (**Fig. 7a**). These values correspond to 2.3, 9.1, and 0.7 mg-N₂O-N/g-N based on the assumption that the total nitrogen content in silage is 16% of the protein content (Collins & Owens, 2003). According to the Crop Production 2022 Summary of the United States Department of Agriculture National Agricultural Statistics Service, the total production volumes of maize, alfalfa, and sorghum for silage in 2022 were 128.6, 17.4, and 5.6 MMT, respectively, comprising 93% of the total silage production combined (USDA, 2023b). Assuming a similar amount of N₂O can be produced from each crop, the total N₂O emission potential amounts to 5.3 MMT CO₂ eq.. This makes forage conservation the third largest N₂O emitter in the agricultural sector, surpassing the field burning of agricultural residues by a factor of 30 (**Table 1**). Notably, N₂O emissions from silage of uncategorized crops (total production volume: 10.7 MMT in 2022, comprising 7% of the total silage production) were not considered in this study (USDA, 2023b). Again, the first step to reducing GHG emissions is to measure them, as policymakers and

decision-makers use GHG inventories to develop strategies and track progress in GHG emission reduction efforts (Gidden et al., 2023; Iyer et al., 2017).

3.2.4.5 Limitations and perspectives

While our assessment provides valuable insights, it is important to acknowledge its limitations, particularly when attempting to extrapolate the data to a national scale. The microbial N₂O production, like any other microbial process, is sensitive to various environmental factors such as temperature and nutritional parameters, which could result in underestimation or overestimation of the outcomes. There are additional uncertainties associated with the heterogeneity of farmers' practices, such as storage types (e.g. silo, bunker, bag), harvest time, inoculant use, moisture content (i.e. wilting), and oxygen exposure. Therefore, N₂O emission measurements from actual silage fermentation systems spanning a range of environmental and management variations worldwide are warranted to achieve more accurate estimation results. Additionally, we demonstrated that a simple treatment could significantly reduce N₂O emissions from silage. With that, we aim to underscore the critical importance of the silage process as a significant source of N₂O emissions, advocating for targeted research and intervention strategies to mitigate this environmental impact. The nitrogen cycle within our study system is complex, where multiple processes such as nitrification, nitrifier-denitrification, and denitrification may occur simultaneously. This complexity introduces potential sources of error in distinguishing the contributions of these concurrent N₂O-producing reactions. Further efforts are warranted to develop more refined methodologies that can accurately assess and differentiate the contributions of each individual process. Our study highlights the need for such advancements to enhance our understanding of N₂O emissions in silage systems. In this study, chlorate was proposed as a simple and effective remedy to significantly reduce N₂O emissions (**Fig. 9a**). Additional

experiments demonstrated that even at low concentrations, such as 0.01% (w/w), chlorate achieved a 92% reduction in N₂O emissions (**Fig. 9c**). The chlorate (as sodium salt) price at the end of 2023 in the United States was 795 USD per ton. The estimated cost to add 0.01% (w/w) chlorate as an additional silage additive is 0.08 USD per ton forage (dry weight), which is only approximately 5–8% compared to the silage inoculant cost (i.e. 1–1.5 USD per ton). Assessing the social cost of GHG emissions has become a common yardstick for estimating the benefits of formulating and implementing abatement policies (Aldy et al., 2021). The social cost estimates of N₂O shown in recent studies range from 16 to 174 USD per kg – N₂O – N (Kanter et al., 2021; Rennert et al., 2022; J. Wang et al., 2023). Assuming that chlorate (0.01%, w/w) is added to achieve 90% N₂O reduction (**Fig. 9c**), the social cost saving could be 81–882 million USD with the expected total cost of 12 million USD for applying chlorate as an additive.

Chapter 4 - Impact of BNI on methanotrophs

4.1 Background

Nitrogen use efficiency (NUE) of agricultural plants, defined as the yield of grain achieved per unit of nitrogen available to the crop from soil and applied fertilizer, is notoriously low.

Globally, about 50-70% of the nitrogen fertilizer applied to cropping systems is lost to the environment mainly due to microbial nitrification (Cassman et al., 2002; Ladha et al., 2005).

Nitrification, a key microbial process in the global nitrogen-cycle, transforming immobile NH_4^+ to highly mobile NO_3^- , is known to enhance losses of fertilizer nitrogen by leaching and denitrification (Coskun et al., 2017). Nitrification has been known to be a two-step process where NH_4^+ is first oxidized to NO_2^- by AOB and/or AOA, and subsequently to NO_3^- by NO_2^- -oxidizing bacteria, although recent studies demonstrated that *Nitrospira* spp. can carry out both reactions (Daims et al., 2015; van Kessel et al., 2015). Further, nitrification is one of the primary processes inducing N_2O emissions in soils (R. Liu et al., 2016), which is the third most important GHG, next to CO_2 and CH_4 .

For these reasons, there have been attempts to repress nitrification in agriculture. Since the mid-1950s, studies on nitrification inhibitors have been conducted, and hundreds of SNIs have been identified. However, these artificially synthesized nitrification inhibitors are not widely applied in practice, because (1) their biological stability is limited leading to shorter effective time of nitrification inhibition, (2) some of these artificially synthesized nitrification inhibitors have toxicity and adversely affect beneficial soil microorganisms, resulting in their limited usability, and (3) artificially synthesized nitrification inhibitors are in general expensive (Coskun et al., 2017; Subbarao, Ito, et al., 2006). On the other hand, despite many early studies, it has recently become apparent that certain plant species are capable of suppressing microbial

nitrification by releasing inhibitory phytochemicals from roots, which is termed BNI (Subbarao et al., 2009a; Sun et al., 2016; Tesfamariam et al., 2014). This phenomenon has emerged as a promising new area of research not only because of its potential to improve NUE in cropping systems (Coskun et al., 2017; Subbarao et al., 2013), but also its implications for agricultural soil management, i.e., reduced N₂O emissions (Byrnes et al., 2017; Peters et al., 2013; Subbarao et al., 2017). For example, N₂O emission was also suppressed by 90% in the field of high-BNI *Brachiaria* pasture compared with soybean (*Glycine max*), which lacks BNI capacity (Subbarao et al., 2009a). Since the first report of BNI in *B. humidicola*, BNIs have been discovered in several crops, including sorghum (*Sorghum bicolor* L.) and wheat (*Triticum aestivum* L.) (O'Sullivan et al., 2016a; Zakir et al., 2008a). More recently, one of the major cereal crop rice (*Oryza sativa* L.) exudate was reported as crops which can produce detectable BNI activity (Pariasca Tanaka et al., 2010).

The first step of nitrification, the oxidation NH₃ to NO₂⁻ via hydroxylamine (NH₂OH), is catalyzed by two enzymes: AMO and HAO. Many studies measure BNI activity using nitrification inhibition assay (Subbarao, Ishikawa, et al., 2006) which measures NO₂⁻ concentration to identify the activity of several BNI exuded from roots with *Nitrosomonas* spp. in pure culture. The mode of inhibitory action of the BNIs varies for each species. For example, recent research has demonstrated that MHPP in sorghum and 1,9-D in rice specifically block the AMO step in bacterial ammonia oxidation (Coskun et al., 2017; Pariasca Tanaka et al., 2010; Zakir et al., 2008a). Also, Brachialactone, which is the exudate of *B. humidicola*, inhibited both AMO and HAO but had a greater inhibitory effect on AMO than the HAO (Subbarao et al., 2007b).

Meanwhile, CH₄ can be oxidized by MOB, a.k.a., methanotrophs, the only group of bacteria that utilize CH₄ as their sole carbon and energy source (Hanson & Hanson, 1996). pMMO, catalyzing the first step of CH₄ oxidation, is evolutionarily related to AMO. In these distinctive groups of Gram-negative bacteria, pMMO and AMO are both membrane associated enzymes and share many similarities in their encoding genes (*pmoCAB* and *amoCAB* respectively), structure ($\alpha_3\beta_3\gamma_3$ and subunit composition known for pMMO and possibly for AMO), metal content (Cu and possibly Fe), substrate ranges and inhibition patterns (Arp et al., 2007; Hakemian & Rosenzweig, 2007), suggesting that certain BNI compounds may inhibit MMO as well as AMO (Holmes et al., 1995), leading to increased CH₄ release to atmosphere.

4.2 Unexpected inhibition of methanotrophy by BNI: Mechanistic insights and ecological implications

4.2.1 Introduction

Plant–microbe interactions are fundamental drivers of terrestrial ecosystems, shaping nutrient cycles and influencing global climate dynamics. One critical and environmentally relevant interaction occurs in the plant rhizosphere, where plants and microbes compete intensely for nitrogen. Despite heavy nitrogen fertilizer use in modern agriculture, approximately 50–70% of applied nitrogen is not utilized by crops, instead lost primarily due to nitrification (Cassman et al., 2002; Coskun et al., 2017)—a microbial process converting immobile NH₄⁺ to highly mobile NO₃[−], contributing to nitrogen loss through leaching and denitrification. This transformation ultimately increases emissions of N₂O, a potent GHG (USEPA, 2023) and primary ozone depleting substance (Ravishankara et al., 2009a).

To mitigate these losses, SNIs have been developed; however, they face limitations due to limited biological stability, potential toxicity to beneficial soil organisms, and high costs

(Coskun et al., 2017; Subbarao, Ito, et al., 2006). In contrast, recent research has identified a natural plant-based alternative: BNI, where plants release specific root exudates capable of suppressing nitrifying microbes (Coskun et al., 2017; Subbarao et al., 2009b, 2021; Subbarao & Searchinger, 2021). BNI represents a promising strategy to enhance NUE and reduce N₂O emissions (Byrnes et al., 2017; Peters et al., 2013; Subbarao et al., 2017). For instance, introducing a chromosomal segment from wild wheat into high-yielding cultivars significantly enhanced nitrogen uptake by 30% and reduced N₂O emissions by 25% without compromising grain yield and quality (Subbarao et al., 2021).

While BNI shows promise in improving agricultural sustainability (Coskun et al., 2017; Ghatak et al., 2023; Stein, 2023), its broader ecological level consequences remain insufficiently understood. Nitrification is a two-step process where NH₄⁺ is first oxidized to NO₂⁻ by ammonia-oxidizing microorganisms and subsequently to NO₃⁻ by NO₂⁻-oxidizing bacteria. The entire nitrification process is usually limited by the initial step, i.e., NH₄⁺ oxidation to NO₂⁻ via NH₂OH, which is sequentially catalyzed by AMO and HAO. Various BNI compounds from different chemical groups have been successfully isolated, and their mode of inhibition has been identified (Nardi et al., 2020; X. Wang et al., 2021). For example, 1,9-D from rice (*Oryza sativa*), MHPP from sorghum (*Sorghum bicolor*) specifically inhibit AMO (Sun et al., 2016b; Zakir et al., 2008b), similar to most SNIs (Coskun et al., 2017). In contrast, LA, derived from the pasture grass *B. humidicola*, inhibits both AMO and HAO (Subbarao et al., 2008).

Meanwhile, CH₄, another potent GHG, is oxidized by (MOB; hereafter referred to as methanotrophs), which serve as the sole biological CH₄ sink and play a critical role in mitigating CH₄ emissions (Hanson & Hanson, 1996). Given the chemical similarity between CH₄ and NH₄⁺, pMMO—the enzyme catalyzing the first step of CH₄ oxidation—is evolutionarily related to

AMO and shares extensive genetic, structural, and catalytic homology (Holmes et al., 1995). Consequently, many SNIs—including nitrapyrin (Topp & Knowles, 1984), thiourea, allylthiourea (Hubley et al., 1975), and acetylene (Bédard & Knowles, 1989)—also inhibit MOB activity. These parallels raise the possibility that naturally produced BNIs, similar to some SNIs, may unintentionally suppress CH₄ oxidation. Understanding the impact of BNI on controlling GHG emissions is thus crucial, especially in wetland and rice paddy ecosystems (Calvo Buendia et al., 2019) where plants, MOB, and ammonia oxidizers coexist. These environments significantly contribute to global GHG budgets, underscoring the importance of comprehensively evaluating BNI's coupled impacts on nitrogen and carbon cycling.

To address this critical knowledge gap, we undertook a multi-scale investigation into the effects of BNI on CH₄ oxidation. We first employed controlled plant growth chamber experiments to examine how BNI-producing plants influence methanotrophic activity and community structure. Subsequently, pure culture assays were conducted to directly evaluate the inhibitory effects of specific BNI compounds on MOB. To elucidate the underlying inhibition mechanisms, we combined molecular docking, dynamic simulations, and chemical analog experiments to pinpoint the interactions between BNIs and the key CH₄-oxidizing enzyme, pMMO. Finally, RNA-seq analysis was conducted to gain molecular-level insights into the transcriptional responses of MOB exposed to BNIs. Our findings reveal, for the first time, that BNIs can significantly suppress CH₄ oxidation activity in addition to their established role in inhibiting nitrification. This dual action provides mechanistic insight into how specific BNI compounds interfere with CH₄-oxidizing enzymes and highlights the need to assess the previously unrecognized impacts of BNI activity on CH₄ oxidation and its broader implications for coupled carbon and nitrogen cycling in terrestrial ecosystems.

4.2.2 Material and Methods

4.2.2.1 Soil characterization.

The soil mixture was composed of 14% sand, 58% silt and 28% clay, and classified as silty clay loam according to the United State of Department of Agriculture (USDA) system for soil taxonomy. The soil physico-chemical characteristics were pH of 7.3; TC of 1.43%; TN of 0.13%; TP of 390 ppm; NO_3^- -N of 8 ppm; NH_4^+ -N of 7 ppm; Ca of 3,111 ppm; Cu of 1 ppm; Mg of 131 ppm; Mn of 9 ppm; Na of 13 ppm; K of 258 ppm; Zn of 0.3 ppm; and Fe of 29 ppm.

4.2.2.2 Soil Column Experiment: Plant-Mediated Effects on Methanotrophy

Acrylic soil columns (10 in i.d., 22 in length) (**Fig. 13**) were fabricated from transparent tubing with airtight lids. Each column base was filled with a 2-cm layer of 3-mm glass beads to ensure uniform gas distribution. Soil was packed to a depth of 25 cm, yielding a bulk density of $1.9 - 2.0 \text{ g}\cdot\text{cm}^3$. A gas mixture of CH_4 and CO_2 (50:50 v/v) was delivered from the base at $5 \pm 0.5 \text{ mL}\cdot\text{min}^{-1}$, while humidified ambient air was continuously flushed into the headspace at $20 \text{ mL}\cdot\text{min}^{-1}$ to simulate natural diffusion and wind. Soil moisture was monitored at 5 cm depth with an EC-5 probe (Decagon Devices, Pullman, WA, USA), and an automated irrigation system maintained 45–55% water-holding capacity. Ambient temperature and headspace relative humidity were monitored by a wireless thermo-hygrometer (Model H5074001, Govee, Hong Kong). Lighting was supplied by LED strips mounted under the reactor lids, providing $200 \mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ photosynthetically active radiation under a 16 h light / 8 h dark cycle. Experimental treatments ($n = 3$ per condition) included: (1) control soil; (2) NH_4^+ amendment (25 mg NH_4^+ -N $\cdot\text{kg}^{-1}$ soil); and (3) NH_4^+ amendment with ten *Brachiaria* seedlings (purchased from Hancock Seeds, Dade City, FL, USA). Seedlings were transferred at the 3–4 leaf stage. After two months, soils were collected from the rhizosphere (10-15 cm depth). CH_4 oxidation kinetics were assayed

within 24 h. Subsamples (30 g wet weight) were incubated in 160 mL serum bottles with 1% CH₄ (1.6 mL headspace). Bottles were incubated at 30 °C in the dark. Headspace samples (100 µL) were withdrawn every 40 min with gas-tight syringes and analyzed on an Agilent 7890 GC-FID. CH₄ oxidation rates were calculated from depletion curves, and Michaelis-Menten parameters ($V_{\max}$, K_m) were fitted to the Michaelis-Menten model using OriginLab.

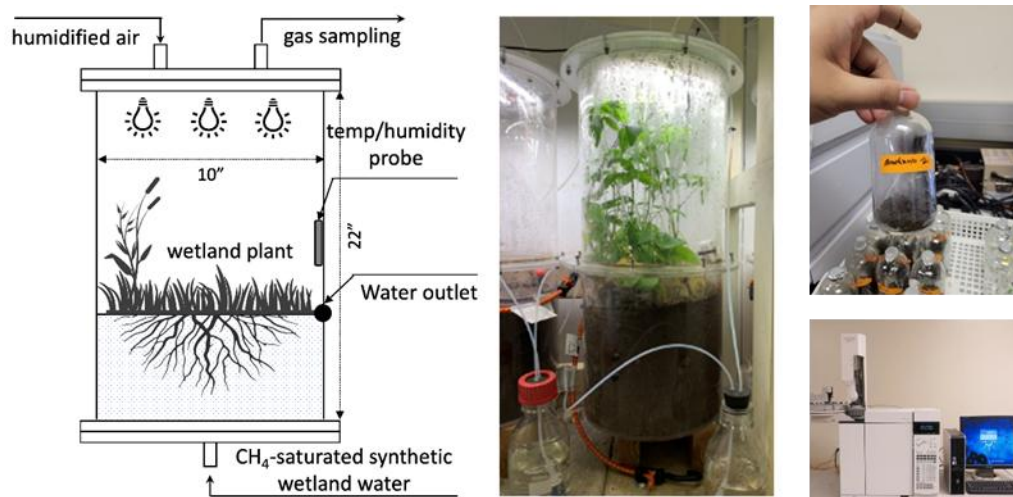


Figure 13 Acrylic soil columns to plant-mediated effects on methanotrophy and kinetic experiments set up.

4.2.2.3 Axenic culture experiment

BNI Compound Preparation.

Three BNIs were tested: 1,9-D (Activate Scientific, 95%), MHPP (Thermo Scientific, 98%), and LA (Thermo Scientific, 99%). Stocks were freshly prepared in DMSO (Thermo Scientific, 99%) and sterilized through 0.45-µm filters. Final solvent concentrations were 0.01% (methanotroph) or 0.1% (*N. europaea*). Additional experiments confirmed that these DMSO levels had no effect on growth, CH₄ oxidation, or NO₂⁻ formation. DMSO-containing controls served as appropriate baselines.

Direct Inhibition of Methanotrophy.

M. trichosporium OB3b and *M. album* BG8 were cultivated in NO_3^- mineral salts (NMS) medium supplemented with $10\ \mu\text{M}$ CuSO_4 at $30\ ^\circ\text{C}$. For induction of sMMO, cultures were subcultured twice on copper-free NMS agar before transfer into liquid medium. Cultures were grown in 1 L NMS medium in 4 L Erlenmeyer flasks under a CH_4/air atmosphere with shaking at 120 rpm. CH_4 was replenished whenever depleted. Growth continued until $\text{OD}_{600} \sim 0.3$ (exponential phase). Cells were harvested by centrifugation ($5,000 \times g$, 10 min, $20\ ^\circ\text{C}$), washed with fresh medium, and resuspended to $\text{OD}_{600} = 0.3$. For inhibition assays, 50 mL aliquots were transferred into 250 mL side-arm flasks for OD_{600} monitoring, sealed, and supplemented with 1% CH_4 (v/v). Inhibitor ranges were: 1,9-D ($0.29 - 5.74\ \text{mM}$), MHPP ($0.028 - 2.78\ \text{mM}$), and LA ($5.8 - 58.2\ \mu\text{M}$). Positive controls contained 0.01% DMSO without BNIs. CH_4 depletion was monitored by GC-FID, and oxidation rates were fitted to Michaelis-Menten curves with OriginLab. V_{max} values were normalized to total protein concentration (Bradford assay).

Comparative Sensitivity of *Nitrosomonas europaea*.

N. europaea (ATCC 19718) was grown in ATCC medium 2265 at $30\ ^\circ\text{C}$ in the dark with shaking (120 rpm). Cultures were harvested in exponential phase, centrifuged ($5,000 \times g$, 10 min, $20\ ^\circ\text{C}$), washed twice, and resuspended in fresh medium to $\text{OD}_{600} = 0.1$. Assays were performed in 96-well plates with negative controls (medium only) and positive controls (0.1% DMSO without BNIs). Cultures were exposed to varying concentrations of BNIs. Parallel plates were used for baseline and 1 h incubations. Ammonia oxidation was stopped by adding 0.1 M allylthiourea (AT). NO_2^- accumulation was quantified by the Griess method, reading absorbance at 548 nm with an Epoch2 Microplate Reader (BioTek). Each treatment was performed in triplicate. Inhibition (%) was plotted against inhibitor concentration, and IC_{50} values were estimated using OriginLab.

Reversibility

To evaluate whether BNI-mediated inhibition of methane oxidation was reversible, *M. trichosporium* OB3b cultures were exposed to BNI compounds at their respective IC₅₀ (MHPP and 1,9-D) or IC₄₀ (LA) concentrations under the same conditions described above. Methane oxidation kinetics were first measured in triplicate cultures to quantify inhibited activity ($V_{\max}$). Following this measurement, triplicate cultures were pooled prior to centrifugation ($5,000 \times g$, 10 min) to minimize variability associated with cell harvesting. Cells were washed once with fresh NMS medium to remove residual inhibitors, centrifuged again, and resuspended in fresh medium. The resuspended culture was then redistributed into triplicate serum bottles, amended with methane, and methane oxidation kinetics were re-measured as described above.

Copper-dependent inhibition assays

To assess whether inhibition was associated with pMMO, additional assays were conducted under copper-replete (+Cu) and copper-limited (–Cu) conditions to modulate expression of pMMO and sMMO, respectively. OB3b cultures used for copper-limited (–Cu) assays were preconditioned on Cu-free NMS agar plates prior to liquid cultivation to minimize residual copper carryover and facilitate sMMO expression (Lee et al., 2006). Copper-replete conditions were established using NMS medium supplemented with 10 μ M Cu, consistent with the conditions used in the kinetic assays, whereas copper-limited conditions were established using NMS medium without added Cu. Methane oxidation kinetics were then measured following exposure to BNI compounds at the same IC₅₀ or IC₄₀ concentrations as described above. $V_{\max}$ was measured for BNI-treated and control (BNI-free) cultures under both copper-replete (+Cu) and copper-limited (–Cu) conditions, and relative activity was calculated by

normalizing $V_{\max}$ values of BNI-treated samples to their corresponding control within each condition.

Molecular docking and dynamics simulations

The trimeric pMMO complex (PmoA–PmoB–PmoC) from *M. trichosporium* OB3b was modeled using both SWISS-MODEL(Waterhouse et al., 2018) and AlphaFold3(Abramson et al., 2024).

Structural templates were selected based on sequence coverage and model quality metrics, including Global Model Quality Estimation (GMQE) and Quaternary Structure Quality Estimation (QSQE). The resulting models were evaluated by comparison with the available crystal structure (PDB ID: 3CHX),(Tucci et al., 2023) and the model with the lowest root-mean-square deviation (RMSD) was selected for subsequent analyses. Molecular docking was performed to identify potential binding sites of BNI compounds. Docking simulations were conducted using SwissDock (Bugnon et al., 2024; Grosdidier et al., 2011) with the AutoDock Vina algorithm in blind docking model, (Eberhardt et al., 2021) allowing exploration of the entire protein surface exposed to the periplasm. Ligand structures were provided as SMILES inputs. Binding poses were ranked based on predicted binding energy (kcal mol^{-1}), and clusters of low-energy conformations were used to identify candidate binding regions. Binding sites that were inaccessible from the protein exterior, including internal cavities and membrane-embedded regions, were excluded from further analysis. Docking results were visualized and analyzed using PyMOL. (Schrödinger LLC, 2015) In addition to native BNI compounds, structural analogues of 1,9-D, including dodecanoic acid and 1-decanol, were selected to probe structure–activity relationships and evaluated using the same methane oxidation inhibition assay described above, under identical experimental conditions. IC_{50} values were determined by testing a range of concentrations and fitting the resulting data to the Michaelis–Menten model.

Nucleic acids extraction, qPCR, and RT-qPCR

Soil DNA was extracted using the DNeasy PowerSoil kit (Qiagen, Hilden, Germany). DNA and RNA from axenic cultures were extracted using the MagMAX Microbiome Ultra Kit (Applied Biosystems, USA) on the Duo Prime automated platform (Thermo Fisher). RNA was DNase-treated (TURBO DNase, Thermo Fisher) and quantified by microvolume spectrophotometry (Epoch2, Agilent Technologies). RNA was reverse-transcribed with SuperScript IV Reverse Transcriptase (Thermo Fisher). qPCR and RT-qPCR were performed on a CFX Opus 96 (Bio-Rad) in 20 μ L reactions containing 10 μ L SsoAdvanced SYBR Green Supermix (Bio-Rad), 300 nM primers, and 2 μ L template. Standard curves were generated using cloned plasmids or gBlocks (IDT). All reactions were run in triplicate. Primer sets and cycling conditions are listed in **Table.6**.

Table 6. qPCR/RT-qPCR primers

Target gene	Primer	Nucleotide sequence (5'-3')	Reference
Universal 16S rRNA	1055F	ATGGCTGTCAGCT	(Orschler et al., 2019)
	1392R	ACGGGCGGTGTGTAC	
pmoA	A189f	GGNGACTGGGACTTCTGG	(McDonald et al., 2008)
	Mb661r	CCGGMGCAACGTCYTTACC	
mmoX	mmoX 206f	ATCGCBAARGAATAYGCSCG	(Antony et al., 2010)
	mmoX 886r	ACCCANGGCTCGACYTTGAA	

where, N=G,A,T,orC;Y=C or T;M=A or C;K=G or T;S=G or C;R=G or A;B=C,T or G;W=A or T;V=A, C or G

Metagenomic analyses

Paired-end metagenomic reads were first subjected to quality control using Novogene's in-house pipeline. Low-quality bases and adaptor sequences were trimmed, and high-quality reads were merged with FLASH (Fast Length Adjustment of Short reads) under default parameters (minimum overlap length: 10 bp, maximum overlap length: 65 bp, mismatch ratio: 0.25).

MergeFASTQ files were converted to FASTA format using Seqtk (<https://github.com/lh3/seqtk>). The FunGene (Functional Gene Pipeline and Repository) database previously served as a curated reference for functional gene classification but is no longer maintained. To enable accurate classification of methanotrophic sequences, we constructed a custom *pmoA* reference library by compiling published *pmoA* sequences from all known methanotroph genera and species. Sequences were grouped into four phylogenetic types (Type I, Type II, and Type III) based on established frameworks. Environmental metagenomic reads were aligned to this custom reference using BLASTN (e-value $< 1 \times 10^{-6}$, minimum sequence identity $> 80\%$). For each query sequence, only the top-scoring hit was retained, and type-level assignments were aggregated to estimate the relative abundances of methanotroph groups.

Transcriptomic analyses.

For transcriptomic profiling, *M. trichosporium* OB3b cultures were exposed to each BNI compound MHPP, 1,9-D, and LA—at their respective estimated IC_{50} concentrations determined in this study, alongside untreated controls. Cultures were harvested in mid-exponential phase. Total RNA was extracted, and RNA quality was verified by agarose gel electrophoresis and an Agilent 2100 Bioanalyzer (RIN > 7). Ribosomal RNA was removed using the Ribo-Zero rRNA Removal Kit (Illumina). Strand-specific cDNA libraries were prepared following Illumina's protocols and sequenced by Novogene (Beijing, China) on an Illumina NovaSeq 6000 platform, yielding 150 bp paired-end reads. Raw reads were processed by Novogene's pipeline, including adaptor trimming, removal of low-quality bases, and exclusion of reads with $> 5\%$ ambiguous nucleotides. Clean reads were aligned to the *M. trichosporium* OB3b reference genome using HISAT2 with default parameters. Read counts were generated using featureCounts, and differential expression analysis was performed with DESeq2 (v1.34.0) in R. Counts were

normalized with the median-of-ratios method, and dispersion was estimated with the default shrinkage approach. Genes with adjusted p -values (Benjamini-Hochberg correction) < 0.05 and $|\log_2 \text{fold-change}| > 1$ were considered significantly differentially expressed. Functional enrichment analysis was performed using Gene Ontology (GO) annotations. Overrepresented GO categories among up- and downregulated genes were identified using Fisher's exact test with Benjamini-Hochberg correction ($\text{padj} < 0.05$). For visualization, enrichment scores were calculated as $(\text{Up} - \text{Down}) / (\text{Up} + \text{Down})$ per GO category, and the top enriched categories for each compound were plotted as heatmaps.

Statistical Analysis

All statistical analyses were performed using OriginLab (OriginLab Software, Northhampton, MA, USA). Nonlinear regression was used to estimate kinetic parameters ($V_{\max}$, K_m) and inhibitory concentrations (IC_{50}) from CH_4 oxidation and NO_2^- accumulation assays. For comparative analyses among treatments, one-way ANOVA followed by Tukey's post hoc test was applied unless otherwise noted. Data are presented as mean $\pm$ s.d. from at least three biological replicates. Statistical significance was accepted at $P < 0.05$ or $P < 0.01$, as indicated in the figure legends. Correlations between predicted binding affinities and inhibition intensities were tested using Pearson's correlation coefficients.

Analytical procedures

CH_4 concentrations in the headspace were measured by an Agilent 7890 GC equipped with a DB-624 capillary column ($60 \text{ m} \times 0.32 \text{ mm} \times 1.8 \mu\text{m}$) and an FID. For each measurement, 100 μL headspace gas was withdrawn and manually injected. Calibration curves ($R^2 > 0.99$) were generated before each experiment.

4.2.3 Results

4.2.3.1 BNI-producing plants suppress CH₄ oxidation in soil bioreactors

To determine whether BNI-producing plants influence CH₄ oxidation in complex soil-plant systems, we established soil bioreactors under three conditions: unamended control, NH₄⁺ amendment (25 mg NH₄⁺-N·kg_{-soil}⁻¹), and NH₄⁺ amendment with *Brachiaria*. NH₄⁺ addition significantly enhanced CH₄ oxidation ($P < 0.001$), with the maximum CH₄ consumption rate ($V_{\max}$) increasing from 18.0 (± 4.4) to 105.7 (± 8.8) nmol·min⁻¹·g_{-soil}⁻¹. In contrast, *Brachiaria* incubation strongly suppressed CH₄ oxidation (13.5 nmol·hr⁻¹·g_{-soil}⁻¹), reducing activity to levels below those observed in the unamended control, despite NH₄⁺ amendment ($P < 0.001$) (**Fig.14a**), and thus were not considered further.

Metagenomic shotgun sequencing was performed to assess methanotrophic community shifts under the three treatments. Across the dataset, 21,283,174 sequences were obtained from the control, 14,678,467 from the NH₄⁺-amended soil, and 12,969,658 from the NH₄⁺ plus *Brachiaria* soil. Screening for methanotroph-specific *pmoA* genes recovered 21,976 sequences, which were classified into Type I (9,882 sequences), Type II (4,084), Type III (20), and Type IV (7,990) groups. The relative abundance of Type I MOB was significantly reduced in the *Brachiaria* treatment compared to both the control and NH₄⁺-only treatments ($P < 0.01$). Type II MOB exhibited a modest decline, while Type III and IV groups were largely unaffected. (**Fig.14b**) These results demonstrate that BNIs from *Brachiaria* suppress CH₄ oxidation in soils, with metagenomic evidence pointing to a disproportionate effect on Type I MOB.

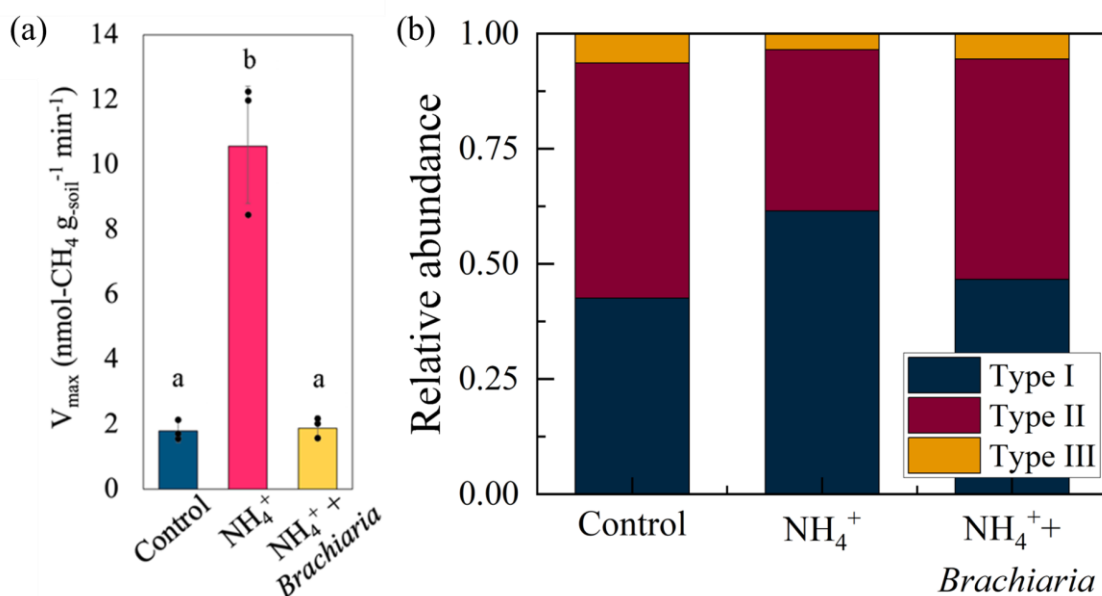


Figure 14. (a) Maximum methane oxidation rates ($V_{\max}$) measured in soils recovered from the bioreactors. Bars indicate mean $\pm$ s.d. and dots represent individual biological replicates ($n = 3$). NH_4^+ amendment stimulated methane oxidation, whereas *Brachiaria* largely abolished this effect. Different letters indicate statistically significant differences ($P < 0.05$, one-way ANOVA with post hoc test). (b) Community profiles for the bioreactor with *B. humidicola* species. Sequences were used to screen related FunGene *pmoA* sequences with an e-value threshold of $1e^{-6}$ and minimum identity of 80%. Based on this screening, separate and non-overlapping databases for each *pmoA* type were constructed.

4.2.3.2 Axenic culture experiments: Methanotrophs inhibition assays

To evaluate the inhibitory effects of BNI compounds on methanotrophic activity, *M.*

trichosporium strain OB3b was incubated with varying concentrations of 1,9-D (0.29–5.74 mM),

MHPP (0.28–2.78 mM), and LA (5.7–57.1 μM), and Michaelis–Menten parameters were

determined. For all three compounds, both the maximum CH_4 oxidation rate ($V_{\max}$) and the

substrate affinity constant (K_m) decreased with increasing inhibitor concentration, consistent with

an uncompetitive mode of inhibition (**Fig. 15a and b**). In contrast, LA exhibited measurable

inhibition at micromolar concentrations (**Fig. 15c**), indicating that LA interacts with OB3b at

substantially lower concentrations than 1,9-decanediol or MHPP. At its aqueous solubility limit

(5.7 μM), LA reduced CH_4 oxidation by $\sim 20\%$. As prior studies in *N. europaea* reported $\sim 80\%$ inhibition at 55 μM —above the aqueous solubility threshold tested here—(Subbarao et al., 2008) we additionally tested LA at 57.1 μM (nominal concentration exceeding aqueous solubility) for comparison. Under these conditions, inhibition increased to $\sim 42\%$, but did not reach 50%, precluding reliable IC_{50} estimation. These data indicate that LA displays higher apparent potency but limited maximal inhibition under the experimental conditions tested. These results suggest that LA can inhibit CH_4 oxidation more strongly at artificially elevated concentrations, but under physiologically relevant conditions its overall inhibitory capacity remains limited compared with 1,9-D and MHPP.

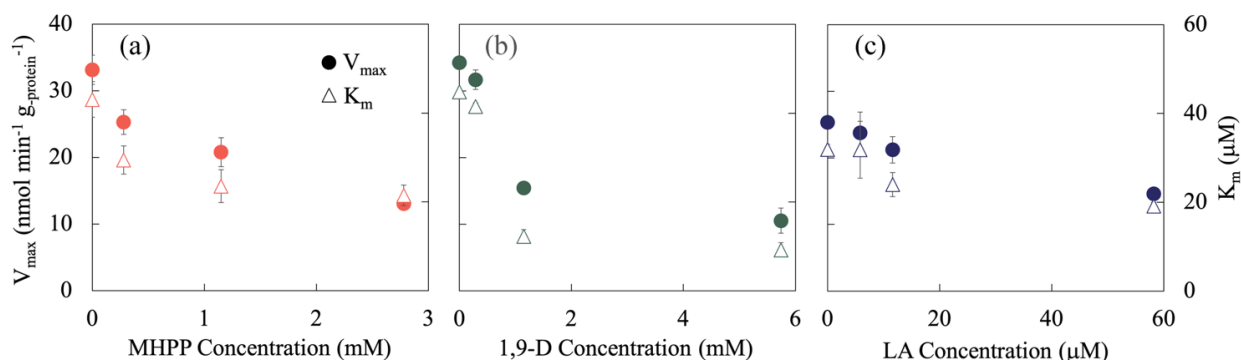


Figure 15. Effects of individual BNI compounds on methane oxidation kinetics. Increasing concentrations of each compound resulted in concurrent decreases in $V_{\max}$ and K_m , consistent with an uncompetitive-like inhibition pattern.

4.2.3.3 Type I methanotroph BG8 exhibits enhanced sensitivity to BNIs

Because metagenomic analysis of the soil bioreactors indicated that Type I methanotrophs were disproportionately reduced under *Brachiaria* incubation, inhibition assays were performed with *Methylobacterium album* BG8, a representative Type I methanotroph, to determine whether this lineage-level pattern could be reproduced in pure culture. Across all compounds tested, BG8 was more sensitive to BNIs than *M. trichosporium* OB3b. The IC_{50} for MHPP was lower in BG8 (432 μM) than in OB3b (730 μM). Similarly, the IC_{50} for 1,9-D was lower in BG8 (391 μM) than in

OB3b (1076 μM). For LA, the IC_{40} was also lower in BG8 (14 μM) than in OB3b (53 μM).

Although these results are based on representative strains, this pattern is consistent with the preferential reduction of Type I taxa observed in soil metagenomes under *Brachiaria* incubation.

(Fig. 16).

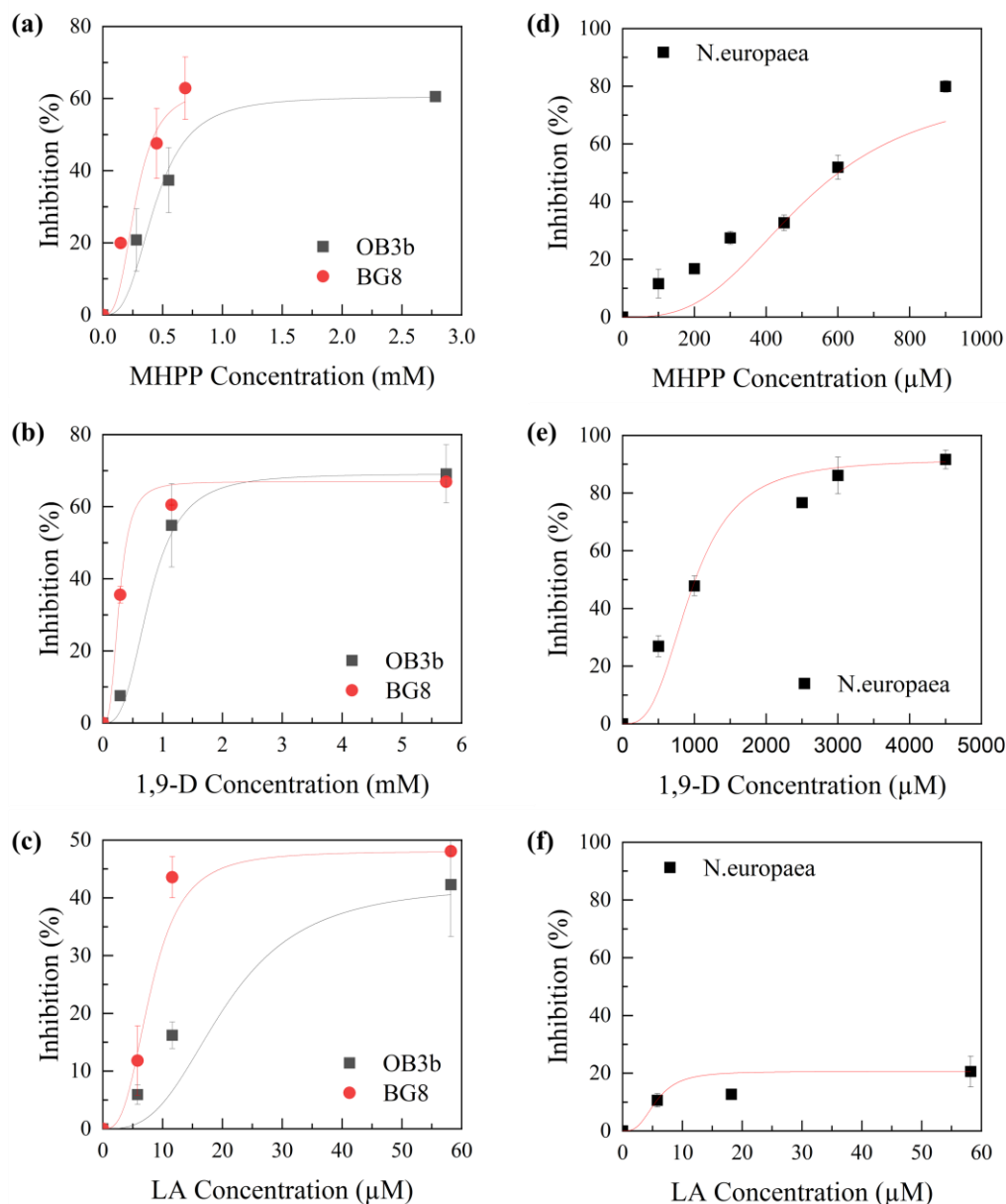


Figure 16. Kinetic analyses comparing the V_{max} inhibition in three different BNI compounds in *M. trichosporium* OB3b, *M. album* BG8, and *N. europaea*. (a) *M. trichosporium* OB3b and *M. album* BG8 of V_{max} inhibition for various concentrations of MHPP. (b) *M. trichosporium* OB3b and *M. album* BG8 of V_{max} inhibition for various concentrations of 1,9-D. (c) *M. trichosporium* OB3b and *M. album* BG8 of V_{max} inhibition for various concentrations of LA. (d) *N. europaea*

$V_{\max}$ inhibition for various concentrations of MHPP. (e) *N.europaea* $V_{\max}$ inhibition for various concentrations of 1,9-D. (f) *N.europaea* $V_{\max}$ inhibition for various concentrations of LA.

4.2.3.4 BNI is largely pMMO-specific with compound-dependent nonspecific effects

To determine enzyme specificity, strain OB3b was assayed under copper-limited conditions that repress pMMO and induce sMMO. While IC_{50} concentrations were estimated under copper replete conditions, inhibition was markedly alleviated under copper limitation, resulting in only 12-20% residual inhibition for MHPP and LA. In contrast, 1,9-D exhibited the highest residual values, reaching up to 38%, suggesting a partial inhibitory effect independent of pMMO (Fig.17).

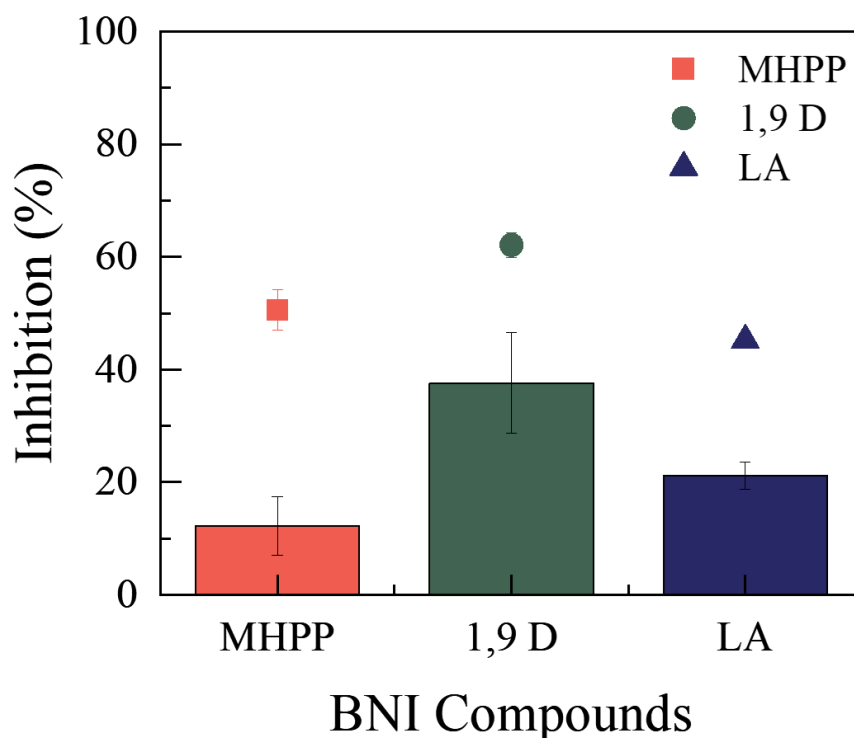


Figure 17. *M. trichosporium* OB3b inhibition analysis under copper-limited conditions with IC_{50} concentrations for BNIs compounds. Dotted line is inhibition ratio of same concentration of each BNI compounds for *M.trichosporium* OB3b with copper.

4.2.3.5 Reversibility of BNI-mediated inhibition

To determine whether BNI-mediated inhibition was reversible or indicative of irreversible enzyme inactivation, OB3b cultures were incubated with each compound at their respective IC_{50} concentrations and to LA at its IC_{40} concentration, followed by removal of the compounds through washing and resuspension in fresh inhibitor-free medium prior to kinetic analysis. Following washout, methane oxidation activity recovered to 85–95% of untreated controls across treatments. (**Fig. 18**). These results indicate that BNI-mediated inhibition is reversible under the conditions tested and does not reflect suicidal inactivation of methane monooxygenase.

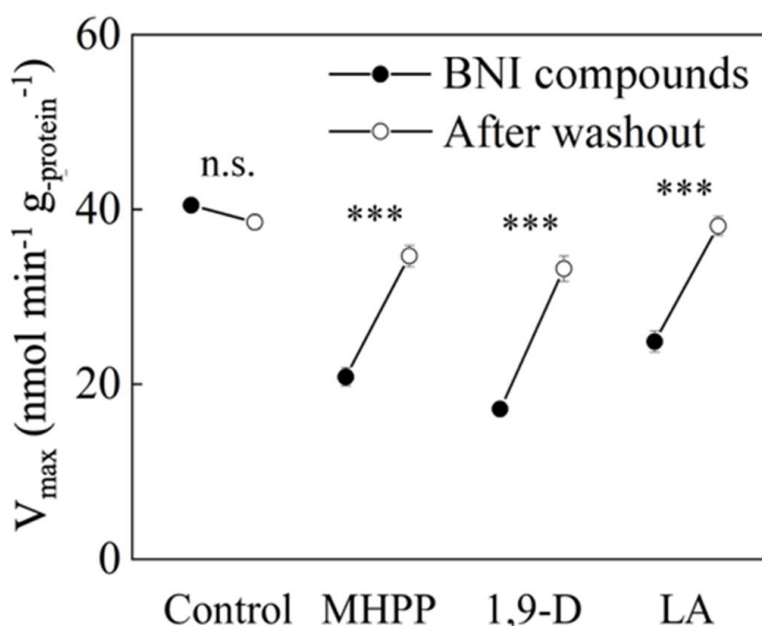


Figure 18. Reversibility of inhibition by BNI compounds following inhibitor removal. Methane oxidation activity recovered substantially after washout of each BNI compound, indicating that inhibition was reversible. Control samples showed no significant change following washout, whereas BNI-treated samples exhibited significant recovery of activity (paired two-tailed t-test, $n = 3$; *** $P < 0.001$; ns, not significant). Data are shown as V_{max} values. Error bars indicate s.d. and are shown where visible; in some cases they are smaller than the symbols.

4.2.3.6 Molecular docking and dynamics simulations

To investigate the structural basis of the pMMO-associated inhibition observed in the methanotroph assays, molecular docking was performed using a homology model of *M.*

trichosporium OB3b pMMO (**Fig. 19a**) with three representative BNI compounds: MHPP, 1,9-

D, and LA. Across docking simulations, the highest-affinity poses were consistently located within the PmoB-PmoC interfacial pocket (**Fig. 19b and 19c**). Predicted binding energies for these compounds ranged from -4.6 to -6.1 kcal mol $^{-1}$ (**Fig. 19d**).

To test whether this predicted binding site reflects a broader structural feature of hydrophobic inhibitors, two structural analogues of 1,9-D—dodecanoic acid and 1-decanol—were subsequently examined. Docking simulations predicted that both analogues bind within the same PmoB–PmoC interfacial pocket. Consistent with this prediction, both compounds inhibited CH $_4$ oxidation in *M. trichosporium* OB3b (IC $_{50}$ values of 909 μ M and 1290 μ M for dodecanoic acid and 1-decanol, respectively). These results support the predicted interaction of hydrophobic inhibitors with the PmoB–PmoC interfacial pocket. Notably, all five chemically distinct compounds converged on the same PmoB–PmoC interfacial pocket in docking simulations.

When the two structural analogues were included, predicted binding energy across all five compounds was strongly associated with experimentally determined IC $_{50}$ values (**Fig. 19e**), indicating that both the native BNIs and their structural analogues follow the same structure–activity relationship. Binding energy exhibited a perfect monotonic relationship with IC $_{50}$ (Spearman’s $\rho = 1.00$, two-tailed $p = 0.017$) and a strong linear correlation between binding energy and IC $_{50}$ (Pearson’s $r = 0.97$, $p = 0.006$). Although docking energies provide relative estimates rather than direct measures of binding affinity, the observed correlation supports the functional relevance of the PmoB–PmoC interfacial pocket as a plausible interaction site for BNI compounds.

The identified interfacial pocket lies adjacent to a channel-like cavity extending toward the Cu $_C$ /Cu $_D$ region of the PmoC subunit, proposed to harbor the catalytic copper center of pMMO, (Koo et al., 2022; Tucci et al., 2023) and is also located near the Cu $_B$ site (**Fig. 19c**).

Ligand binding at this site may influence molecular flux through this channel, for example by affecting substrate access or product egress, without directly occupying the catalytic center. This spatial arrangement is consistent with the inhibition pattern observed in the kinetic assays.

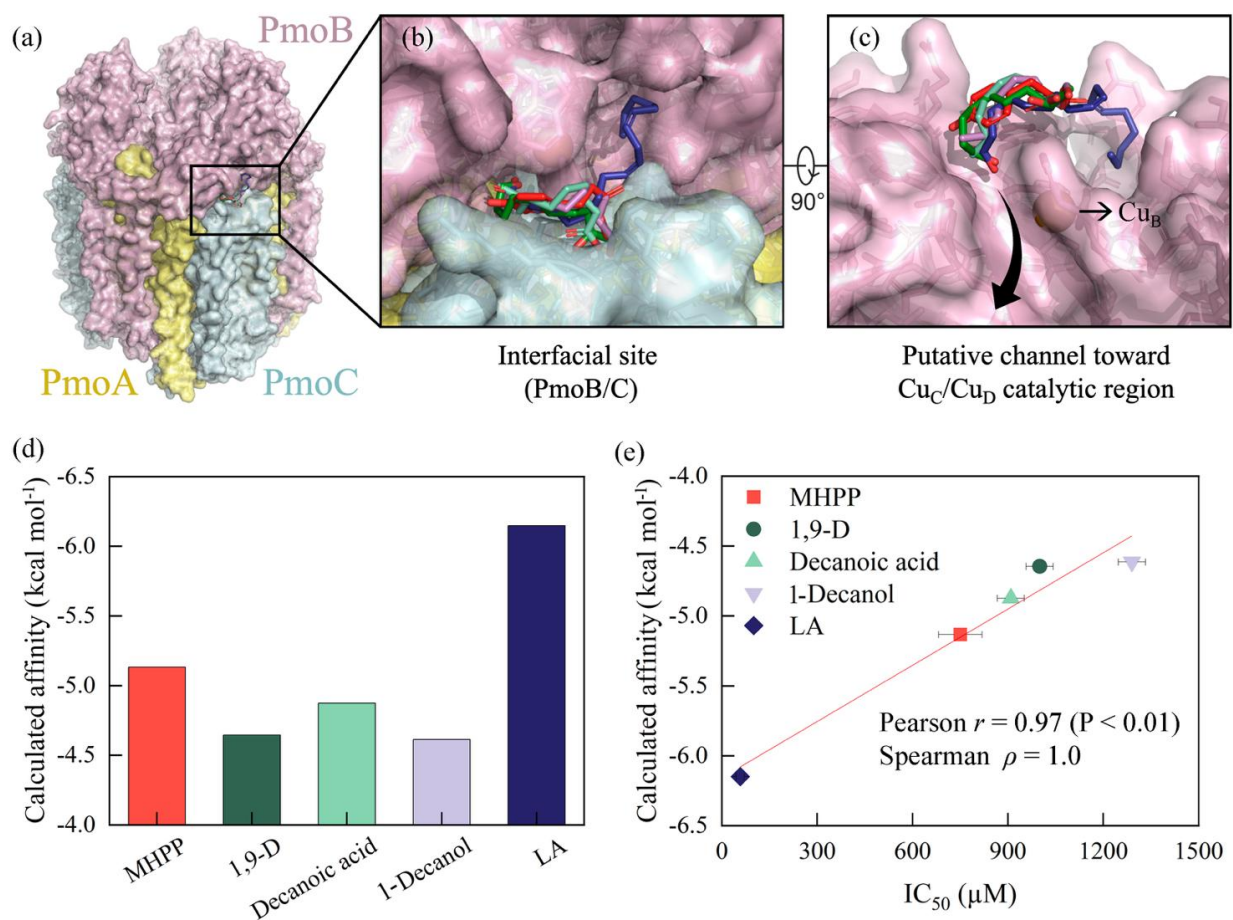


Figure 19. Molecular docking identifies a putative BNI-binding site in pMMO and links binding affinity to inhibitory potency. (a) Predicted homology model of pMMO showing the location of a putative binding site at the PmoB–PmoC interface.

(b) Close-up view of docked poses for MHPP (red), 1,9-D (dark green), decanoic acid (light green), 1-decanol (light purple), and LA (navy blue) within the interfacial pocket.

(c) Rotated view of the same pocket, showing its position near a putative channel toward the Cu_C/Cu_D catalytic region and in proximity to the Cu_B site.

(d) Calculated binding affinities of the docked compounds.

(e) Relationship between calculated binding affinity (kcal mol⁻¹) and inhibitory potency (IC₅₀, except LA where IC₄₀ is shown). Stronger binding was associated with lower IC₅₀ values.

Colors are consistent across panels.

4.2.3.7 Mechanistic insights from RNA-Seq analysis

To further examine how methanotroph cells respond to BNI exposure at the transcriptional level, RNA-seq analysis was performed. Principal component analysis revealed clear separation of MHPP-treated samples from controls, whereas 1,9-D induced broader dispersion consistent with widespread transcriptional perturbation. LA-treated samples showed a more moderate shift in global transcriptomic space (**Fig. 20a**). Differential expression analysis further highlighted these differences (**Fig. 20b-d**). 1,9-D triggered the broadest response, affecting a large number of genes with moderate fold changes across diverse functional categories, including stress responses, transporters, and membrane-associated processes. MHPP induced a more structured transcriptional shift, with coordinated changes in specific metabolic pathways. In contrast, LA elicited fewer differentially expressed genes but produced high-amplitude and high-significance expression changes in a subset of loci. These patterns were further resolved at the gene level (**Fig. 20e**), where LA induced pronounced upregulation of *nuoL* and *dksA* and strong downregulation of *pmoA* and *kdp* operon genes, consistent with a selective but high-impact transcriptional response. Cultures were exposed to MHPP and 1,9-D at their estimated IC₅₀ concentrations and to LA at its IC₄₀ concentration. Under these conditions, methane oxidation was inhibited by 48.5%, 58.6%, and 38.6% for MHPP, 1,9-D, and LA, respectively. Taken together, these results indicate that BNIs induce fundamentally distinct transcriptional programs in methanotrophs: 1,9-D acting as a broad stressor, MHPP producing a structured metabolic shift, and LA eliciting a selective but high-amplitude response.

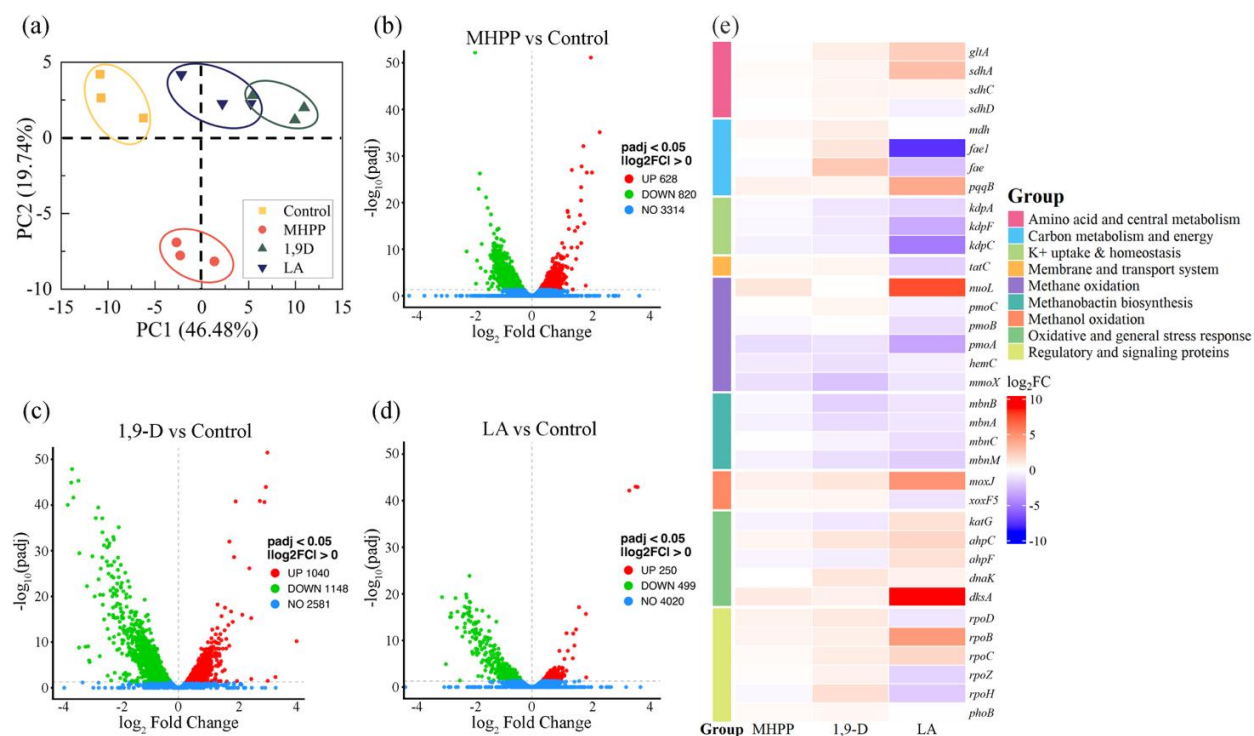


Figure 20. Compound-specific transcriptomic responses of *Methylosinus trichosporium* OB3b to BNI exposure.

(a) Principal component analysis (PCA) showing global transcriptomic differences among treatments.

(b–d) Volcano plots showing differential gene expression relative to control (adjusted $P < 0.05$).

(e) Heatmap of selected genes involved in methane oxidation, central metabolism, transport, and stress response. Notable responses include upregulation of *nuoL* and *dksA* and downregulation of *pmoA* and *kdp* operon genes under LA treatment. Cultures were exposed to MHPP and 1,9-D at their estimated IC₅₀ concentrations and to LA at its IC₄₀ concentration, corresponding to 47.5%, 58.6%, and 38.6% inhibition of methane oxidation, respectively.

4.2.3.6 Comparative sensitivity of ammonia-oxidizing and CH₄-oxidizing bacteria

To directly compare the inhibitory effects of BNIs on ammonia-oxidizing bacteria and methanotrophs, parallel inhibition assays were performed with *N. europaea*, a model ammonia-oxidizing bacteria, using the same three BNI compounds evaluated with methanotrophs: MHPP, 1,9-D, and LA. Because inhibitory responses depend on the ratio of inhibitor concentration to microbial biomass (Udekwu et al., 2009; X. Wang et al., 2023), inhibitory potency was normalized to cell abundance (16S rRNA copy number corrected) prior to comparison. In these

assays, *N. europaea* cultures were tested at $OD_{600} \approx 0.1$, whereas methanotrophs assays were performed at higher biomass ($OD_{600} \approx 0.3$). For MHPP and 1,9-D, inhibitory potency was expressed as the negative $\log_{10}$ of the cell-normalized IC_{50} values ($-\log_{10} IC_{50}$) to facilitate comparison across orders of magnitude. Under this normalization, both methanotrophic strains were markedly more sensitive than *N. europaea*, indicating that BNIs inhibit CH_4 oxidation at lower per-cell exposure levels than those required to suppress nitrification. For LA, inhibition did not reach 50% in either ammonia-oxidizing bacteria or methanotrophs assays, precluding IC_{50} estimation. In *N. europaea*, inhibition reached its maximum ($\sim 13\%$) at $57.1 \mu M$, corresponding to a biomass-normalized exposure of $3.6 \times 10^{-8} \mu M \text{ cell}^{-1}$. Achieving the same normalized exposure in methanotrophs assays as in the *N. europaea* experiment would require substantially higher LA concentrations, estimated at approximately $734 \mu M$ for OB3b and $985 \mu M$ for BG8. However, inhibition in both methanotrophic strains had already reached a plateau within the tested concentration range (approximately 42% for OB3b and 48% for BG8), with maximal suppression observed at $\sim 57.1 \mu M$. Thus, even at normalized exposures far below those required to match the ammonia-oxidizing bacteria assay, methanotrophs exhibited markedly stronger inhibition than *N. europaea*. Together, these results suggest that BNIs may suppress methanotrophic activity to a similar or greater extent than nitrification.

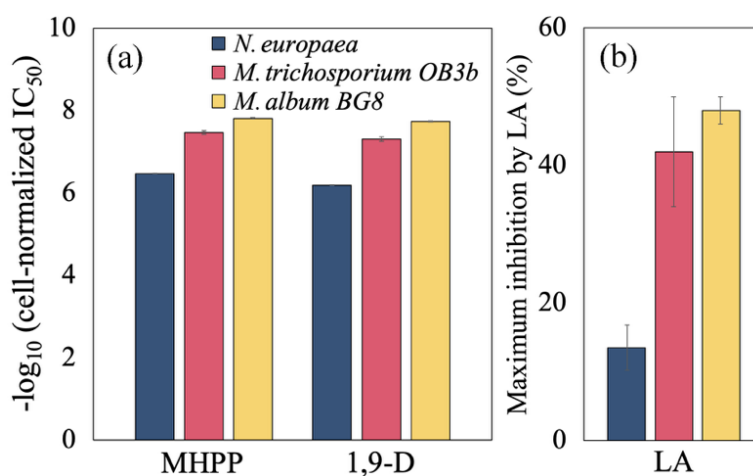


Figure 21. Comparative sensitivity of ammonia- and methane-oxidizing bacteria to BNI compounds.

(a) Cell-normalized inhibitory potency of MHPP and 1,9-D toward *Nitrosomonas europaea*, *Methylosinus trichosporium* OB3b, and *Methylobacterium album* BG8. Inhibitory potency is expressed as $-\log_{10}(\text{IC}_{50})$ of the cell-normalized values (M cell^{-1}). Error bars represent propagated standard deviations following log transformation of IC_{50} values.

(b) Maximum inhibition by LA observed within the tested concentration range. Error bars represent standard deviations of triplicate measurements.

4.2.4 Discussion

This study reveals that BNIs, naturally released by plants, not only suppress ammonia oxidation but also curtail CH_4 oxidation by MOB. By integrating evidence from soil bioreactors, metagenomic profiling, pure-culture kinetics, and molecular docking, we show that BNIs can substantially reduce the microbial capacity for CH_4 oxidation in soil systems. These findings uncover a previously unrecognized ecological tradeoff (Florio et al., 2021; Peters et al., 2013; Subbarao, Rao, et al., 2013; Zhang et al., 2023),: while BNIs limit nitrification and thereby lower N_2O emissions, they may simultaneously diminish CH_4 consumption. Because nitrification and methanotrophy are central microbial processes linking the global nitrogen and carbon cycles, the dual action of BNIs has consequences that extend beyond local soil environments. Our comparative assays extend this tradeoff beyond nitrification. Although BNIs were originally characterized as inhibitors of AMO in AOB, we found that MOB were equally or more sensitive

on a per-cell basis (**Fig. 21**). In particular, LA, which was constrained by solubility and produced only weak inhibition in *N. europaea* (~15%), substantially suppressed CH₄ oxidation in MOB (40–50%). These findings highlight that BNIs can simultaneously constrain both nitrogen and carbon transformations, amplifying their potential to reshape coupled C–N cycling and GHG fluxes.

Building on these observations, we next examined the mechanistic basis of inhibition. Kinetic assays with *M. trichosporium* strain OB3b revealed an uncompetitive mode of inhibition with simultaneous decreases in $V_{\max}$ and K_m (Fig. 15). Molecular docking supported this pattern, identifying a hydrophobic region near the entrance of the pMMO active site where multiple BNIs localized, consistent with peripheral binding rather than direct competition with CH₄. This aligns with recent structural work revealing a hydrophobic cavity adjacent to the *PmoC* copper site in active pMMO (Koo et al., 2022) and with prior observations that hydrophobic cavities between *PmoC* and *PmoB* may guide substrate to the pMMO active site (Ross & Rosenzweig, 2017). Analog experiments confirmed that hydrophobic chain interactions were the primary determinant of binding, with terminal functional groups playing a minor role. Importantly, predicted binding affinities correlated strongly with experimentally determined IC₅₀ values across all compounds tested (**Fig. 19**), providing evidence for a structure–activity relationship in which hydrophobicity and binding strength govern inhibitory potency. However, compound-specific differences were also apparent: MHPP acted in a primarily pMMO-specific manner, 1,9-D exhibited additional nonspecific inhibition (residual inhibition under sMMO conditions, reduced *E. coli* growth, and weaker predicted binding affinity), and LA showed limited maximal inhibition. It should be noted that 1,9-D was previously reported to affect AMO only (Sun et al., 2016b), underscoring that its broader inhibitory effects in MOB diverge from earlier findings. In addition, peripheral

binding near the pMMO active-site entrance is consistent with scenarios such as perturbation of access/egress channels or allosteric effects on catalytic cycling, both of which could explain the observed uncompetitive kinetics, as has been reported in other systems where product release is slowed. (Kokkonen et al., 2021).

Our choice of MHPP, 1,9-D, and LA was motivated by prior studies with *N. europaea*, where MHPP and 1,9-D were identified as AMO (Sun et al., 2016b; Zakir et al., 2008b), whereas LA inhibited both AMO and HAO (Subbarao et al., 2008). In those studies, LA was tested at concentrations far exceeding its aqueous solubility (1.6 mg L^{-1} , $\sim 5.7 \text{ }\mu\text{M}$), with in vitro assays conducted up to 16 mg L^{-1} ($\sim 57.1 \text{ }\mu\text{M}$) and reported IC_{80} values at this level (Subbarao et al., 2008). By contrast, in our experiments with MOB (OB3b, BG8), inhibition by LA plateaued at its solubility limit, with no further suppression at higher tested concentrations. While LA reduced CH_4 oxidation only up to $\sim 40\%$ (**Fig. 16**), the calculated IC_{40} was the lowest among the three BNIs tested, indicating strong inhibitory potency at low concentrations despite limited maximal inhibition. This contrasts with MHPP and 1,9-D, which exhibited more classical dose–response curves consistent with pMMO-specific inhibition. (**Fig. 16**) Transcriptomic profiles further reinforced these distinctions, with 1,9-D eliciting broad stress-associated responses, MHPP producing a more targeted shift consistent with pMMO inhibition, and LA causing comparatively modest transcriptional changes, in line with its limited maximal inhibition. Taken together with prior findings in ammonia oxidizers, our results demonstrate that BNIs represent a chemically diverse class of inhibitors, each with distinct inhibition profiles and ecological implications.

Moving beyond individual compounds, we next evaluated the ecological consequences of BNI release. These were evident in our soil bioreactor experiments, where CH_4 oxidation was

suppressed in the presence of *Brachiaria* and metagenomic analysis revealed a disproportionate decline in Type I MOB. This lineage-level effect was further supported by pure culture assays: *M. album* BG8 (Type I) exhibited greater sensitivity to MHPP, 1,9-D and LA than the Type II strain OB3b (**Fig. 16**), Type I organisms being more vulnerable in some cases. Because Type I and Type II MOB occupy distinct ecological niches—Type I often dominating in nitrogen-rich soils and Type II in nitrogen-limited environments (Hanson & Hanson, 1996; Knief, 2015b)—such selective inhibition could shift methanotroph community structure in BNI-affected systems. One possible factor contributing to this pattern is the auxiliary CH₄ oxidation capacity of Type II MOB, many of which can express the sMMO under copper limitation (Liebner & Svenning, 2013; Semrau et al., 2010). Although convincing evidence for sMMO expression in terrestrial environments is still lacking, it has been suggested that its apparent rarity may reflect primer bias and methodological constraints in detecting *mmoX* (Semrau et al., 2010). Thus, while the ecological role of sMMO remains uncertain, the potential for its conditional expression raises the possibility that Type II MOB may, under some circumstances, be less sensitive to BNIs than Type I.

Having established the effects of BNIs at the cellular and community levels, it is important to consider their broader implications for the global CH₄ cycle, particularly given the widespread occurrence of BNI across the plant kingdom, ranging from agricultural crops (Coskun et al., 2017; Nardi et al., 2020) to forest trees (Andrianarisoa et al., 2010; Laffite et al., 2020). Wetlands, the largest natural source of atmospheric CH₄, and rice paddies, the largest anthropogenic source, both rely on methanotrophy in surface soils and rhizospheres as a key sink that attenuates gross CH₄ production (Conrad, 2009; Saunio et al., 2020). Our findings raise the possibility that BNIs released from plants may interfere with this sink function, thereby

modulating net CH₄ fluxes in these ecosystems. In wetlands, BNI exudation by native vegetation could influence not only nitrification and other processes in the nitrogen cycle but also methanotrophy, thereby modulating the balance of GHG fluxes. Accordingly, global and regional CH₄ models should represent BNI-driven constraints on methanotrophy in wetlands and flooded ecosystems to more accurately capture net fluxes. In rice paddies, the relevance is more direct, as rice itself produces BNI compounds such as 1,9-D (Sun et al., 2016b). Thus, BNI-based breeding programs aimed at enhancing NUE and reducing N₂O emissions should also evaluate their impacts on CH₄ uptake, ensuring that progress in nitrogen management does not come at the expense of CH₄ sink strength. Together, these considerations highlight the need to integrate BNI effects into our understanding of CH₄–climate feedbacks.

Several limitations of this study also point toward promising avenues for future research. The bioreactor and pure-culture experiments, although mechanistically informative, may not fully capture the complexity of plant–soil–microbe interactions in natural environments, where fluctuating oxygen availability, soil redox dynamics, microbial competition, and spatial heterogeneity in root exudation could all influence the magnitude of BNI effects. While our molecular docking analyses support the kinetic results, they rely on predicted structures and simplified binding assumptions, and experimental structural confirmation of BNI-enzyme interactions will be an important next step. Moving forward, extending these findings to field-scale systems will be essential, combining CH₄ flux measurements with high-resolution microbial and transcriptomic analyses in BNI-rich environments such as rice paddies and wetlands. Incorporating BNI effects into process-based and Earth-system models will also be critical for evaluating their contribution to the global CH₄ budget. As BNI-based strategies continue to be developed for improving nitrogen-use efficiency and reducing N₂O emissions,

interdisciplinary studies linking agronomy, microbiology, and climate science offer a promising path to ensure that their broader impacts on GHG dynamics are fully understood.

Chapter 5 - Conclusion

Agricultural activities are the predominant source, responsible for over two-thirds of global anthropogenic N₂O emissions and one-third of total CH₄ emissions. And these systems are mostly controlled by microbial processes. To make a comprehensive understanding and control of both GHGs, we performed two different studies: 1) Building on the recognition of unquantified N₂O emissions from forages and control, 2) Impact of BNI on methanotrophs. Based on the mini-silo experiment and the Crop Production 2022 Summary of the United States Department of Agriculture National Agricultural Statistics Service, the total N₂O emission potential amounts to 5.3 MMT CO₂ eq. Mechanistic tests showed no evidence for a nitrification contribution (no effect of O₂ pulses or low-dose acetylene; no *amoA* transcripts) and instead identified denitrification as the dominant pathway: chlorate suppressed *narG* expression and cut N₂O by up to 99%, with effective mitigation even at 0.01% (w/w) and further improvement when paired with a modest external carbon source (acetate) to raise C/N. These results highlight a practical, cost-effective mitigation lever—judicious chlorate dosing (order-of-cents per ton DW) with carbon balancing—that can deliver large abatement relative to silage inoculant costs and yield substantial social-cost savings at scale. While extrapolation uncertainties remain (climate, storage type, management heterogeneity), the core findings are actionable: (i) incorporate forage-related N₂O into national inventories to close a major accounting gap; (ii) Lab-scale interventions that combine chlorate-based denitrification control with C/N management; and (iii) refine process attribution with expanded multi-omics and flux measurements across real silage systems. Measuring what matters enables management—quantifying this neglected source and demonstrating feasible controls provides an immediate pathway to meaningful N₂O reductions from a ubiquitous agricultural practice.

Second, this study also shows that plant-derived BNIs can unintentionally suppress CH₄ oxidation, revealing a key tradeoff in coupled N-C cycling: while BNIs curb nitrification and N₂O, they also diminish soil CH₄ sink strength. Across soil columns, *Brachiaria* reduced CH₄ consumption and selectively depleted more on Type I methanotrophs, a lineage-level pattern recapitulated in pure culture where *M. album* BG8 (Type I) was more sensitive than *M. trichosporium* OB3b (Type II). Kinetic assays and docking converge on an uncompetitive inhibition, pMMO-proximal mechanism in which diverse BNIs bind a hydrophobic pocket near the active-site entrance; MHPP acts largely pMMO-specifically, 1,9-D exhibits additional nonspecific cytotoxicity, and LA shows strong low-dose potency but limited maximal inhibition due to solubility. Comparative tests indicate methanotrophs are at least as sensitive per cell as ammonia oxidizers, implying BNIs can constrain both nitrification and methanotrophy. Together, these findings highlight that BNI-based agronomic strategies promising for nitrogen-use efficiency—must also account for potential weakening of the CH₄ sink, particularly in wetlands and rice systems, and motivate field-scale validation, structural confirmation of BNI–pMMO interactions, and integration of BNI effects into process and Earth-system models to more accurately predict GHG fluxes.

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