# BEAM-FOIL LIFETIMES OF CrII

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by

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#### Introduction

In 1962, the first formal proposal for a beam-foil experiment was made by Kay<sup>1</sup>. Beam-foil Spectroscopy (BFS) has since come into wide use. Along with other applications BFS provides a new method for measuring the lifetimes of atomic states from which radiative transition rates may be derived. Previously arc, spark, and absorption spectra were the primary sources for this information. BFS presents problems of its own, of course, but it does not have the problem associated with it of the assumption of thermal equilibrium in the light source. In addition, one need not worry about plasma stratification, which is a difficulty encountered in shock tube experiments.<sup>2</sup>

A BFS lifetime measurement is performed by passing a beam of accelerated ions through a thin foil. As atoms pass through the foil the valence electrons are excited. Downstream of the foil one may observe the number of photons emitted as a function of distance from the foil. The velocity of the beam after the foil is then used to convert the distance scale to a time scale. Thus the final result is a plot of the number of photons emitted versus time. The slope of this plot is the lifetime of the emitting level.

Lifetimes of atomic energy levels may be used with branching ratios to find atomic transition probabilities, which are important to the study of elemental abundances in the solar and stellar atmosphere. Recent calculations of the chromium abundance, based on BFS lifetimes of CrI, 3 make it desirable to calculate a similar abundance based on the lifetimes

of CrII. The reason for this is that at solar photospheric temperatures the principle ionic species is CrII. Therefore, a small error in the temperature-pressure stratification might lead to a large error in the fractional abundance of CrI and would produce a significant error in determining the total chromium abundance from CrI. The ratio of CrII to total chromium abundance should be much less sensitive to small errors in the temperature-pressure stratification.

The work presented in this thesis may be conveniently divided into four phases:

- 1) Some question as to the accuracy of BFS measurements has been raised in connection with the energy loss incured by the beam. 4 The problem has been examined to some extent in this work. The energy loss for 89 keV Ar+ has been measured and agrees surprisingly well with stopping power calculations.
- 2) In preparation for the lifetime experiments, spectral scans of the BFS spectrum have been taken between 1900 and 4100 A. Lines originating in CrI and CrII exist in this region and have been identified.
- 3) Lifetimes of five lines representing four levels in CrII have been measured. These measurements were made at two energies to ensure adequate knowledge of the energy loss.
- 4) A preliminary calculation of the solar abundance of chromium has been made based on our CrII results.

#### Spectroscopic Background

The following discussion will define common spectroscopic term-inology and relate the Einstein coefficient (A) to the oscillator strength (f), and to the atomic radiative lifetime ( $\tau$ ). The following definitions are based on the shell model of the atom and the LS coupling

scheme, and will consider only radiative transitions of the valence electrons. 5 In this coupling scheme, a spectroscopic "term" is characterized by the total orbital angular momentum (L) and the total spin (S) of the valence electrons. The conventional notation for a term is 2S+1L, where 2S+1 is called the "multiplicity" of the term. The multiplicity is the number of possible orientations of S relative to L. One may think of the multiplicity as the number of ways L and S can couple to form the total angular momentum, J, but this is only true for  $L \geq S$ . For example, if  $S = \frac{1}{2}$  and L = 0 as in the hydrogen atom, the term is S and there are two possible orientations of S, but only one possible coupling to form J (J =  $\frac{1}{2}$ ). A "level" designates one possible value for J, and is written  $^{2S+1}L_{\tau}$ . One specifies the "sublevel" that an atom is in by specifying the orientation of the atom, M. Sublevels are degenerate in energy except in the presence of a magnetic or electric field. A transition between two sublevels produces a "line component". The blend of line components produced by all transitions between two particular levels form a "line". The set of all lines produced by transitions between two terms is called a "multiplet".

The lifetimes of atomic levels are the principal quantities with which this work is concerned. However, the literature often refers to measurements of "gA-values or gf-values" as a measure of electromagnetic transition probabilities. Therefore, it is appropriate to discuss these two quantities as an alternative means of expressing transition probabilities. 6

Let us express the initial sublevel as i  $(\alpha'J'M')$ , where  $\alpha'$  represents all internal quantum numbers, and the final sublevel as  $f(\alpha'JM)$ . Also let us define  $A(\alpha'J'M'\alpha'JM)$  to be the transition rate for the transition

i->f, that is the transition probability per unit time for i->f. Since, in the absence of an electric or magnetic field the components blend to produce a single line, it is important to know the transition rate for the line. For each level there will be 2J+1 sublevels, corresponding to the 2J+1 possible values of M. To calculate the transition rate for the level (A ( $\alpha'J'-\alpha J$ )) it is necessary to sum the sublevel transition rate over all M and M'. Thus, A( $\alpha'J'-\alpha J$ ) is defined by the equation

$$N(J') A(\alpha'J'-\alpha J) = \sum_{M'M} N(J'M') A(\alpha'J'M'-\alpha JM),$$

where N(J') is the population of the level and N(J'M') is the population of each of the sublevels  $i(\alpha'J'M')$ . Since the transition rate of the sublevel is not dependent on the orientation of the atom (M'), it may be factored from the sum over M'. Thus, supressing M' in the transition rate,

$$N(J') A(\alpha'J'-\alpha J) = \sum_{M} A(\alpha'J'-\alpha JM) \sum_{M'} N(J'M').$$

Since it is clear that

$$\sum_{M'} N(J'M')/N(J') = 1,$$

we may write

$$A(\alpha'J') = \sum_{M} A(\alpha'J' - \alpha JM).$$

In a thermally excited source the sublevels are assumed to be equally populated. Thus, we may write

$$N(J') = (2J'+1) N(J'M') = gN(J'M'),$$

which is the defining equation for the statistical weight g. The observed intensity I is equal to the number of atoms multiplied by the transition rate. The intensities and populations are the quantities normally measured in thermal sources. One may express the intensity as

$$I = N(J') A(\alpha'J'-\alpha J) = gA(\alpha'J'-\alpha J) N(J'M').$$

The gA-values may be found experimentally by measuring the populations of the sublevel and the intensity of the line.

In BFS the lifetimes of atomic levels are measured directly. Thus a relation must be sought that will provide the total transition rate (A(a'J')) from one particular level. This may be done by summing all the transition rates out of the level of interest. Therefore one has

$$A(\alpha'J') = \sum_{J,\alpha} A(\alpha'J'-\alpha J).$$

The lifetime ( $\tau$ ) of a level is defined to be the inverse of the transition rate, and it is the time needed for the population of the level to decrease by 1/e of its original value.

By observing the lifetime of a single line originating in the level of interest, when all transitions from that level are taking place, one is provided with the total lifetime of the level. This is true because

the number of transitions per unit time from the level of interest to a particular final level is proportional to the number of atoms  $(N(\mathbf{a}'J'))$  in the level at that time. But N obeys the time dependant equation

$$dN/dt = N/7$$
,

and will be given by  $NA(A'J'-\alpha J)$ . Therefore, when all allowed transitions occur each transition out of level  $(\alpha'J')$  will be characterized by a decay constant which is the total lifetime of the level. In order to transform the measured lifetime into the corresponding gA-values for a specific transition. More experimental information is needed. This is the branching ratio which is defined as the ratio of the transition rate of  $I(J') \geq I(J)$  to the total transition rate.

As has been shown, atomic lifetimes and transition rates are different quantitative measures of the same phenomena. The oscillator strength or f-value is yet another way of describing the rate of an atomic transition. The oscillator strength may be defined by comparing the strength of absorption by a quantum mechanical atom with that which would be incurred by an electron moving in a potential characterized by the resonant absorption frequency (). If, for a particular frequency, there are N such classical absorbers per unit volume, and N' atoms per unit volume, then f is defined by N = N'f, and is the number of classical absorbers per atom. In terms of the transition rate the f-value is given by

$$gf = gA mc^3/8 \pi^2 e^2 + 2 = 1.4493 \times 10^{-5} m^{-2} gA/4^2$$

where m is the mass of the electron and e is the charge on the electron in esu. Since the natural lifetime (T) of a classically absorbing electron is

$$T = 3mc^3/8e^2\pi^2v^2$$

the preceeding equation may be written as

$$gf = gA T/3.$$

This relationship indicates the intrinsically classical nature of f-values. It is important to realize that the two values of g are not the same and may not be factored from the equation.

### The Energy Loss: Procedure

The energy loss suffered by the chromium ions traversing the foil is of great importance in the calculation of the level lifetimes, as it is necessary to express the distance from the foil as a function of time. This distance is related to the time of flight of the ion by

$$t = d (m/2(E-\Delta E))^{\frac{1}{2}},$$

where E and m are the energy and mass respectively of the incident ion, d is the perpendicular distance between the foil and the axis of the lens, t is the time it takes for the ion to traverse d, and  $\Delta E$  is the energy loss.

The problem of calculating the stopping power of a low energy ion traversing matter has been explored by Lindhardt. 8,9 Calculations

of energy losses were made using the method of Lindhardt and a program written by Brand, Fox, and Keller. <sup>10</sup> The results for several relevant choices of ion, energy, and thickness of carbon foil are listed in Table I.

Table I

Theoretical energy loss calculations for 89 keV Ar, 85 keV Cr, and 100 deV Cr

Ion	$x(gm/cm^2)$	E(keV)	E(keV)
Ar	2.4	89	11
Cr	2.0	85	13
Cr	2.0	100	13

In order to investigate the validity of these calculations, the energy loss for an 89 keV Ar ion was measured. Argon was used rather than chromium because the argon beam is easily produced. Since it has an atomic number (Z=17) close to that of chromium (Z=24), the results of the experiment provide a useful comparison to the calculation.

The beam for the energy loss experiment was supplied by a 150 kV linear accelerator, equipped with a Physicon universal ion source. The Physicon source is capable of producing ionized gas from either gaseous or solid materials. The argon beam was produced by aklowing argon gas to enter the source at a rate which produced a partial pressure in the accelerator of approximately  $6\times10^{-6}$ mm of Hg. The gas was ionized in the source and extracted from it by a -10 kV potential on an extractor electrode. The beam was then accelerated to an energy of 89 keV, mass analyzed at the magnet, and directed into the target chamber, as shown

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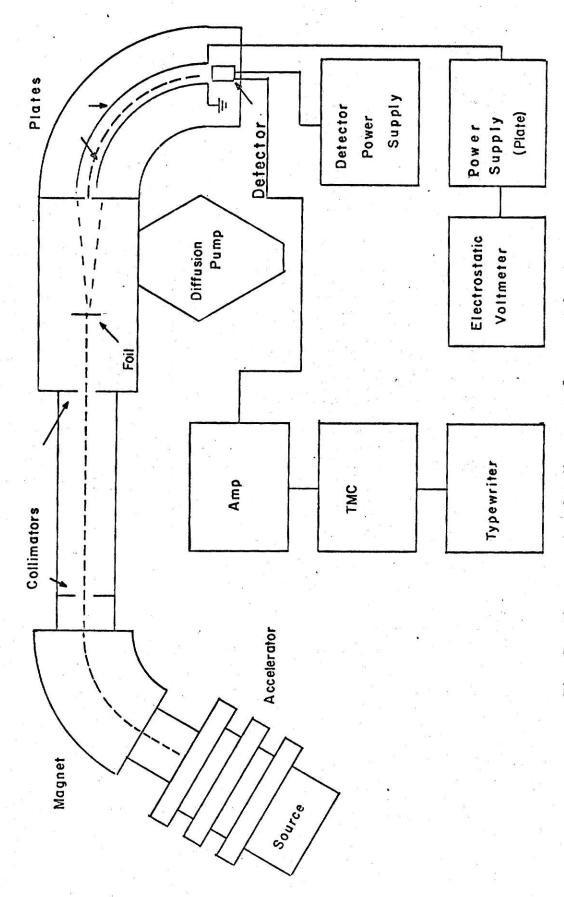


Fig. I. -- The apparatus for the energy loss experiment

in Figure I. The argon beam was identified by maximizing the beam current of the most intense beam at a Faraday cup just downstream of the magnet. The argon leak was then closed; if the beam current dropped dramatically, it was assumed that the argon beam had been identified.

The size of the beam spot was restricted by the use of two \(\frac{1}{4}\) in. collimators, and vertical control of the beam was accomplished with the use of electrostatic deflection plates powered by a 400 V variable voltage supply. (The plates and power supply are not shown in Figure I.) A vacuum was maintained in the target chamber by a 4 in. diffusion pump located several inches downstream of the foil.

An electrostatic analyzer, built by Curnutte, was used to measure the energy of the beam before and after passage through the foil, the energy loss being the difference between the two values. The analyzer consisted of two curved parallel plates preceded by a 0.5 mm aperture. Particles were detected by a Bendix Spiraltron electronmultiplier located at the exit aperture of the analyzer.

The signal from the Spiraltron was amplified by a spectroscopic amplifier and recorded on a TMC (Technical Measurements Corporation) multi-channel analyzer, model 404-6. The voltage for the analyzer plates was supplied by a Spellman high voltage power supply, and was calibrated with an electrostatic voltmeter to ± 0.1 kV (corresponding to approximately ± 1.06 keV in particle energy). In order to record the energy spectrum on the TMC, the plate voltage was varied manually at a rate of .03 kV/sec, while the TMC address was advanced internally by 1 channel/sec.

The beam was electrostatically analyzed without a foil from which the ratio of the beam energy to the plate voltage of the electrostatic analyzer was determined to be 0.0945 kV/keV of particle energy.

Several energy loss measurements were then made on a single foil. It was noted that as time elapsed the energy loss of the beam increased. Therefore energy loss measurements were made using a new foil noting carefully the time elapsed during each run. The foil used for these runs had been damaged in mounting and thus small holes were present in the foil. Under these circumstances, the energy spectrum contained a peak for the beam of particles that passed through the foil and a second peak for the unperturbed particles, which passed through the holes. Three additional runs were made with unused foils to make possible a calculation of the average energy loss.

The foils used throughout the experiments were provided by the Arizona Foil Company. The thicknesses quoted here, ranging from  $10\,\mu\text{gm/cm}^2$  to  $2.0\,\mu\text{gm/cm}^2$ , are those measured by the supplier. Thin carbon foils are not durable if self-supporting, and therefore were mounted on 80% transmission electroplated nickel mesh. Several foils could be placed in the chamber simultaneously, and could be moved in and out of the beam while the system was under vacuum.

#### The Energy Loss: Results

The results of the energy loss experiment may be divided into two parts: first, the thickening of the foils; second, the actual energy loss incured by the beam.

Figure II shows the gradual thickening with time of a carbon foil, which was initially 2.4  $\mu$ gm/cm<sup>2</sup> in thickness. This time dependence is probably better described as a dependence on the amount of charge

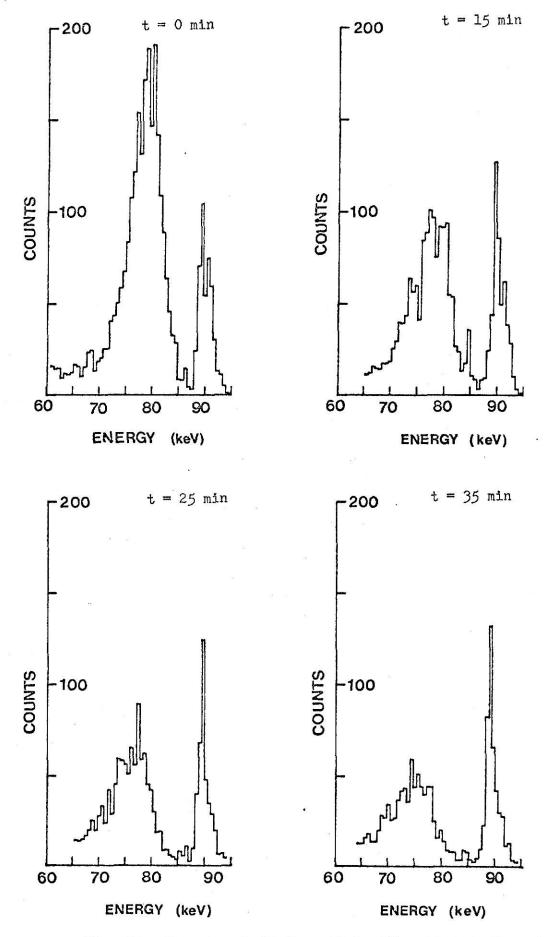


Fig. II. -- The energy spectrum of Ar after passage through a carbon foil

delivered to the foil. The incident beam was 89 keV Ar+, and corresponds to the peak of higher energy in each graph. The pressure in the system was  $10^{-5}$ mm of Hg. The time indicated is the time at which each run began, a typical run duration being 3 minutes. The beam current could not be measured directly while the electrostatic analyzer was in use, but it was measured at a forward Faraday cup before the experiment. The percentage of current transmitted to the foil from this cup is normally about 30%, but transmission has been as high as 50%. Therefore from the measurement of 13 MA at the Faraday cup, one may conclude that the foil actually received a current on the order of several microamperes. It was found that the energy loss increased from 10 keV to 14 keV in a period of 35 minutes. This implies that the foil thickened at a rate of 0.015 mgm/cm2-min. This effect is attributed to the presence of hydrocarbons (pump, oil, etc.) in the system. The build up probably occurs when a molecule very close to the foil is hit by an incoming ion. Dissociated, the molecule may form new bonds with the foil or other molecules. It is hypothesized that better vacuum would reduce the size of the effect. Precautions were taken both in subsequent energy loss experiments and in the lifetime experiments to avoid this change in foil thickness. For the energy loss measurements each foil was used only once, while in the lifetime experiments the time each foil was exposed to the beam was kept below 15 minutes.

An average energy loss of 10 keV was measured using 2.4 \mu\_gm/cm<sup>2</sup> foils. The foils were bombarded for approximately 2 min. The agreement between this measurement and the previous calculation of 11 keV (see Table I) strongly supports the validity of our use of the Lindhardt calculations for the chromium lifetime analyses.

## The Spectral Scans: Apparatus

The same acceleration, collimation, and pumping systems were utilized for the spectral scans as for the energy loss experiment (see Figure III). Solid chromous chloride was used in the source to produce the chromium beam. The Cr<sup>+</sup> beam was identified by first producing an Ar<sup>+</sup> beam as in the energy loss experiment. A Hall probe was then used to measure the magnetic field of the deflection magnet. The voltage measured across the Hall probe is proportional to the magnetic field. The magnetic field needed to turn an ion of given energy through a specified radius of curvature is proportional to the square root of the mass of the ion. Thus, for Ar<sup>+</sup> and Cr<sup>+</sup> beams the following formula gives the projected Hall probe voltage (V<sub>2</sub>) for the chromium beam,

$$V_2 = 1.14 V_1$$

where  $\mathrm{V}_{\mathrm{l}}$  is the Hall probe voltage measured when argon was used.

An adjustable LiF lens, with a focal length of 4.5 in. at 4000 A, was placed perpendicular to the beam line and downstream of the foil. The purpose of this lens was to focus the light emitted by the decaying ions on the slits of the monochromator. Survey scans were made using 6  $\mu$ gm/cm<sup>2</sup> foils. The position of the foil was such that a maximum amount of light was focused on the slits of the monochromator. The monochromator used was a McPherson model 218 vacuum ultraviolet monochromator. Two gratings were available for use with the monochromator; a 1200 line per mm grating blazed at 3000A which produced a dispersion in the system of 26.5 A/mm of slit width, and a 2400 line per mm grating blazed at 1500 A which produced a dispersion of 13.3 A/mm.

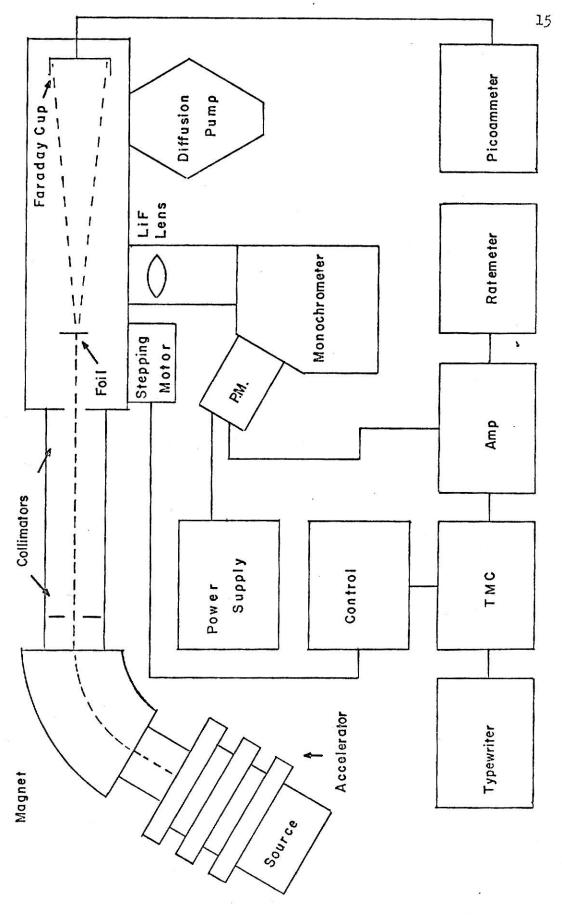
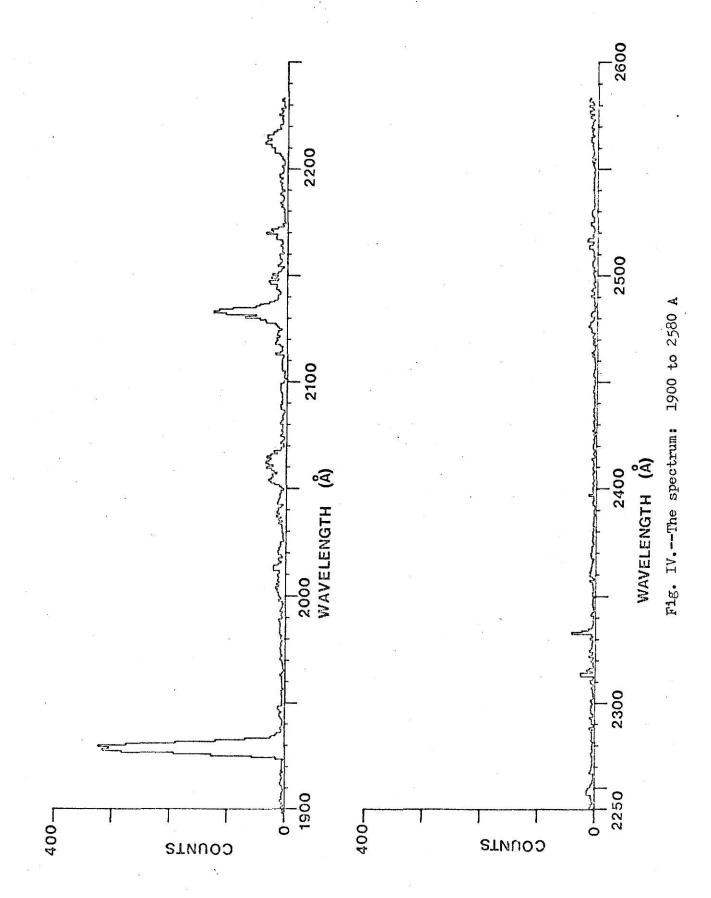


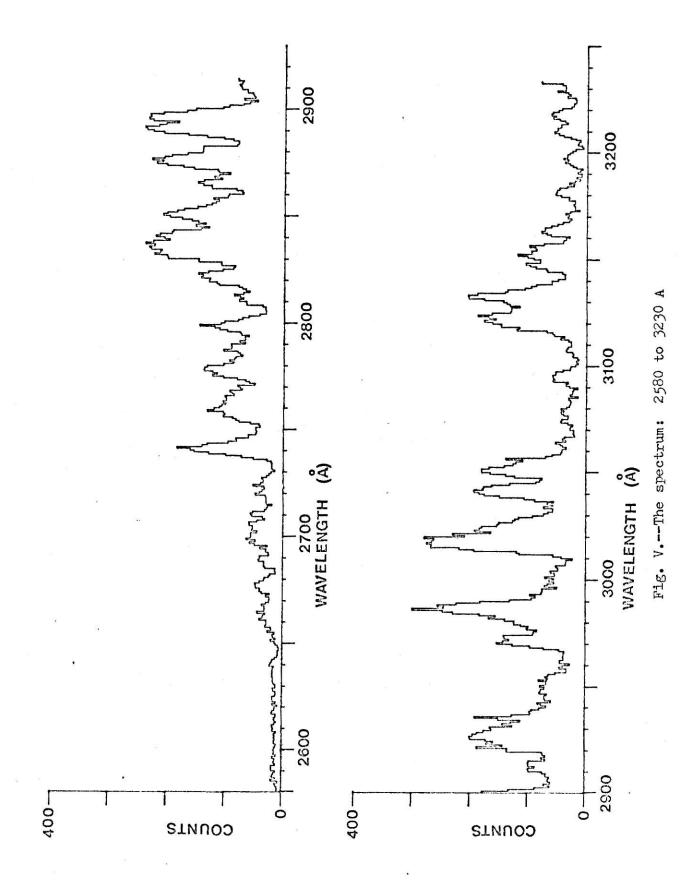
Fig. III. -- The apparatus for the spectral scans and for the lifetime experiment

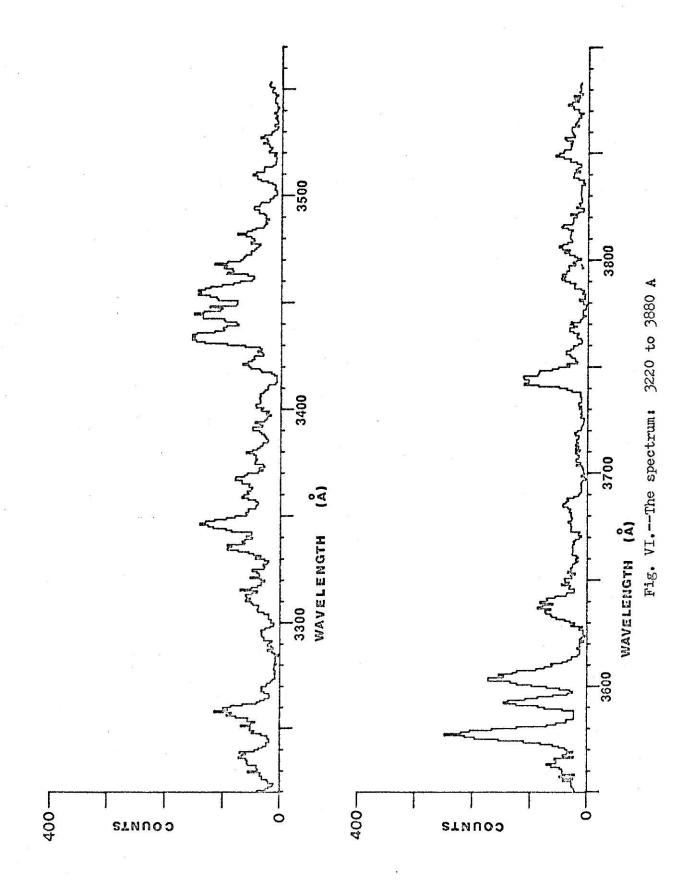
Photons of the proper wavelength were detected with an EMI 6256S photomultiplier tube. The tube was enclosed in a copper case and was cooled by means of a thermoelectric cooling device. A voltage of 1.5 kV was supplied to the tube by a Fluke model 404M power supply. A Nuclear Data Corporation dual channel amplifier and single channel analyzer model ND500, was used to amplify the photomultiplier signal and discriminate against low level noise. This amplifier supplies two output signals, positive pulses which were counted by the TMC, and negative pulses which were monitored on a count rate meter. The electrostatic analyzer had to be removed, and was replaced by a Faraday cup which was used to monitor the beam current.

Two sets of scans were made:

- 1) Low resolution scans were made using the 2400 line/mm grating. The monochromator slits were set at 250 microns to produce an instrumental resolution of 3 A. The scanning rate was 50 A/min. and the TMC was operated in the internal channel advance mode at one channel per second. Thus the TMC recorded all counts produced within a wavelength range of 5/6 A in a single channel. Seven overlapping scans of 400 channels (330 A) each were taken to cover the wavelengths between 1900 and 4100 A. The results of these scans are presented in Figures IV IX.
- 2) High resolution scans of selected regions were taken in a similar manner using the 1200 line/mm grating. Slit widths for these scans were reduced to 150 or 200 microns depending on the resolution needed. The TMC was paced at 1 channel/sec., and the monochromator scan rate was 5 A/min. The length of these scans was approximately 50 A. The resulting spectra are shown in Figures IX XIII.







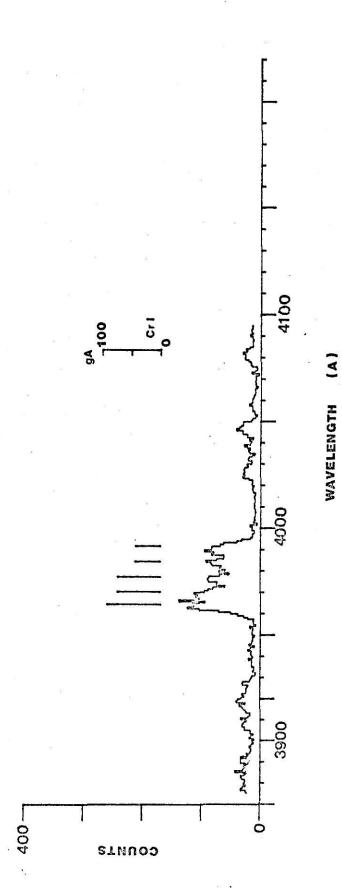


Fig. VII--The spectrum: 3870 to 4100 A

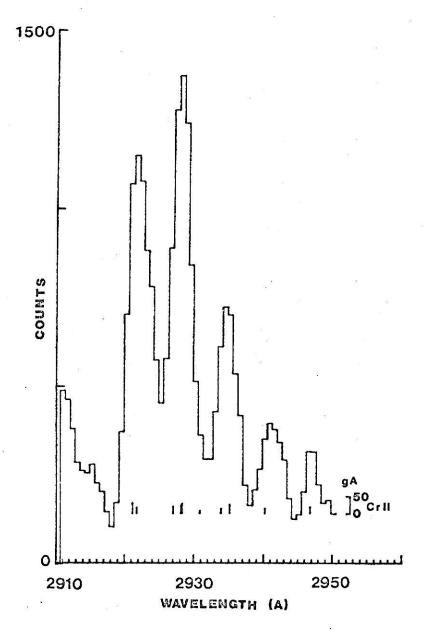


Fig. VIII. -- The high resolution spectra of CrI and CrII: 2910 to 2950 A  $\,$ 

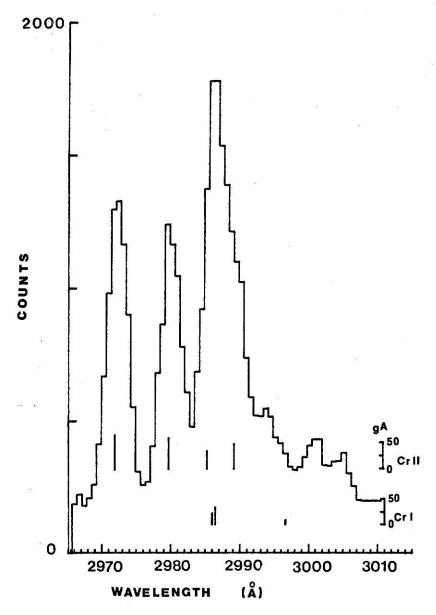


Fig. IX.--The high resolution spectra of CrI and CrII: 2970 to 3010 A

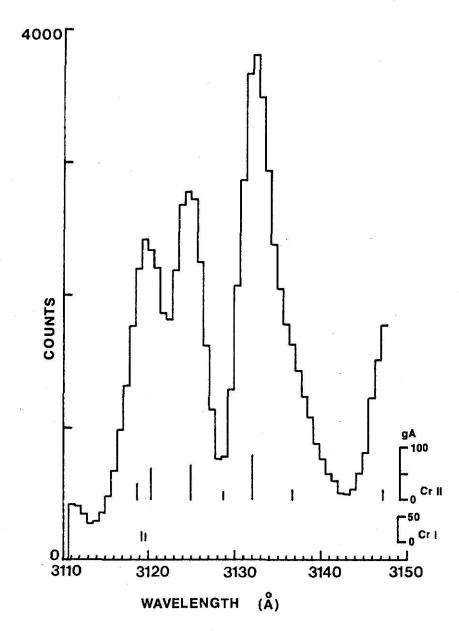


Fig. X.--The high resolution spectra of CrI and CrII: 3110 to 3150 A

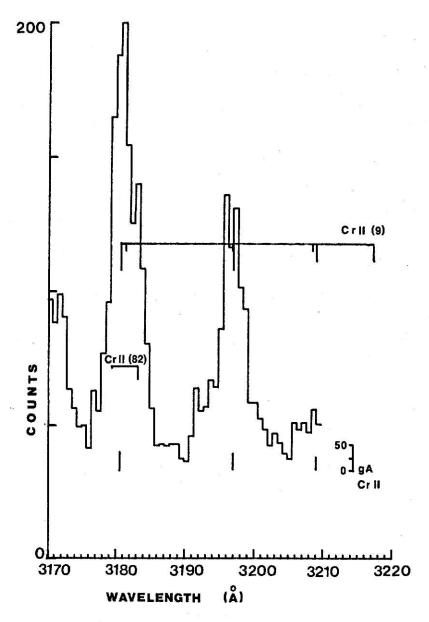


Fig. XI.--The high resolution spectra of CrI and CrII: 3170 to 3220 A

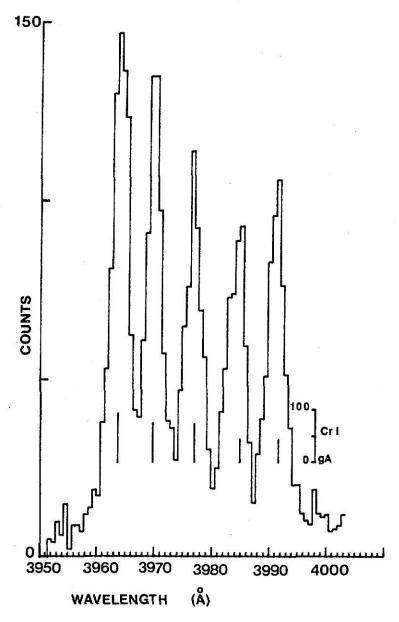


Fig. XII.--The high resolution spectra of CrI and CrII: 3950 to 4000 A

#### The Spectra: Analysis

The spectral analysis provides the experimenter with the basic information needed to decide which lines are appropriate for lifetime analysis. The procedure used in this analysis consisted of three steps:

- 1) A graph of the TMC output was made (counts versus channel, and the appropriate wavelength scale was superimposed on the graph. This produced a graph of counts versus wavelength. Selected gA values (gA > 10<sup>9</sup> sec<sup>-1</sup>) of CrI and CrII, taken from the tables of Corliss and Bozman CB), <sup>10</sup> were then plotted on the same graph.
- 2) Preliminary line identifications were made on the basis of the correspondence between intensity and gA-value. The multiplets to which these lines belong were then found by consulting Moore's A Multiplets Table of Astrophysical Interest and An Ultraviolet Multiplet Table. 13 If other members of identified multiplets were present with the required intensity, the identification was assumed to be correct. Those CrII multiplets identified in our spectra which seemed to be well resolved and of sufficient intensity for lifetime analysis were noted.
- 3) High resolution scans were taken in the vicinity of those multiplets chosen to detect any unresolved structure that could influence the lifetime analysis. Table II gives a list of CrI and CrII lines identified.

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	Lines of Cr	I and CrII	identified in	this spectru	ım
Wavelength (A)	(10 <sup>8</sup> nsec <sup>-1</sup> )	Wavelength (A)	gA (10 <sup>8</sup> nsec <sup>-1</sup> )	Wavelength (A)	gA (10 <sup>8</sup> nsec <sup>-1</sup> )
2739.38 2741.07 2742.17 2986.00 2986.47	26. 36. 37. 23. 34.	3639.80 3743.58 3743.88 3757.66 3768.24	77. 42. 50. 20. 18.	3050.14 3118.65 3120.37 3124.94 3132.06	44. 32. 60. 66. 85.
3014.92 3017.57 3021.56 3039.78 3053.88	21. 44. 47. 29.	3804.80 3963.69 3969.75 3976.66 3983.91	74. 93. 75. 74. 45.	3136.68 3147.23 3180.70 3197.08 3209.18	20. 20. 35. 32. 25.
3109.34 3110.86 3119.25 3119.71 3148.44	10. 10. 18. 15. 29.	3991.12	цц. CrII	3217.40 3295.43 3360.30 3368.05 3378.34	20. 12. 35. 46. 12.
3155.15 3163.76 3237.73 3251.84 3257.82	35. 35. 20. 41.	2055.52 2061.49 2065.42 2677.16 2800.77	9.1 7.3 4.8 132. 24.	3379.83 3421.21 3422.74	21. 18. 25.
3259.98 3346.02 3346.74 3433.60 3436.19	31. 29. 28. 44. 26.	2812.01 2818.36 2822.01 2822.37 2830.47	27. 21. 15. 56. 55.		
3445.62 3453.33 3455.60 3467.02 3467.72	28. 30. 20. 11.	2840.02 2851.36 2897.67 2911.68 2921.24	36. 39. 19. 16. 33.		
3469.59 3481.30 3481.54 3494.97 3510.54	13. 14. 15. 15. 21.	2927.08 2928,15 2930.85 2935.14 2940.22	23. 25. 11. 28. 17.		
3566.16 3578.69 3593.49 3605.33 3636.59	38. 8.3 7.0 5.2 43.	2946.84 2971.91 2979.74 2985.32 2989.19	22. 66. 61. 36. 49.	# #*	e .

## The Lifetimes: Experimental Procedure

In addition to the apparatus used for the spectral scans, the lifetime experiment required the use of a stepping motor and a control box (see Figure III). The stepping motor was used to step the foil up and down stream, the length of one step being 0.258 mm. The control box provided stepping voltage to the motor and a channel advance to the TMC. Since all lines chosen for analysis were in the region of 3000 A, the grating blazed at 3000 A was used for the lifetime runs. The foils used were measured by the Arizona Foil Company to be 2.0  $\mu gm/cm^2$  in thickness.

A lifetime measurement was made at 85 and 100 keV for each of the lines. Spectral lines were located by setting the monochromator for the appropriate wavelength and maximizing the count rate meter in that wavelength region. The slit widths were set at 100 or 150 microns depending upon the resolution required. The foil and the TMC were stepped at the same rate, i.e. 1.04 sec/channel and 1.04 sec/motor step. At this rate one run of 400 channels took approximately six minutes.

One run consisted, in most cases, of stepping the foil upstream a distance of approximately 3 cm and back downstream the same distance. This produced two decay curves for one run. In some cases the counting rate was such that the reversal was unnecessary. During each run the chamber was isolated from the accelerator for several seconds in order to measure the dark count. The dark count was measured to be from four to six counts per channel.

The beam current was monitored continuously in order to note any large fluctuations in the amount of beam delivered to the foil. When

changes of more than 15% occurred the run was discontinued. During most runs the beam current was quite steady and fluctuated no more than 5%. Because of the hydrocarbon buildup mentioned in the energy loss experiment the foils were not used for more than 15 min. The time elapsed on each foil was noted and is stated in the heading of each computer output (see the Appendix).

#### Lifetimes: The Results

Lifetime data was taken on five lines originating from four atomic levels in CrII. Figure XIII is a partial Grotrian Diagram showing the transitions whose lifetimes were measured and possible cascades into these levels of interest. A measurement was made on each line at two energies (85 and 100 keV). Each decay curve was analyzed separately, and an average over all of the data for each level was used to obtain the stated lifetimes.

The lifetimes were calculated from the decay curves using a four parameter least-squares fit to the double exponential function

$$F(t) = A \exp(-t/\tau_1) + B \exp(-t/\tau_2) + D$$

where D is the measured dark count, and A and B are the amplitudes of the decays. The computer program used was written by Brand, Fox, and Keller. The Appendix is composed of photographic reductions of the final computer output.

Brand 14 has indicated that the fitting program will resolve a short component and a longer lived "cascade" component, but that the program will not resolve two short lived components. Therefore it is imperative that one know whether or not strong cascading occurs. Table III provides

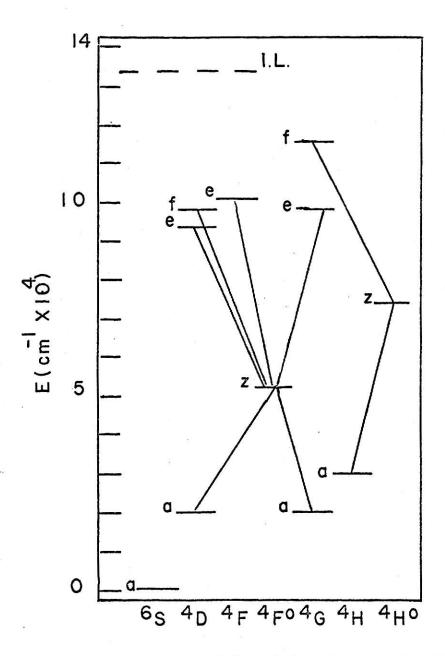


Fig. XIII. -- A partial Grotrian Diagram of CrII

Table III

Possible cascades into levels of interest

Wavelength I	Level	Configuration
Transitions into	0=	3d <sup>4</sup> (a <sup>3</sup> H)4p
2415.23 5w	f <sup>4</sup> G <sub>5<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a. <sup>3</sup> G)5s
Transitions into	z <sup>4</sup> H <sup>o</sup> 5½	3d <sup>4</sup> (а <sup>3</sup> н)4р
2404.72 2w	f <sup>4</sup> G <sub>5</sub> 1/2	3d <sup>4</sup> (a <sup>3</sup> G)5s
2408.02 3w	f <sup>4</sup> G <sub>4<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> G)5s
Transitions into	z4F0	3d <sup>4</sup> (a <sup>5</sup> D)4p
3049.49 10w	11	3d <sup>4</sup> (a <sup>5</sup> D)5s
2569.40 152	<b>)</b> 1	3d <sup>4</sup> (a <sup>5</sup> D)4d
2577.74 10w	e <sup>4</sup> F <sub>3</sub> 1	3d <sup>4</sup> (a <sup>5</sup> D)4d
2661.22 50w	e <sup>4</sup> G <sub>5</sub> 1/2	3d <sup>4</sup> (a <sup>5</sup> D)4d
2674.26 7w	LL	3d <sup>4</sup> (a <sup>5</sup> D)4d
Transitions into	z <sup>4</sup> F <sup>0</sup> 3½	3d <sup>4</sup> (a <sup>5</sup> D)4p
3056.66 8w,	LL LL	3d <sup>4</sup> (a <sup>5</sup> D)5s
2642.60 2w	f <sup>4</sup> D <sub>3</sub> 1	3d <sup>4</sup> (a <sup>5</sup> D)4d
2652.78 3w,	1 f <sup>4</sup> D <sub>2<sup>1</sup>/<sub>2</sub></sub>	3d4(a5D)4d
2567.50 5w	e <sup>4</sup> F <sub>3<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>5</sup> D)4d
2576.45 2w	e <sup>4</sup> F <sub>2<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>5</sup> D)4d
2663.28 30w	,1 e <sup>4</sup> G <sub>μ</sub> ,	3d <sup>4</sup> (a <sup>5</sup> D)4d
2674.07 8w	e <sup>4</sup> G <sub>3½</sub>	3d <sup>4</sup> (a <sup>5</sup> v)4d

a listing of all transitions into the levels studies given by Keiss. 15
There are no lines listed that occur in the spectral scans with a significant counting rate. The generally low intensities given by Keiss for these lines also suggest that one should not expect short lived cascades. Under these circumstances it was assumed that no strong cascades need be considered.

Table IV provides a summary of the lifetime analyses. The average value for each line are given for 85 and 100 keV, and an average lifetime for each level is stated. Note that there is no systematic relationship between the measured lifetime and the energy at which it was taken. This indicates that if a trend of this kind is present, its magnitude must be within the limits of the experiment. The errors quoted in Table V are based on the standard deviations of the lifetime measurements, and on an estimated possible error in the energy loss of 50%. The standard deviations ran between 0.1 and 0.4 nsec., while the error in energy loss produced an error in the lifetimes of 0.2 nsec.

Table IV.

The lifetimes measured at 85 and 100 keV for five lines

Wavelength (A)	Level	Lifetimes (nsec.) 85 keV 100 keV		Level	
2971.9	z <sup>4</sup> H <sup>0</sup> 6½	5•3	4.6	5.0 <u>+</u> 0.6	
2979.7	z <sup>4</sup> H <sup>0</sup> 5½	4.9	4.7	4.8 <u>+</u> 0.3	
3124.9	$z^{4}F^{0}3^{\frac{1}{2}}$	5.0	4.9	5.0 ± 0.4	
3132.1	$z^{4}F^{0}4^{\frac{1}{2}}$	4.9	5.2	1.0.4.0.1	
3180.8	$z^{4}F^{0}4^{\frac{1}{2}}$	4.8	4.7	4.9 ± 0.4	

Since branching ratio measurements are needed before gf values can be derived for the transitions, the BF data cannot be compared directly to previous measurements of gf-values. To avoid this difficulty, lifetimes were derived from all gA-values given by CB for transitions out of the level of interest. A comparison was then made on the basis of these lifetimes (see Table V). These lifetimes were derived by summing all gA-values for transitions out of the level of interest and dividing by the g of the upper level. This is the total transition rate out of the level and its inverse is the lifetime of the level. The ratio of these lifetimes to BFS lifetimes should be equivalent to a ratio of gf-values. Thus the ratios of gf-values given in Table V compared to the ratio of lifetimes provides a method of intercomparing our lifetime results with gf-values measured by other authors. Shackleford 14 has measured a gf-value for one line and Byard 15 has measured the gf-values for two lines originating from the  $z^4 F^0_{L^{\frac{1}{2}}}$  level. It may be seen that these results give gf/gf(CB) in fair agreement with our  $\gamma$  (CB)/ $\gamma$  (BF) for that level.

One should note that the ratio of lifetimes is a factor of two higher for the lines at 2971.9 A and 2979.7 A. The reason for this discrepancy may be that some transitions out of the z H term were not seen by Corliss and Bozman. If this were true  $\tau$  (CB) would be reduced, producing a more uniform ratio of lifetimes. In particular, lines at 2297.17 and 2307.19 A were not observed in the present spectrum nor does CB list gA-values for them, but Keiss lists them with substantial intensities (Table VII-VIII). This result may indicate experimental problems in observing this part of the spectrum (2250 to 2650 A) since there seems to be no structure in this region (see

Figure V). Although, the same arguments could be made for the  $z^4F^0$  levels, it is believed not to be the case since there is good agreement with gf-values measured by Schackleford and Byard gf-values for those transitions.

Table V.

A comparison of the measured lifetimes to the results of CB, Shackleford, and Byard

Wavelength (A)	(BF)	ブ(CB) (nsec)	<b>プ</b> (CB)/ <b>プ</b> (BF)	gf(OSU)/gf(CB)	gf(S)/gf(CB)
2971.9	5.0	1.8	•36	W 6)	,
2979.7	4.8	1.7	.38		
3124.9	5.0	0.8	.16	**	<i>18.</i>
3132.1		0.7	.14	.09	.10
3180.8	4.9	υ.γ	• 14	.16	

## The Solar Photospheric Abundance of Chromium

Two measurements are necessary before the calculation of the chromium abundance is possible. First, the branching ratios for the measured levels must be found, and second, the equivalent widths for the lines out of these levels must be measured in the solar spectrum. The branching ratios can easily be measured with conventional light sources. Problems exist in the measurement of the equivalent width of ultraviolet lines in the solar spectrum. The intensity of the continum in the UV region is low and the number of lines is large. Thus the exact height of the continum is not well known. This is not as much of a problem in the visible region and, as can be seen in Tables VI-IX,

Table VI Transitions from the  $z^4 H^0_{5\frac{1}{2}}$  (3d $^4$ (a $^3$ H)4p) level

Wavelength	I	gA	Level	Configuration	Solar Eq. Width
3552.42	2		ъ <sup>2</sup> н <sub>5<sup>1</sup>/<sub>2</sub></sub>	3a <sup>5</sup>	+
3540.23	2		ь <sup>2</sup> н <sub>ц1</sub>	3a <sup>5</sup>	+
3442.98	ı		a <sup>2</sup> H <sub>5<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> H)4s	+
3421.62 <sup>(a)</sup>	5		a <sup>2</sup> H <sub>4<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> H)4s	2.3 min.
3315.28	12		ъ <sup>4</sup> G <sub>5<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> G)4s	blended
3307.02	50	9.2	ъ <sup>4</sup> G <sub>4½</sub>	3d <sup>4</sup> (a <sup>3</sup> G)4s	blended
3225.44	8		ь <sup>4</sup> ғ <sub>4<mark>1</mark>2</sub>	3d <sup>5</sup>	blended
3063.82	7		a4F41	3d <sup>4</sup> (a <sup>3</sup> H)4s	78 m
2988.04	12		a4H <sub>61/2</sub>	3d <sup>4</sup> (a <sup>3</sup> H)4s	blended
2979.73*	80	61	a4H <sub>5½</sub>	3d <sup>4</sup> (a <sup>3</sup> H)4s	?
2972.67	7w		а <sup>4</sup> Н <sub>4 1/2</sub>	3d <sup>4</sup> (a <sup>3</sup> H)4s	?
2307.19	35		a4G41	3d.5	+
2306.81	10		a4G <sub>5<sup>1</sup>/<sub>2</sub></sub>	3d.5	*+

<sup>\*</sup>Lifetime measured in this experiment.

<sup>+</sup>Not observed according to MMH.

<sup>?</sup>Principal contributor to blend.

<sup>(</sup>a)  $a^2H_{5\frac{1}{2}}-z^4H^0_{6\frac{1}{2}}$  may be present.

Table VII Transitions from the z  $^4{\rm H}^o_{6\frac{1}{2}}$  (3d  $^4({\rm a}^3{\rm H})4{\rm p})$  level

Wavelength	I	gA	Level	Configuration	Solar Eq. Width
4195.40	4		b <sup>2</sup> I <sub>6<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>1</sup> I)rs	11 m
3529.73	2		ъ <sup>2</sup> н <sub>5<sup>1</sup>/<sub>2</sub></sub>	3a <sup>5</sup>	, + P
3421.62 <sup>(a)</sup>	5		a <sup>2</sup> H <sub>5<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> H)4s	23
3295.42	50	12	ъ <sup>4</sup> G <sub>5½</sub>	3d <sup>4</sup> (a <sup>3</sup> G)4s	53
2971.90*	75	66	a <sup>4</sup> H <sub>6</sub> 1/2	3d <sup>4</sup> (a <sup>3</sup> H)4s	?
2950.69	7		a <sup>2</sup> I <sub>6<sup>1</sup>/<sub>2</sub></sub>	3a <sup>5</sup>	+
2950 <b>.</b> 10 <sup>(b)</sup>	10		a <sup>2</sup> I <sub>5<sup>1</sup>/<sub>2</sub></sub>	3a <sup>5</sup>	+
2297.17	50		a4G <sub>5½</sub>	3d <sup>5</sup>	+

<sup>\*</sup>Lifetime measured in this experiment.

?Principal contributor to blend.

<sup>+</sup>Not observed by M.M.H.

 $<sup>(</sup>a)_a^2 H_{4\frac{1}{2}} - z^4 H_{5\frac{1}{2}}^0$  transition may also be present.

<sup>(</sup>b)  $_{b}^{\mu}G_{\mu^{1}_{\overline{2}}-z^{2}}I^{o}_{5^{1}_{\overline{2}}}$  transition may also be present.

Wavelength	I	gA	Level	Configuration	Solar Eq. Width
5502.07	40		ъ <sup>4</sup> G <sub>41</sub>	3d <sup>4</sup> (a <sup>3</sup> G)4s	23
5472.60	12		ъ <sup>4</sup> G <sub>3½</sub>	3d <sup>4</sup> (a <sup>3</sup> G)4s	+
5280.08	30		$b^{4}F_{2\frac{1}{2}}$	3a <sup>5</sup>	blended
5279.88	50		b <sup>4</sup> F <sub>4<sup>1</sup>/<sub>2</sub></sub>	3a <sup>5</sup>	18
5274.99	75		b <sup>4</sup> F <sub>3<sup>1</sup>/<sub>2</sub></sub>	3a <sup>5</sup>	<b>blende</b> d
4860,20	20		$a^{4}F_{4\frac{1}{2}}$	3d <sup>4</sup> (a <sup>3</sup> F)4s	9
4848.24	75		a <sup>4</sup> F <sub>3<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> F)4s	52
4836.22	25		a <sup>4</sup> F <sub>2<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> F)4s	blended
3738.38	35		c <sup>4</sup> D <sub>2<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> D)4s	blended
3736.56	1		ь <sup>4</sup> Д <sub>Э<sup>1</sup>/<sub>2</sub></sub>	3a <sup>5</sup>	+
3336.16	. 2		$a^{4}P_{2\frac{1}{2}}$	3d <sup>5</sup>	+
3197.08	75	32	a <sup>4</sup> G <sub>41/2</sub>	3d <sup>5</sup>	blended
3196.93	. 20		a <sup>4</sup> G <sub>3½</sub>	3a <sup>5</sup>	blended
3196.35	3		a <sup>4</sup> G <sub>2½</sub>	3a <sup>5</sup>	39
3147.22 <sup>(a)</sup>	50	20	a <sup>4</sup> D <sub>3<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>5</sup> D)4s	blended
3124.94*	40	66	a <sup>4</sup> D <sub>2<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>5</sup> D)4s	149
2544.26	15		$a^{6}D_{\mu^{\frac{1}{2}}}$	3d <sup>4</sup> (a <sup>5</sup> D)4s	+
2531.84	25	81	a <sup>6</sup> D <sub>3<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>5</sup> D)4s	+

<sup>\*</sup>Lifetime measured in this experiment.

<sup>+</sup>Not observed by MMH.

<sup>?</sup>Principal contributor to blend.

<sup>(</sup>a)  $_{b}^{\mu}G_{\mu^{\frac{1}{2}}}-z^{\mu}G^{o}_{\mu^{\frac{1}{2}}}$  transition may be present.

Wavelength	I	gA	Level	Configuration	Solar Eq. Width
7311.60	2		c <sup>4</sup> D <sub>3<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> D)4s	5
6478.37	50		b <sup>4</sup> G <sub>5<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> G)4s	+
5455.86	8		ъ <sup>4</sup> G <sub>4½</sub>	3d4(a3G)4s	+
5237.35	100		$\mathbf{b}^{4}\mathbf{F}_{4\frac{1}{2}}$	3a <sup>5</sup>	49
5232.54	20		b. F31/2	3a <sup>5</sup>	12
4824.12	100		a <sup>4</sup> F <sub>4<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> F)4s	94
4812.34	25	8	a <sup>4</sup> F <sub>3<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>3</sup> F)4s	41
3715.18	25		ь <sup>4</sup> р <sub>3½</sub>	3d <sup>5</sup>	58
3181.42	20	4.6	a.4 <sub>G.4½</sub>	3d <sup>5</sup>	blended
3180.70*	75	35	a <sup>4</sup> G <sub>5<sup>1</sup>/<sub>2</sub></sub>	3d <sup>5</sup>	<b>ble</b> nded
3132.05*	100	85	a <sup>4</sup> D <sub>3<sup>1</sup>/<sub>2</sub></sub>	3d4(a5D)4s	137
2534.33	40	6.4	$a^{6}D_{4\frac{1}{3}}$	3d <sup>4</sup> (a <sup>5</sup> D)4s	+
2522.01	4		a <sup>6</sup> D <sub>3<sup>1</sup>/<sub>2</sub></sub>	3d <sup>4</sup> (a <sup>5</sup> D)4s	+

<sup>\*</sup>Lifetime measured in this experiment.

<sup>+</sup>Not observed by M.M.H.

<sup>?</sup>Principal contributor to blend.

which are a compilation from Keiss<sup>14</sup> and Moore, Minnaert, and Houtgast<sup>18</sup>, lines from the levels measured do exist in the visible. Thus, a relatively straight forward experiment to measure the branching ratios will produce many gf values for lines which may be useful in the solar abundance problem.

Although, the measured lifetimes alone do not determine the abundance unambigouously, one may derive an approximate abundance by comparison. Warner  $^{19,20}$  has measured gf values in CrII and has derived an abundance of log  $N_{\rm Cr} = 5.47$ , where log  $N_{\rm H} = 12$ , based on those gf-values. Warner's gf-values may not be compared to the BF lifetimes directly, as there are no lines in common, but a comparison of both measurements to CB will produce a correction to the abundance given by Warner.

Table X compares gf-values taken from CB with those of several authors and with the results of this experiment. The value of 0.82 dex for the CB-BF comparison was arrived at on the assumption that strong transitions out of the  $z^{\mu}H^{0}$  levels do exist and were not observed by CB. The correction is thus the log of the average of the two ratios listed in Table VI for the  $z^{\mu}F^{0}$  transitions.

The value obtained by Cocke, Stark, and Evans  $^3$  based on BFS lifetimes for neutral chromium is 5.80. The correction of 0.82 dex from the present study would indicate a value of 5.54 for log  $N_{\rm Cr}$ . Although the branching ratios must be measured before any firm conclusions are possible, one might seek an explanation for the large disparity in the gf-values of log  $N_{\rm Cr}$ , either in the ionization equilibrium for chromium or in the abundance of 5.47 given by Warner.

Table X

A comparison of the measured lifetimes to the gA values of others

0.75 dex
0.89 dex
0.95 dex
0.94 dex
0.60 dex
0.82 dex

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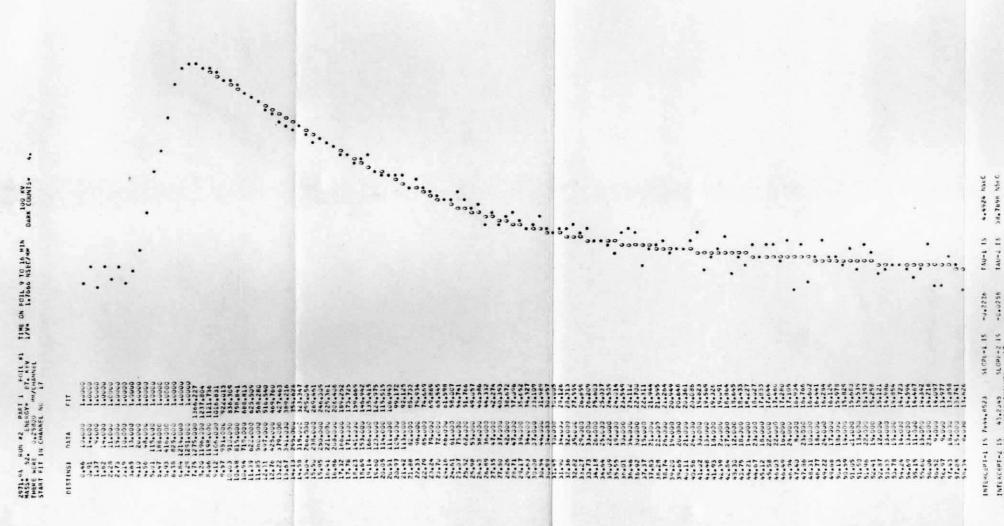
- <sup>17</sup>Byard, P., J.Q.S.R.T., <u>8</u>, 1543 (1968).
- Moore, C. E., M. G. J. Minnaert, and J. Houtgast N.B.S. Monograph, No. 61 (1966).
  - <sup>19</sup>Warner, B., Memoirs R.A.S. <u>70</u>, 165 (1967).
  - 20 Warner, B., Obs. <u>89</u>, 107 (1969).

Figure XIV. A. Part one of the 100 keV lifetime fit for the line at 2971.9 A

## ILLEGIBLE DOCUMENT

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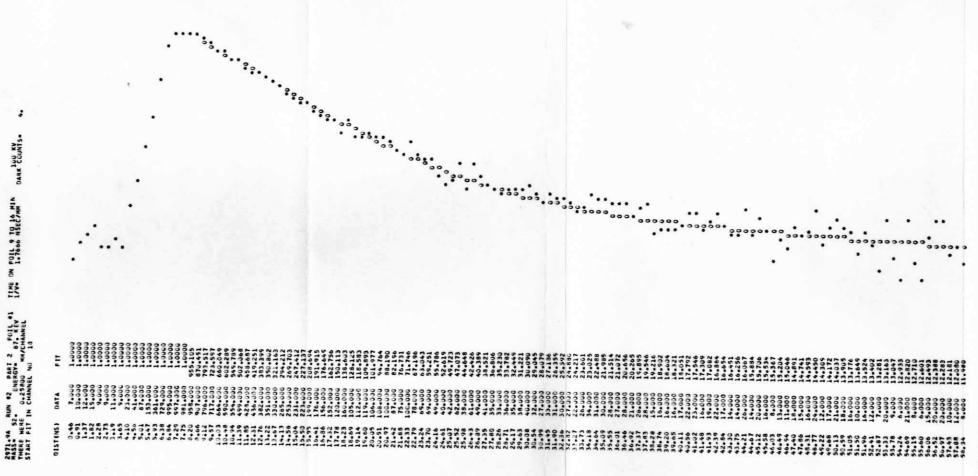


INTERCEPT-1 IS 7444-8523 SLOPE-1 IS -4-2226
INTERCEPT-2 IS 43-23-49 SLOPE-2 IS -0-0256
THE DIAN IS 4 US 40.64 CHANNEL IT
THE DIAN IS 4 US 40.64 CHANNEL IT
RUN PROCESSED 3 JULY 72.

1.77336770 00 1.77336770 00 1.7561560 00 1.74606710 00

-wne

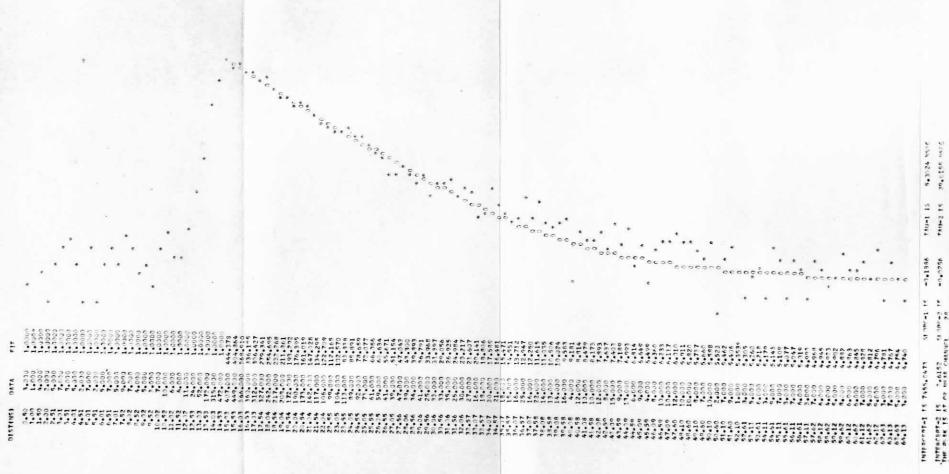
Figure XIV. B. Part two of the 100 keV lifetime fit for the line at 2971.9 A



INTERCEPT-1 IS \$183-6820 SLOPE-1 IS -0.2106 TAU-1 IS 4.7479 NSEC INTERCEPT-2 IS 37.7253 SLOPE-2 IS -0.0267 TAU-2 IS 37.4831 NSEC THE FERL IS 4 OR NEAR CANNEY.
THE RATIO OF COEFFICIENTS AT THE FOIL ISS 33.03660

1.52030650 00

Figure XIV. C. Part one of the 85 keV lifetime fit for the line at 2971.9 A

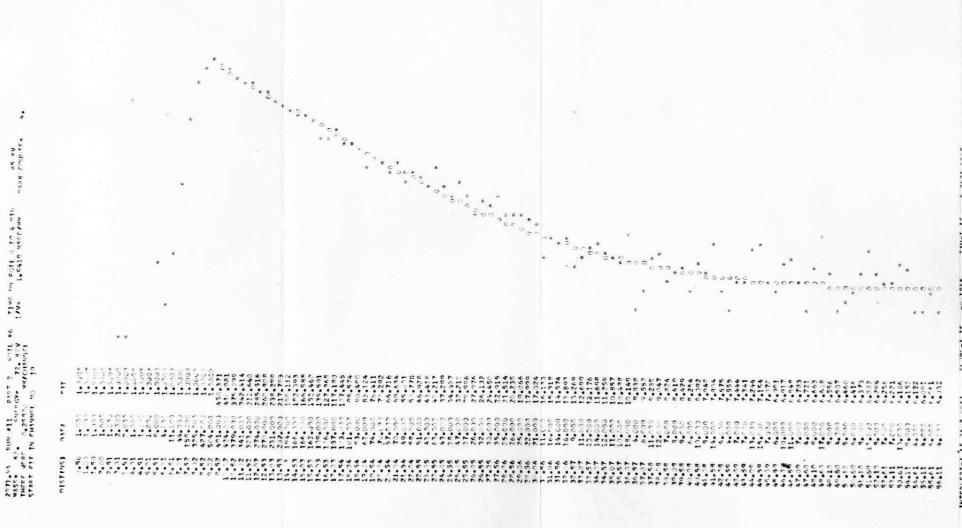


PS KV

2072-0A BINN #11 DART 1 FOIL #6 TIME DV FOIL O TO 4 MIN MASS 57 EVERN 72. KTV 1/V 1.0410 NSFC/AM THERE WERE 0.25800 WOT 70. MINUTEL 1.0410 NSFC/AM START FIT IN CHANNEL NO 70.

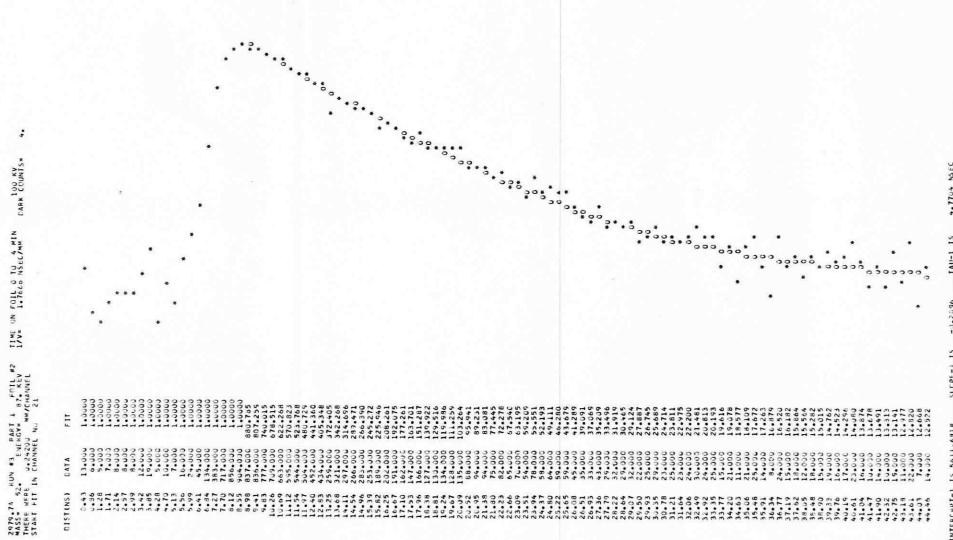
5888 ---

Figure XIV. D. Part two of the 85 keV lifetime fit for the line at 2971.9 A



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Figure XV. A. Part one of the 100 keV lifetime fit for the line at 2979.7 A  $\,$ 

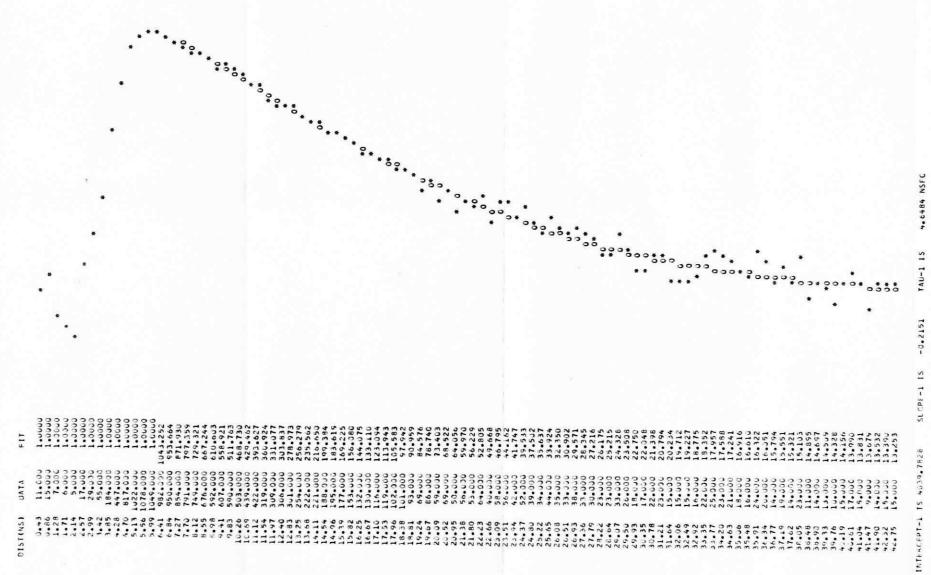


INTERCEPT-1 IS 5611.6918 SLCPE-1 IS -0.2096 TAU-1 IS 4.7704 NSEC INTERCEPT-2 IS 28.646 SLCPE-2 IS -0.4286 TAU-2 IS 34.9206 NSFC THE PERK IS AT OR NEAR CHANNEL 20 THE RATIO OF COFFICIENTS AT THE FOIL IS: 41.08316 SULY 72.

CH150R

1 2-10516350 JU 2 2-10388899 U0 3 2-10269740 U0 4 2-10151299 U0

Figure XV. B. Part two of the 100 keV lifetime fit for the line at 2979.7 A



4

DARK COUNTS=

TIME ON FOIL O TO 4 MIN

2979-74 RUN #3 PART 2 FELL #2 ANSS# 52. BT. KEV THERE WERE 0.24200 MMZCHANNEL START FIT IN CHANNEL NJ 15 INTERCEPT-1 IS 4039-7828 SLOPE-1 IS -0.2151 TAU-1 IS 4.6484

INTERCEPT-2 IS 65.4671 SLOPE-2 IS -0.0256 TAU-2 IS 38.9981

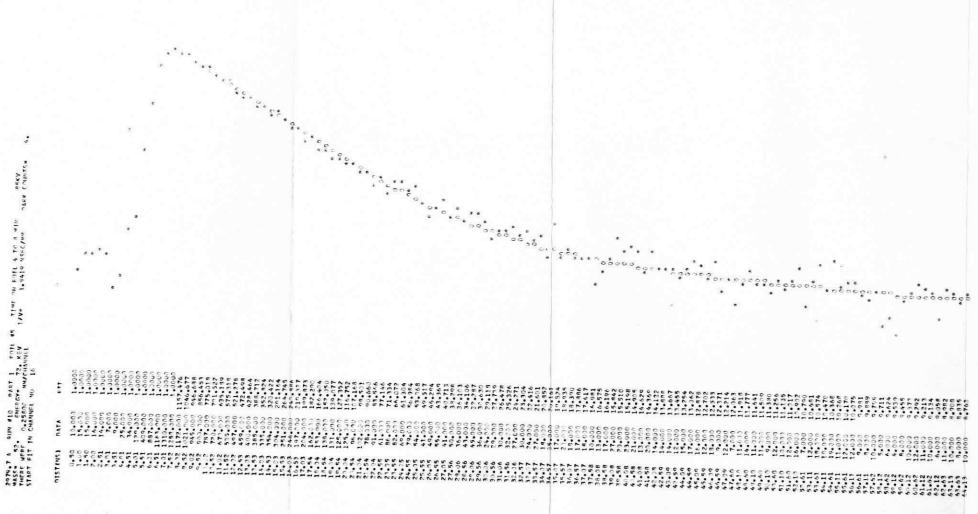
THE PPAK IS AT CR NEAR CHANNEL 13

THE RATIO OF COEFFICIENTS AT THE FOIL IS: 53.24791

RUN PROCESSED 3 JOLY 72.

CHISOR

Figure XV. C. Part one of the 85 keV lifetime fit for the line at 2979.7 A



INTERCEDT-1 IS 5862.4638 SLOBE-1 IS -0.2055 TAU-1 IS 4.8655 HSTC INTERCEDT-2 IS 31.4042 SLOBE-2 IS -0.0294 TAU-2 IS 33.9603 HSTC THE BEST IS A TO WILES CHANNEL IS 50.0294 TAU-2 IS 33.9603 HSTC RIP BATTO DE COMPETITIONES AT THE FOLIO IS 40.70569

Figure XV. D. Part two of the 85 keV lifetime fit for the line at 2979.7 A  $\,$ 

2079,7 & GIN MID DART 2 ESTI ME TIME DA FOIL 3 TO 8 MIN MASS 55, SHORY 77, KEV 170\* 1,0419 NSECZMA THERE METER 0,255013, WITCHANNEL, START FIT IN CHENNEL ND 19 DISTINCT

Figure XVI. A. Part one of the 100 keV lifetime fit for the line at 3124.9 A

4.7537 NSEC 23.9313 NSEC DAKK COUNTS= 3124. A RUN 84 PART 1 FULL 82 TIME UN FULL 4 TO 10 HIN FERENCE 32590 MACCANNEL 1705 1.7666 NSFC/MP START FIT IN CHANNEL NO 18 0000000000 DISTINS

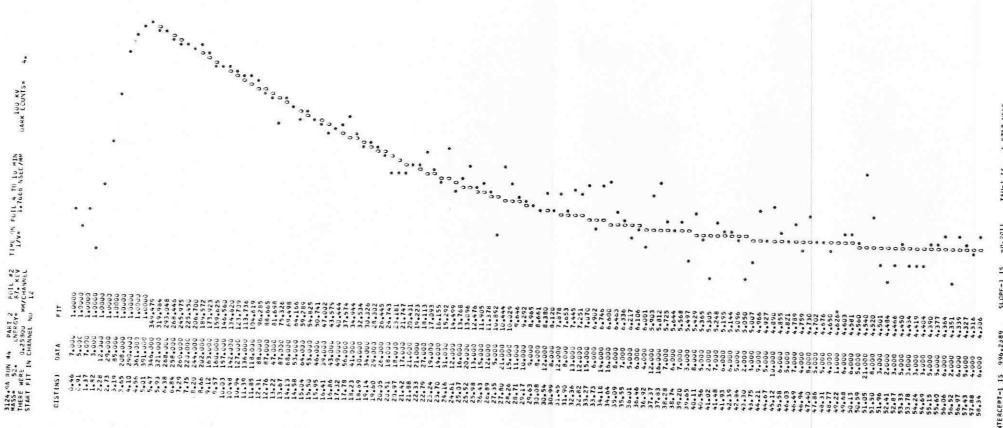
TAU-1 15 TAU-2 15 57.39299 -0.0418 INTERCEPT-2 IS 15.2183 SLCPE-2 IS
THE PEAK IS AT UR NEAR CHANNEL 16
THE RATIU OF COEFFICIENTS AT THE FOIL IS:
RUN PHOCESSED 3 JULY 72. St UPE-1 15 INTERCEPT-1 15 2936.156

2223

1.54181400 1.54181400 1.54062530 1.53949820

4004

Figure XVI. B. Part two of the 100 keV lifetime fit for the line at 3124.9 A

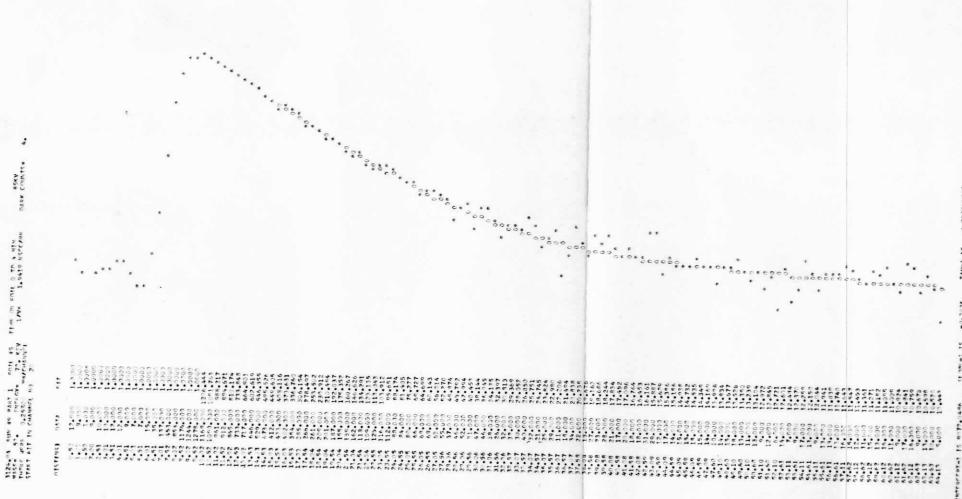


INTERCEPT-1 IS 990.2389 SLOPE-1 IS -0.2011 TAU-1 IS 4.9737 NSEC
THE PERK ISA TOON NEAM CHANNEL

IL 12.5.590.26
RUN PROCESSED 3 JULY TEX TO THE PER PARTO OF CORFFICIENTS AT THE FOIL IS: 25.590.26

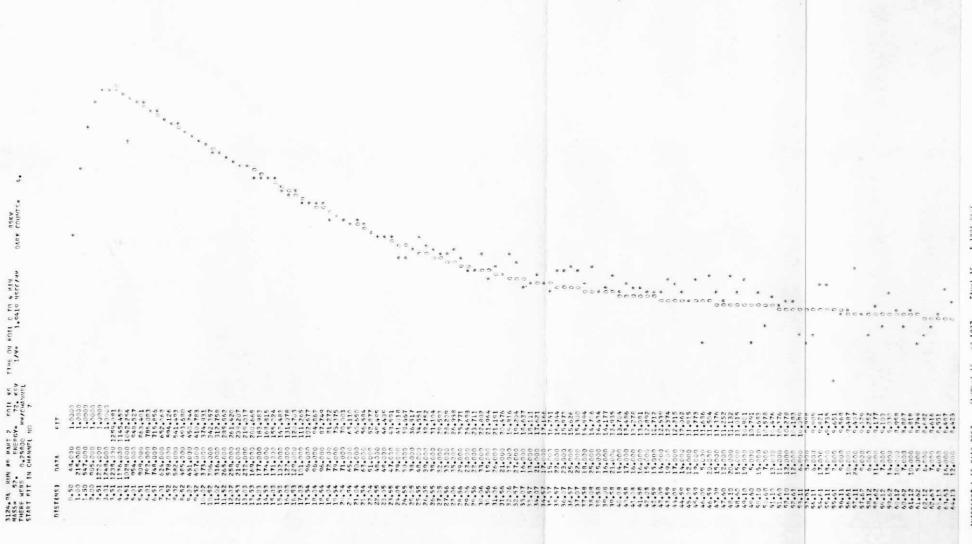
CHISOR

Figure XVI. C. Part one of the 85 keV lifetime fit for the line at  $3124.9~\text{\AA}$ 



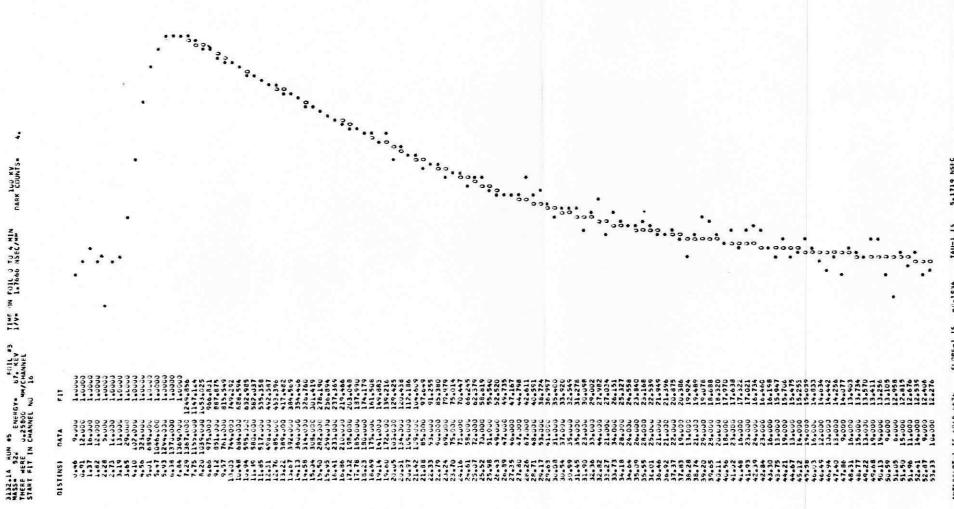
Introduction is unitable state. In -0.2038 Table in 4.013 [Witnessen of the control of the contr

Figure XVI. D. Part two of the  $85~\rm{keV}$  lifetime fit for the line at  $3124.9~\rm{A}$ 



Herreroral is 24,420 seen the relation of the 21 the 21 the 22 the 23 the 23 the 24,420 the 24 the 2

CHISQR 1 - MA137200 00 2 1-64701400 00 3 1-67619710 00 4 1-41709640 00



INTERCEPT-1 IS 4905-0470 SLOPE-1 IS -0+1634 TAU-1 IS 5+1719 NSEC INTERCEPT-2 IS 44-948 SLOPE-2 IS -0+0308 TAU-2 IS 32-4330 NSEC THE PEAK IS 41 UN NEAR CHANNEL THE ATTO 0F COFFICIENTS AT THE FCIL IS! 41-92534 NATIO 0F COFFICIENTS AT THE FCIL IS! 41-92534

CHISGR 1 1-25879380 UD 2 1-228795677 00 3 1-24819700 UU 4 1-24911260 UU

Figure XVII. B. Part one of the 85 keV lifetime fit for the line at 3132.1 A

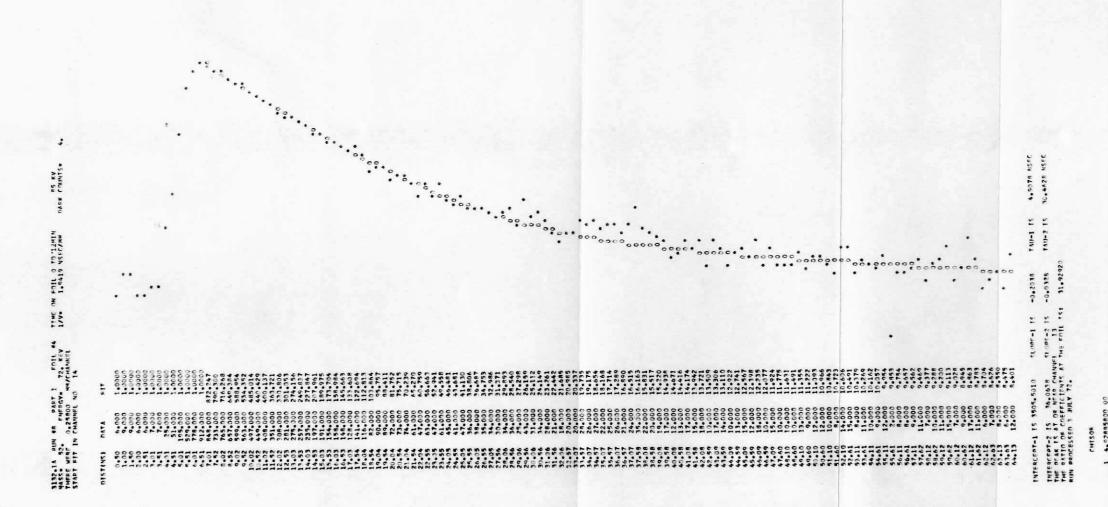
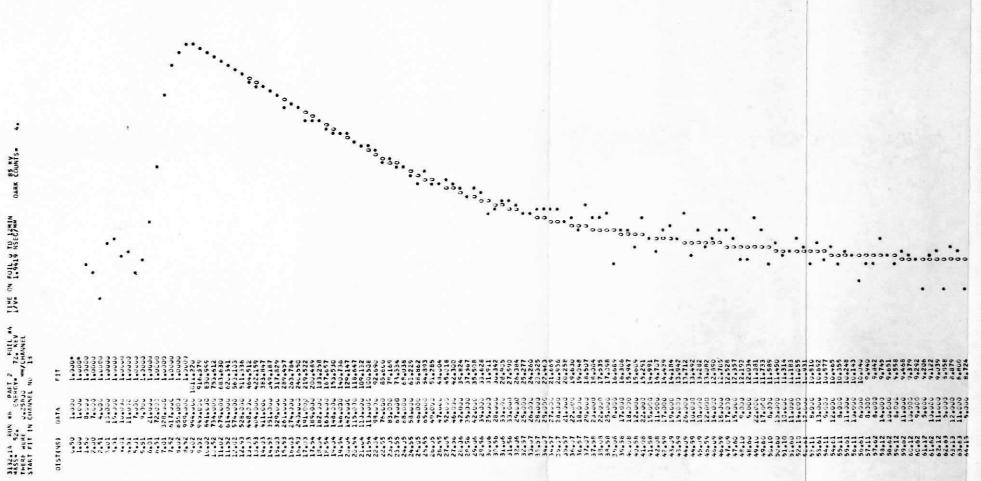
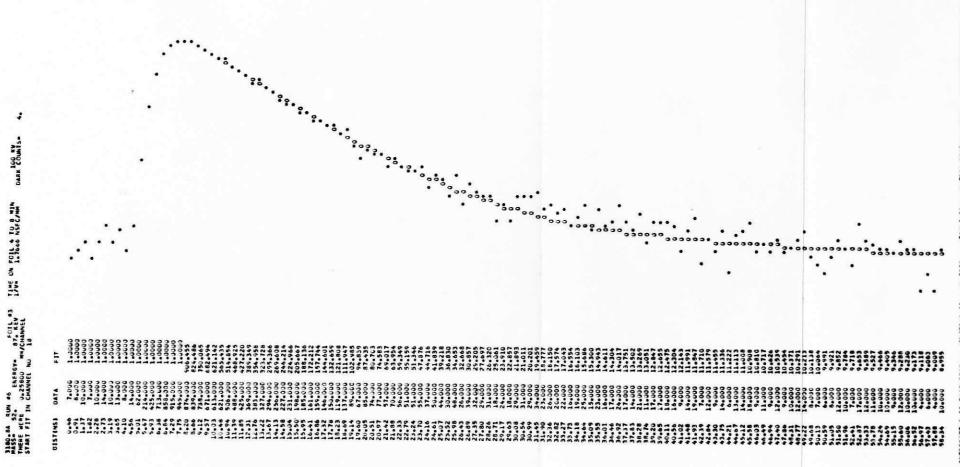


Figure XVII. C. Part two of the 85 keV lifetime fit for the line at 3132.1 A

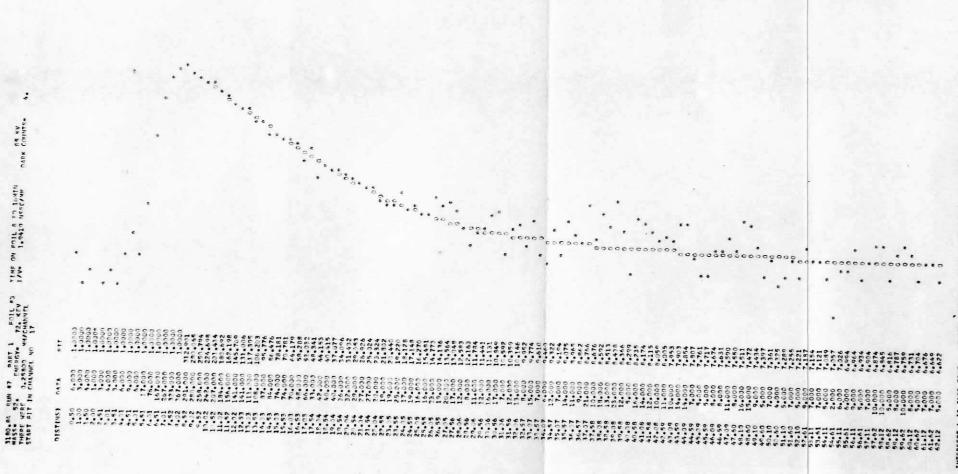


INTERCEPT-1 IS 6043-6626 SLOPE-1 IS -0.2017 TAU-1 IS 4.8979 HSEC INTERCEPT-2 IS 35-8669 SLUPF-2 IS -0.0317 TAU-2 IS 31.9890 NSEC THE PER AL ST ON WAR CHANNEL 17 THE WATHO OF COEFFICIENTS AT THE FOIL ISS 39.99943

9-87522630-01 2-9-8752630-01 3-77519690-01 4-73470740-01



INTERCEPT-1 IS 5062-2504 SLUPE-1 IS -0.2123 TAU-1 IS 4.7104 MSEC INTERCEPT-2 IS 18-5500 SLUPE-2 IS -0.0227 TAU-2 IS 44-1046 MSEC THE FAMIN OF COFFFICIENTS AT THE FOIL IS! 62-89762



INTERCOT—1 IS 2132-7667 SLORE—1 IS -0.2798 TAU-1 IS 4.3511 P. NUTBERCOT—2 IS 0.1500 SLORE—2 IS -0.0201 TAU-2 IS 40.4404 P. PER ENALS AT THE MATTER THE WATTO COMPETENTS AT THE FOLL IS 43.35765

## BEAM-FOIL LIFETIMES IN CrII

by

JAMES RICHARD METTLING

B.A., Southwestern College, 1971

AN ABSTRACT OF A MASTER'S THESIS

submitted in partial fulfillment of the

requirements for the degree

MASTER OF SCIENCE

Department of Physics

KANSAS STATE UNIVERSITY Manhattan, Kansas The lifetimes of four atomic levels of CrII have been measured at 85 and 100 keV using beam-foil methods. In preparation for the lifetime experiment the beam-foil spectrum of 85 keV chromium from 1900 to 4100 A was taken. The energy loss incured by 89 keV Ar+ passing through a 2.4  $\mu$ gm/cm<sup>2</sup> carbon foil has been measured and found to be in adequate agreement with stopping power calculations. From our lifetime measurements a preliminary solar abundance for chromium of log  $N_{\rm Cr}$  = 5.54 (where log  $N_{\rm H}$  = 12.0) has been calculated.