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A STUBY OF CEPTAIN PACTOPS IN GAS HEATING AFTECTING THE EPPICIENCY OF ALUMINUM RETTLES/

by

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B. S., College of Industrial Arts, 1916

A THESIS

submitted in partial fulfillment of the

requirements for the degree of

MASTER OF SCIENCE

KANSAS STATE AGRICULTURAL COLLEGE

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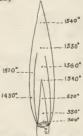
INTPODUCTION

Problam

The problem of standardisation of household equipment though recognized for more than two decades has only recently been actively engaging the attention of the home consumer. Because of its relative newness it is not surprising that the question of standardisation for cooking utensils has had as yet little attention. Any standardization of size of cooking utensil, which will be suited to the amount of food to be cooked and the burner to be used, must consider how heat can be utilized best and how waste of fuel can be eliminated. This study was made as an attempt to establish certain relations between the size of kettles of a given type, the mass of water to be heated, the kind of natural gas burners used, and the gas consumption under certain gas pressures.

Review of Literature

Previous investigations emphasize that in making any study of utensil efficiencies it is impossible to separate utensil efficiency from that of the stove, and the conditions under which it is operated. The construction and operation of the stove are vital factors in producing uniform results in kettles. For example the position of the burner relative to the utensil has a marked effect on the effielency of the utensil. The Bureau of Standards (3) found that as the distance between utensil and burner is increased there is a decrease in efficiency due to flame contact. The decrease in efficiency caused by moving the utensil from 3/4 inches to a position 1½ inches away, amounts to about 15% irrespective of velocity of gas. Should the velocity of gas be increased at the 5/4 inch position there is a lowering of efficiency. This is due to loss of heat by radiation and convection when the flame extends to the sides of the utensil. Figure 1 shows the temperatures of the various parts of a gas flame as is given by McFherson and Henderson. (10)





If the burner is adjusted so that the flame artends to the sides of the kettle, the hottest part of the flame is not in contact with the kettle. A good flame depends on the adjustment of the air shutter, the velocity of the gas and condition in the mixing chamber. All ordinary fuel gases are a mixture of naturally occurring or artificial gases. Hatural gas was used in this experiment, and by far the larger per sent is methane. The following equation given by Hefferron and Henderson (10) shows the relationship of volumes necessary for maximum consustion:

CH4 + 20g → C0g + 2H20

1 vol + 2 vol → 1 vol + 2 vol

Since air is only about 1/5 oxygon it is obvious that approximately 10 volumes of air are necessary for complete combustion of 1 volume of the natural gas. The composition of the gas determines the volume of air necessary to properly support combustion. The Bureau of Standards (3-4) sites the position, length, shape and size of the crifice, the design of the injection tube, dimension and shape of burner head and the total area of the flame ports as factors which affect the heating value of a gas flame. If these factors are controlled it is then possible to determine the conditions for the maximum efficiency of the utensil.

Landreth and Hutchinson (9) in studying the effect of

the weight of the kettle used five aluminum kettles of varying weights on an electric stove. They found that the thermal efficiency of heavy and light weight aluminum kettles with practically the same capacity and diameter was approximately the came. They also found that evaporation of water from the battle depends upon the tightness of fit of the lid rather than on the weight of the mettle. They conclude also that waterless cooking may be done in light weight kettles as well as heavy provided the lid fits well. Good (6) in making similar tests on a gas stove with four four-quart stew bettles, one of light and one heavy weight aluminum and one of heavy and one of light weight engage, found the slight difference of 3% in thermal efficiency of the light and heavy weight enamel mettles. For the aluminum mettle the heavy weight showed the greater efficiency. It is interesting to note that after the light weight aluminum hatthe was roughened with sand paper it showed an increase of 1.6% in efficiency, while the same kattle blackened on the bottom showed an increase of 5.3% in efficiency. This was probably due to reflection of heat by the brighter surface. Brighan (2) found in her study of electric stoves that an ordinary enamoled or nickled copper kettle held heat better than one of aluminum and she reports also that it absorbed heat more slowly on an open coil then for a closed coil.

She makes the statement that straight sided hottles present less surface for radiation and are more efficient, and that flat bottom pans of the proper size and type are a saving of electricity. A pan with a dismeter at the heating surface equal to or a trifle larger than the heating unit is best. The bottle should be heavy enough to provent buckling so that it will at all times fit flat on the heating unit. Brighan also found that the heavior kottle heats as quickly as the lighter one and that it holds the heat longer. She brings out the fact that close fitted covers are an economy of time and fuel.

Balderston (1) makes the statement that efficiency is not due to the thickness of material of which the utensil is made but rather to size of kettle in relation to size of the burner. She, however, does not cite proof for the statement.

Electric (8), 1017, in his work with electric stoves shows that covering a lettle while heating water gives a gain of from 10 to 25 per cent in efficiency in the utensil. In his test he used two aluminum kettles 6 inches in dismetor and 5 inches in depth. One kettle was new while the other was dissolared and dented. On the open unit the new hettle showed an increase of 2.45% in efficiency over the other while on the emclosed unit the difference in effi-

ciency was much greater, 13.8%. He considers that this difference is due to absorption of heat by the old kettle on the open unit while the bright surface of the new one reflects back some of the heat.

INVESTIGATION

Apparatus

In testing the fuel value of gas a Sargent Automatic Gas Calorimeter, No. 2756, a Sargant Wet Tost Gas Neter, a Laboratory balance, a Torsion balance, style 4420, 35 lbs. capacity graduated to .01 of a pound, a mercury barometer and Fahrenheit thermometer were used. For all the tests a gas stove of standard make, with closed manifold, was used. The burners were star shaped and the tests were made on the two at the front. The diameter of the regular burner was 4-1/4 inches and that of the giant burner was 53 inches. The giant burner used was next to the oven. The burners were 2-3/16 inches from the utenail. The mixer and cock showing the speed and orifice are illustrated in Plate No. 1. The orifice, one of the adjustable, channel type, was 1/16 inch in dissotor and fifted with a spud which regulated the volume of gas going to the mixor head. The air shutters were also adjustable. The temperature of the flame was

tested by means of an iron constantan thermocouple and a Leeds & Northup potentionstor. The bettles were of medium weight aluminum with convex sides and flat bettems. The characteristics of the bettles are showing in Table I.

TRANSFER W

	an agin sure state state state state	2.25 X	22.523 Arts	Balling, 3 2415	TAPATERS', 25% J	1007 JL 71 J	in monadil agé sau dan sér dan di	
Kot- tlo	Diam. Bottom Inches	tom	Diam. Top Inches	Depth Inches	Jante T	2000	Conner- cial Lab.Cap. Quart	Actual Cap. Quart
A	9	28.5	11-1/15	7-1/16	1.761	1.625	12	11.215
B	7	22.0	7-7/8	4-3/4	.8135	.995	6	3.834
C	5.5	17.5	7-1/4	5-5/8	.536	1.045	2	2.136

The apparatus was arranged in a corner of the laboratory where there were no windows and surrounded by a screen of pressed wood, seven feet high. Plate No. II shows the arrangement of apparatus. The time was recorded from a stop watch.

Procedure

The fuel value of gas is essential in obtaining the desired evaporation in utensil efficiencies. Checks on the calcrimeter showed that running the gas at 2 inshes pressure gave the highest fuel value. As the fuel value of the gas changes from time to time it was necessary to make the test

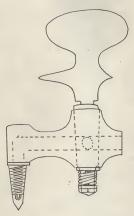
often. At least two tests were made each day and a alight variation in B. T. U. was found each time the test was made. In order to get the greatest heat from the burner the temperature of the flame was tested with the thermoccuple while the shutter governing the volume of air was being adjusted. A blue flame without yellow tips gave the greatest heat, and at the same time did not blackon the kettle. A room had to be arranged to reduce air currents to a minimum because the slightest current caused a noticeable drop in temperature. The three mottles used vere tosted with varying emounts of water. Nottle A with 3, 6, 9, 12, 15 and 18 pounds. B with 12, 5 and 6 pounds and 6 with 12 and 5 pounds. The first set of tests were made on the regular burner with lids on the lottle. A hole was made in the lid through which the thermometer, inserted in a cork, was suspended. Each mattle was weighed with lid, thermometer and cork, and the water was then weighed after the temperature of 700 was obtained. The burner had providuely been tested and lighted, and the gas rate obtained with a stop watch and wet test gas meter. The tosts were made with 5 and 7 inches of pressure. As the hettle was placed on the flame the stop watch was started and motor reading taken. The water in the mottle was heated to 209 degrees, this being the average boiling point for

this attitude. When the boiling point was reached the watch was stopped and meter out off. The water was quickly weighed to determine the amount of evaporation. To determine the rate of heating the temperature of the water was noted at intervals. Each run was repeated at least three times. The temperature of gas, temperature of room and barometric pressure were recorded for each run. Each gas volume was reduced to 60° F, and 30 inches of mercury pressure. The following formula was used for ealculating the efficiency of the hettle:

Efficiency = H x S(to - to) cu. ft. gas x BTU per cu. ft. H = water heated S = specific heat of water which is 1 to highest temperature to which the water was heated which was 200°F to beginning temperature of 70°F

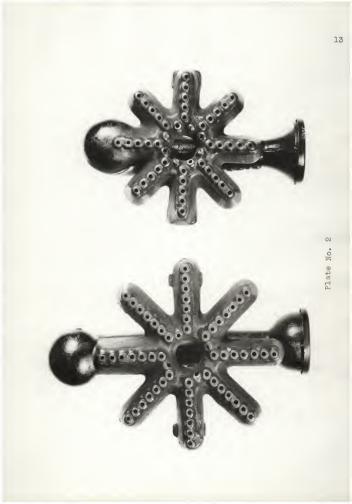
Tests with the same kettles uncovered were made in a similar way. The thermometer bulb in all cases was kept at the center of the mass of water.

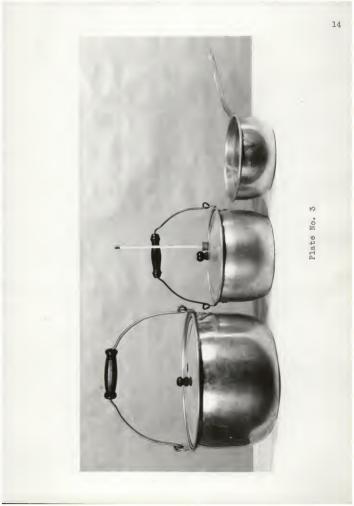
PLATE NO.I.





GAS COCK







Discussion of Results

In the tests made using the three aluminum kettles on the regular burner, it was found that the efficiency in each case was greater with three inches of gas pressure than with 6.2 inches. This was due, evidently, to the fact that more heat was utilized when the flame was shorter. With the higher pressure the flame was longer. The Bureau of Mines (7) classify this longer flame as an effective but wasteful one because so much of the total heat produced is lost to the surrounding atmosphere by being sent outward and down. Kettle C and B showed a greater rate of increase in officiency with three pounds of water when changing from 6.2 inches of gas pressure to three inches than did bettle A. Kettle C showed a greater increase in efficiency than did B. The lesser increase for A may be due principally to greater surface area which allows more evaporation. This evaporation loss may be noted by comparing data in Tables IV, VII and XV; also Tables III, IX and XIV. There was also greater surface of the hottle to radiate and reflect heat. In using three inches of gas pressure a longer time was required to heat the water but less gas was utilized. For example, in heating six pounds of water in hettle A

with the cover on, it took an average of 5.8 minutes longer to heat the water from 70° F. to 200° F. with three inches of gas pressure, and an average of .4 of a cubic foot of gas less than with 6.2 inches of gas pressure.

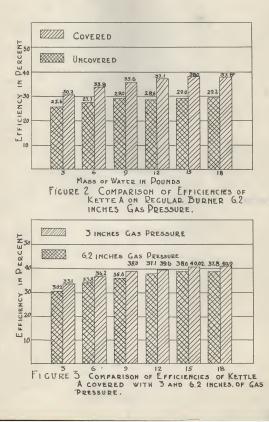
All bettles covered showed greater efficiency than uncovered. When a closed bettle is heated according to Preston (11) there is a point reached at which the molecules which have escaped from the surface of the water are reflected from the sides and from each other in such a way that many find their way back into the water lessening the rate at which evaporation takes place. When the bettle is uncovered the molecules escaping from the surface of the water are carried away by the atmosphere above and at the same time take a larger amount of heat. The Bureau of Standards (3) estimates that with each five drops of water that is taken off as steam it takes from the water enough heat to cool about a pound of water 10 F. This means that an uncovered heattle requires a longer time to reise the tamperature to the desired point.

The tests show that there is a limit to the increase in efficiency with the increase in pounds of water heated. For example the efficiency of bettle A severed increased as the water was increased up to 12 pounds. After this point it remained practically at a level. Then the bettle was un-

covered the increase in efficiency noted stopped at nine pounds.

There was an increase in efficiency with an increase in dismeter of the mettle. This may be noted especially from Table XVII for three pounds of water. The larger area of surface probably explains this fact.

All the bettles showed lowered efficiency on the giant burners due to the fact that the area of the burner was larger and a greater percentage of the heat was carried away by the surrounding air and not absorbed. The giant burner requires slightly longer time and more gas to reach a given temperature in all the tests as yet made.



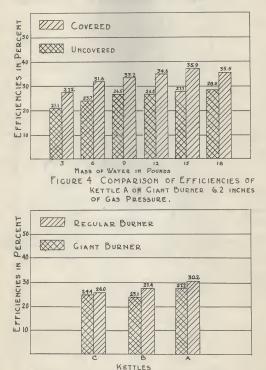
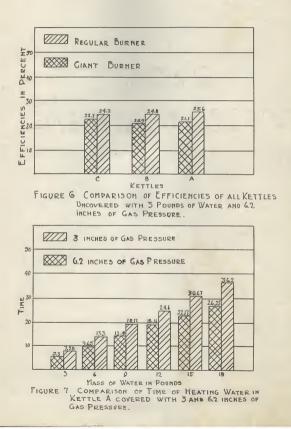


FIGURE 5 COMPARISON OF EFFICIENCIES OF ALL KETTLES COVERED WITH 3 POUNDS OF WATER AND 6.2 INCHES OF GAS PRESSURE.



HOH	Vapor	Cubic Feet	Time Minutes	Cubic Peet Per Hour	Per Cent Efficiency
5	.040	1.425	8.33	10.75	50.4
	.039	1.346	7.10	11.30	53.5
	.082	1.398	7.56	11.00	53.2
	.040	1.341	7.34	10.90	53.4
6	.042	2.350	13.50	10.38	37.1
	.041	2.421	15.11	11.20	36.3
	.047	2.378	13.74	10.35	35.1
	.051	2.341	13.11	10.70	36.1
9	.155	5.360	19.00	10.80	37.4
	.135	36.20	19.00	10.70	36.7
	.046	3.450	18.80	11.02	38.9
	.038	32.60	18.31	10.70	38.9
12	.041	4.390	24.90	10.65	39.4
	.069	4.440	24.10	11.02	40.0
	.042	4.375	25.55	10.32	40.0
	.088	4.380	24.08	10.17	39.0
15	.045	5.370	31.25	10.45	40.0
	.036	5.300	31.03	10.33	41.7
	.034	5.840	30.50	10.34	41.8
	.032	5.340	29.90	10.70	38.9
18	.027	6.130	36.00	10.35	42.4
	.039	6.399	36.30	10.57	41.2
	.038	6.270	36.15	10.60	40.1
	.036	6.620	37.00	10.70	40.0

TABLE II. KUTTLE & COVERED ON REGULAR GAS BURNER (3 Inches Gas Pressure)

TABLE III. KETTLE A UNCOVERED ON REGULAR GAS BURNER (6.2 Inches Gas Pressure)

Pounds HOH		Cubie Feet	Time	Cubic Feet	Per Cent Efficiency
3	.204	1.805	7.03	15.35	23.9
	.158	1.582	6.28	15.13	28.1
	.190	1.905	6.86	15.62	24.7

6	.812	3.23	12.33	15.80	28.7
	.565	3.42	12.73	16.10	26.4
	.261	3.12	11.57	16.20	28.5
9	.472	4.77	18.73	15.30	27.2
	.369	4.50	16.95	15.90	30.0
	.383	4.41	16.92	15.60	30.2
12	.588	6.17	24.34	15.20	28.0
	.556	6.18	23.42	15.63	29.2
	.517	6.18	23.10	16.05	28.7
15	.881	7.87	24.20	15.30	27.4
	.800	7.69	29.98	15.43	29.3
	.555	7.33	28.05	15.75	30.3
18	.986	9.23	36.15	15.32	28.2
	.903	8.99	35.60	15.15	30.1
	.761	9.09	34.90	15.60	29.3

TABLE IV. METTLE & COVERED ON NEGULAR GAS BURNER (6.2 Inches Gas Pressure)

Pounds Holl	Pounds Vapor	Cubic Feet	Time	Cubic Feet Per Hour	Por Cont Efficiency
3	.044	1.395	5.28	16.00	51.6
	.044	1.435	5.28	16.86	50.1
	.051	1.630	5.62	16.50	28.9
	.046	1.565	5.98	15.65	50.4
6.	.064 .043 .044 .048	2.660 2.615 2.618 2.618 2.541	9.73 9.59 9.75 9.85	15.80 16.35 16.25 16.38	84.0 83.4 84.2 83.7
9	+048	5.680	13.80	16.05	35.7
	•041	5.670	13.80	15.95	35.8
	•045	5.790	14.25	16.05	35.5
	•042	5.705	13.86	16.02	35.3
12	.052	4.700	18.28	16.45	36.7
	.041	4.740	18.15	15.65	37.2
	.035	4.700	17.90	15.75	37.2
	.040	4.780	18.12	15.75	37.4

15	.034	5.850	22.08	15.47	37.5
	.049	5.740	22.10	15.50	38.5
	.045	5.750	22.15	15.35	37.6
	.036	5.760	21.75	15.80	38.2
28	-020	6.960	26.02	15.70	37.7
	-045	6.850	26.10	15.75	37.8
	-030	6.250	26.23	15.70	38.5
	-030	7.020	26.25	16.32	37.3

TABLE V. KUTTLE & COVEPED ON CIANT GAS NURMER (6.2 Inches Gas Pressure)

Pounds HOH	Pounds Vapor	Cubia Foot Gas	Time	Cubic Feet Por Hour	
3	.051	1.63	6.00	16.25	27.8
	.051	1.64	5.95	16.41	25.8
	.051	1.55	5,62	16.48	27.2
6	.048	2.84	10.56	16.10	31.6
	.042	2.82	10.03	16.90	31.1
	.055	2.74	10.11	16.32	32.2
9	.041	3.89	14.85	16.13	83.8
	.039	4.01	14.45	16.85	32.2
	.052	3.95	14.26	16.25	85.7
12	.034	5.02	13.39	16.15	34.2
	.030	5.02	13.38	16.41	34.9
	.035	5.13	18.51	16.50	34.6
15	.028	6.04	22.30	16.38	36.8
	.029	3.07	22.20	16.35	36.0
	.027	6.29	22.81	16.42	35.3
13	.033	7.43	27.30	16.30	35.0
	.021	7.16	26.25	16.31	35.4
	.032	7.25	26.10	16.40	36.1

Pounds HOH	Pounds	Gubic Feet		Cubic Feet Per Hour	Per Cent Efficiency
3	.209	2.040	7.72	15.90	21.6
	.946	2.215	8.20	16.15	20.1
	.171	1.990	7.26	16.35	21.7
6	.518 .380 .335 .366	5.580 3.720 3.750 3.660	13.30 13.90 14.00 13.60	16.16 16.05 16.20	25.9 23.7 23.8 23.2
9	.574	5.030	18.50	16.35	26.7
	.558	4.940	17.95	16.50	26.5
	.340	4.770	17.50	16.28	27.5
	.465	5.000	18.98	16.10	25.3
12	.470	6.260	22.80	16.20	27.9
	.435	6.420	23.85	16.10	27.2
	.638	6.970	25.20	16.50	25.3
	.649	6.740	25.10	16.20	25.4
15	.576	7.680	27.75	16.55	28.4
	.555	7.740	28.43	16.21	28.6
	.777	8.420	30.75	16.30	26.2
28	.632	8.970	32.95	16.20	29.3
	.755	9.220	33.67	16.40	28.8
	.865	9.850	36.60	16.15	25.9

TABLE VI. KETTLE A UNCOVERED OF GIANT BURNER (6.2 Inches Gas Pressure)

TABLE VII. KETTLE & COVERED ON A REGULAR GAS BURNER (6.2 Inches Gas Pressure)

	Pounds	Cubic Feet	Time	Cubic Feet	Per Cent Rfficiency
28	.020	.935	3.40	16.00	25.4
	.020	.926	3.47	16.00	25.8
	.036	.905	3.41	16.00	24.5
	.030	.912	3.43	15.98	25.6

8	.016	1.631	5.91	16.55	26.9
	.017	1.560	6.00	15.85	28.1
	.030	1.500	6.00	16.00	27.1
	.082	1.600	6.00	15.85	27.5
8	.012	2.961	10.76	16.45	29.7
	.020	2.900	10.85	16.10	30.0
	.017	2.950	11.12	15.90	30.2
	.028	2.921	10.95	16.05	29.7

TABLE VIII. ENTILE B COVERED OF REGULAR GAS BURNER (3 Inches Gas Pressure)

Pounds	Pounds Vapor	Cubic Feet	Time	Cubic Feet Per Hour	Per Cent Efficiency
11	.017	.796	4.35	10.95	27.4
	.017	.798	4.30	11.02	27.6
	.080	.732	4.24	10.36	23.8
	.016	.758	4.10	11.10	28.4
8	.017	1.416	7.81	10.85	30.8
	.016	1.372	7.60	10.90	31.2
	.017	1.404	7.86	10.70	30.8
	-042	1.485	8.33	10.75	30.4
6	-016	8.540	14.02	10.65	34.0
	.023	2.640	15.12	10.45	33.9
	.011	2.610	14.30	10.95	33.2
	.043	2.865	15.15	10.60	32.7

TABLE IX. EFTLE B DECOVERED ON REGULAR GAS BURNER (5.5 Inches Gas Pressure)

Pounds HOH	Pounds Vapor	Cúbic Feet		Cubic Peet Per Rour	Per Cent Efficiency
12	.061 .063 .065 .052	1.051 .999 1.038 .990	5.87 5.85 5.95 3.95 3.74	16.03 15.85 15.86 15.72	21.1 22.1 21.1 22.0

8	.089	1.730	6.63	15.88	25.1
	.089	1.720	6.33	15.70	25.5
	.093	1.842	6.92	16.00	24.1
	.093	1.810	6.90	15.76	24.4
6	.134	3.260	12.38	15.80	27.2
	.094	3.390	12.60	16.05	26.1
	.136	3.170	12.17	15.80	27.8

TABLE X. EFTLE B UNCOVERED OF GIANT GAS BURNER (6.2 Inches Gas Pressure)

Pounds HOH	Pounds Vapor	Cubio Feet		Cubic Feet Per Hour	Per Cent Efficiency
13	-059	1.150	4.28	16.42	20.0
	-078	1.167	4.43	16.10	18.4
	-058	1.128	4.17	16.30	19.4
8	.091	2.140	7.78	16.23	20.9
	.096	2.155	7.88	16.40	20.8
	.095	2.190	7.617	16.58	20.9
6	.152	8.700	13.81	16.11	23.9
	.145	8.800	13.85	16.45	23.0
	.155	3.830	14.38	16.20	22.7

TABLE XI.

KETTLE S COVERED OF GIART GAS FURNER (6.2 Inches Gas Preserve)

Peands Holf	Pounds Vapor	Cubic Feet		Par Hour	Per Cent Efficiency
28	.023 .021 .021	1.015 1.013 1.115	5.88 5.84 4.02	15.90 16.01 16.88	21.2 21.5 19.0
3	.020 .016 .034	1.890 1.885 1.732	6.96 6.92 6.83	16.48 16.38 16.15	22.3 23.8

6	.012 .001 .080	3.570 3.290 3.310	16.11 16.25 16.10	25.6 25.7 26.0
	-			

TABLE HIL. ENTILE C UNCOVERED ON GIANT GAS DUMMER (0.2 Inches Gas Pressure)

Pounda	Pounda Vapor	Cubis Feet		Cubic Feet Per Hour	Per Cent
13	.068	1.145	4.17	16.45	18.6
	.065	1.135	4.28	16.15	18.7
	.055	1.025	3.93	16.05	20.7
8	.065	1.830	6.71	16.30	24.6
	.085	2.020	7.35	16.38	21.0
	.079	1.880	6.89	16.58	22.5

TABLE XIII. KETTLE C COVERED OF GIANT GAS SURPOR (6.9 Inches Gas Pressure)

Pounds ROII	Founds Vapor	Cubic Poet		Cubic Feet Per Hour	
28	.025 .017	.968 .955	3.51 3.50	16.82	20.3 22.3
3	.029 .016 .025	1.815 1.715 1.905	6.78 6.19 6195	16.10 16.36 16.30	25.7 24.7 122.7

TABLE XIV. EDTLE C DECOVERED OF REGULAR GAS EURERE (6.2 Inches Gas Pressure)

Pounda	Pounds	Oubic Foot	Time	Cubic ?eet	
24	-087	1.029	3.83	16.15	21.6
	-050	1.010	3.77	16.20	22.0
	-088	.990	3.74	15.72	22.0

3	.073	1.365	7.05	15.85	23.5
	.073	1,795	6.63	16.25	84.7
	.069	1.850	6.87	16.20	24.2

TABLE XV. KETTLE C COVERED OF RECULAR GAS SUFFIER (6.2 Inches Gas Pressure)

Pounds	Pounda	Cubic Feet	Time	Cubic Fost	Per Cent
HOH	Vapor		Winnton	Per Hour	Efficiency
28	.011	.872	3.25	16.00	24.7
	.021	.873	5.29	16.05	26.8
	.015	.918	3.48	15.85	23.6
	.018	.895	3.35	16.05	24.1
3	.022	1.717	6.48	15.80	25.1
	.014	1.642	6.16	16.05	26.8
	.012	1.675	6.29	16.00	25.7
	.015	1.640	6.18	15.92	26.3

TABLE XVI. KNITTLE C COVERED OF REJULAR GAS FURHER (3 Inches Cas Pressure)

Pounds HOE	Yaper	Cubic Post	Time Kimtes	Cubic Post Per Nour	Per Cent
13	.018 ;	0.797	4.32	11.05	27.5
	.016	0.802	4.25	11.30	27.3
	.019	0.797	4.35	10.98	27.2
	.015	0.775	4.12	11.10	27.7
3	.016	1.400	7.87	11.23	30.0
	.011	1.475	7.94	11.20	29.7
	.013	1.435	7.90	11.06	30.0
	.015	1.435	7.67	11.15	20.3

	13
	16 1b
LTLES	19 1b
ALL NO	HOR 9 1b
CIES OF	HOH 6 1b
NA I DI AN	ROH S 1b
ABLE OP	HOIL 1Å 1b
XVII. 7	Cover
TABLE	Patro
	Treemar

HOH 19 1b	37.G	40.9	80.8	35.5	28.0
10H 18 1b	38.0	80.6	89.0	35.9	27.7
HOH 18 1b	27.1	30.6	80.6	36.6	26.5
AL 6	35.6	39.0	20.1	53.8	26.8
HOII	83.6	86.8 83.4	27.8	51.6 25.3	85.7 85.8
di S	30.8 27.4 26.0	33.1 30.0 30.0	26.6 24.3 24.1	27.2 23.1 24.4	21.1 20.9 22.7
14 1b	85°.8	88.1 87.4	21.0 21.0	80°6	19.3
Cover	888	888	220	888	110
Purner	Regular Regular	Regular Regular	Regular Regular	Glant Olant Diant	Stant Stant Stant
Trebas	000 000 0000	000	6.8 Regular 6.8 Regular 6.8 Regular	000	0000 0000
Kettle.	480	<80	400	480	480

Summary of Remaits

1. Although from 20 to 25% more time was required when the regular burner was used, the efficiencies were greater for three inches of gas pressure than for 6.2 inches. With the lower pressure the efficiency of kettle A was about 7% greater while the efficiencies of kettles B and C were from 11 to 15% higher.

2. The covered mettles on the regular burner showed greater efficiencies than the uncovered with both the 3 and 6.2 inches of gas pressure. The efficiencies of mettle A increased from 20 to 30% when the cover was used. For mettles B and C covered the increase ranged from 3 to 14%.

3. The efficiency of mattle A <u>uncovered</u> on both burners with 6.2 inches of gas pressure remains about the same for mine pounds or more of water.

 Estile A <u>covered</u> showed only a slight increase in efficiencies for masses of water of 12 or more pounds when three inches of gas pressure were used.

5. With three pounds of water and 6.2 inches of gas pressure the efficiency of the mottle with largest area was greater than that of the two mettles with the lesser areas except when the largest mettle was used uncovered with the giant hurner.

6. Estiles B and C containing 12 pounds of water showed little difference in efficiencies under the same conditions.

7. The covered kettles showed lower efficiencies on the giant burner with 6.2 inches of gas pressure. For bottle 4 on the regular burner the efficiency was about 4% greater with the larger amounts of water and 15 to 20% greater with the smaller amounts. For the mmaller bettles the increase ranged from 8 to 20% when the regular burner was used. The giant burner utilised more gas throughout the test. There was a slight saving of time when the giant burner was used with the smaller bettles.

CONCLUSIONS

1. On the regular burner three inches of gas pressure. gave greator officiencies than 5.2 inches but required longor time.

2. In contracting the regular burner with the giant burner greater efficiencies in all cases were found with the regular burner when 6.2 inches of gas procesure were used.

5. The covered mettles in all cases showed greater efficiencies then the uncovered enes.

4. The kottle having the largest dismeter was more of-

ficient than the smaller once except when used uncovered with the giant burner.

ACKNOWL SPORENTS

Acknowledgmont is due to Hiss Mary Fidelia Taylor who directed the research of the thesis and to Dean Margaret Justin who read the thesis and offered helpful criticism. I wish, also, to make grateful acknowledgment to Miss Stella Mend Marris for her valuable advice and criticisms while the thesis was being written.

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