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1 Thermo-mechanical Extrusion Pretreatment for Conversion of Soybean Hulls to  
2 Fermentable Sugars

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16 **Abstract**

17 Thermo-mechanical extrusion pretreatment for lignocellulosic biomass was  
18 investigated using soybean hulls as the substrate. The enzyme cocktail used to  
19 hydrolyze pretreated soybean hulls to fermentable sugars was optimized using response  
20 surface methodology (RSM). Structural changes in substrate and sugar yields from  
21 thermo-mechanical processing were compared with two traditional pretreatment  
22 methods that utilized dilute acid (1% sulfuric acid) and alkali (1% sodium hydroxide).  
23 Extrusion processing parameters (barrel temperature, in-barrel moisture, screw speed)  
24 and processing aids (starch, ethylene glycol) were studied with respect to reducing sugar  
25 and glucose yields. The conditions resulting in highest cellulose to glucose conversion  
26 (95%) were screw speed 350 rpm, maximum barrel temperature 80°C and in-barrel  
27 moisture content 40% wb. Compared with untreated soybean hulls, glucose yield from  
28 enzymatic hydrolysis of soybean hulls increased by 69.6, 128.7 and 132.2%,  
29 respectively, when pretreated with dilute acid, alkali and extrusion.

30

31 **Keywords:** soybean hulls; lignocellulosic; ethanol; pretreatment; extrusion

32 1. Introduction

33 The National Biofuels Action Plan released in October 2008 states that  
34 expanding annual biofuels production to 36 billion gallons by 2022 would be a key  
35 component in America's movement towards clean, affordable and secure energy  
36 sources. Corn ethanol, which is categorized as a first generation biofuels, has grown  
37 steadily over the last few years to 12.1 billion gallons in 2010 (Service, 2010). However,  
38 corn ethanol alone cannot meet the increasing demand for biofuels because of  
39 sustainability issues (Perlack et al., 2005). The Renewable Fuel Standard 2 (RFS2),  
40 which was published in March 2010, introduced a 15 billion gallon cap for corn ethanol  
41 by 2015 and set the goal of 21 billion gallons for advanced biofuels by 2022. The latter  
42 include cellulosic ethanol, biodiesel and other nonconventional biofuels.

43 Lignocellulosic biomass is abundantly available, and has potential for better  
44 energy balance than corn ethanol. Additionally, cellulosic ethanol provides the benefit  
45 of reducing greenhouse gas emissions by up to 86% (Wang et al., 2007). Potential  
46 feedstock includes perennial energy crops such as switch grass, forest residues such as  
47 wood, crop residues such as sugarcane bagasse, corn cobs, corn stover and wheat straw,  
48 and agricultural co-products such as soybean hulls and wheat bran. In the USA 10.8  
49 billion pounds of soybean hulls are produced every year, as a result of processing 1.8  
50 billion bushels of soybeans for oil, protein and flour (Corredor et al., 2008). This co-  
51 product is used as a low-value protein/ fiber supplement in ruminant animal feed, but its  
52 demand is expected to reduce due to availability of distiller's dry grains with solubles  
53 (DDGS) as a substitute. Soybean hulls contain 46-51% cellulose, 16-18% hemicellulose  
54 and 1.4-2% lignin (Corredor et al., 2008; Blasi et al., 2000), which makes them an  
55 attractive source of fermentable sugars for ethanol production. In addition to their high

56 cellulose content, soybean hulls do not require as extensive a grinding process prior to  
57 pretreatment as some other feedstocks.

58 Cellulosic ethanol production is different from corn ethanol, due to the  
59 ‘recalcitrance’ of the substrate. An additional pretreatment step is required because the  
60 lignocellulosic structure is very complex, highly crystalline, and resistant to enzymatic  
61 degradation in its native state. Despite the ongoing research on pretreatment  
62 technologies and other aspects of cellulosic ethanol production the current production  
63 costs are still too high to compete in the market place (Galbe et al., 2007). In 2010,  
64 production of cellulosic ethanol in the USA, all by demonstration facilities, was  
65 expected to total 25.5 million gallons (Service, 2010). This is far less than original  
66 expectations, due to the economic slump, uncertainty among policymakers, an  
67 oversupply of first generation biofuels, transportation logistics for raw materials, and  
68 technological gaps that make a cellulosic ethanol plant unviable. To overcome at least  
69 part of these obstacles, transformational scientific breakthroughs are urgently needed in  
70 pretreatment procedures, enzyme systems for conversion of pretreated substrate to  
71 fermentable sugars and efficient microorganisms to ferment mixed sugars to ethanol  
72 (Saha and Bothast, 1999).

73 In order to break down the structural integrity of lignocellulosic biomass and  
74 enhance enzymatic action on cellulose, various chemical, mechanical, thermo-chemical  
75 and biochemical pretreatment have been investigated, including acid hydrolysis, alkali  
76 hydrolysis, the organosolv process, steam explosion, ammonia fiber explosion (AFEX),  
77 pyrolysis, hot water treatment and microorganism treatment (Corredor et al., 2008;  
78 Galbe et al., 2007; Galbe and Zacchi, 2007; Sun and Cheng, 2002). There have been  
79 some successes in terms of increased ethanol yield, however no single method has yet  
80 been found suitable for commercial application (Saha and Cotta, 2007). The challenge

81 lies in increasing the efficiency of cellulose conversion to the extent that the overall  
82 ethanol production process becomes cost-effective.

83 This article primarily focuses on pretreatment methods, and to a certain degree  
84 on optimum enzyme systems. Most of the above mentioned pretreatment technologies  
85 have inefficiencies related to batch processing, small capacity and low solids loading. A  
86 recent publication by our laboratory (Lamsal et al., 2010) described the application of a  
87 promising technology, extrusion processing, which can provide a unique and  
88 continuously stirred thermo-chemical reactor environment in combination with  
89 mechanical energy or shear. The shear forces applied by the extrusion screw serve to  
90 continuously remove the softened surface regions of the substrate and expose the  
91 interior to chemical and/or thermal action, thus improving the overall rate of cellulose  
92 conversion. The high throughputs associated with extrusion further increase its  
93 attraction as an effective pretreatment technology. In this study soybean hulls were used  
94 as the model lignocellulosic substrate, with the objectives of development of a protocol  
95 for thermo-mechanical extrusion pretreatment, optimization of the enzyme cocktail for  
96 downstream hydrolysis to sugars, and comparison of conversion efficiency of extrusion  
97 with two traditional pretreatments, acid hydrolysis and alkali hydrolysis.

98

## 99 **2. Materials and Methods**

### 100 *2.1 Lignocellulosic biomass*

101 Soybean hulls (Archer Daniels Midland Company, Salina, KS) were ground to a  
102 particle size less than 1041 $\mu$ m by using an experimental laboratory roller mill ( Ross  
103 Roller Mill, Ross Machine & Mill Supply Inc., Oklahoma City, OK). All experiments  
104 described in this study utilized ground soybean hulls as the substrate. The  
105 lignocellulosic composition of soybean hulls was determined using the ANKOM Fiber

106 Analyzer (ANKOM Technology, NY), and starch content was determined using the  
107 standard glucoamylase method (920.40; AOAC 2010).

### 108 *2.2. Dilute acid and alkali pretreatment*

109 Ground soybean hulls were subjected to two traditional pretreatment methods,  
110 dilute acid hydrolysis and alkali hydrolysis, which were adapted from previous studies  
111 (Corredor et al., 2008; Saha and Bothast, 1999). The substrate (10%, w/w) was added to  
112 1% (w/w) sulfuric acid or 1% (w/w) sodium hydroxide solutions (both chemicals were  
113 obtained from Sigma Aldrich, St. Louis, MO), and autoclaved (SS-325E; Tomy Tech  
114 USA, Inc., Fremont, CA) at a temperature of 121°C for 30 min. The remaining solids  
115 were neutralized and washed three times with 500 ml of deionized water at 85°C. This  
116 was followed by 10 min of centrifugation (10,000×g) and subsequent drying in an air  
117 oven at 45°C for 24 h.

### 118 *2.3. Enzyme cocktail optimization*

119 A combination of cellulase (Celluclast 1.5L),  $\beta$ -glucosidase (Novozyme 188)  
120 and a cell-wall degrading enzyme complex (Viscozyme® L) were used for enzymatic  
121 saccharification of pretreated soybean hull. All enzymes were obtained from Novozyme,  
122 Franklinton, NC. Cellulase was used to break down cellulose into glucose, cellobiose  
123 and higher glucose polymers. The enzyme activity of Celluclast 1.5L was 80 FPU (filter  
124 paper unit)/mL with optimal pH range 4.5-6.0 and optimal temperature range of 50-  
125 60°C.  $\beta$ -glucosidase with enzyme activity of 322 CBU (cellobiase unit)/mL was used to  
126 hydrolyze cellobiose, which inhibits the initial stage of cellulose hydrolysis.  
127 Viscozyme® L contained a wide range carbohydrases, including arabinase, cellulase,  $\beta$ -  
128 glucanase, hemicellulose and xylanase, which act on branched pectin-liked substances  
129 found in plant cell walls. Its activity was 134 FBG (fungal beta-glucanase unit)/mL.  
130 Response surface methodology was used to optimize the amount of each enzyme for

131 maximum reducing sugar yield, using soybean hulls pretreated with acid hydrolysis as  
132 the substrate . The experimental design (Table 1) had fifteen combinations with 3 levels  
133 (-1, 0, 1) of each enzyme and 3 replicates at the center point (Box and Draper, 1987).  
134 Results were analyzed using SAS software (v.9.2, SAS Institute, Cary, NC) at  $p < 0.05$ .  
135 Three dimensional plots of reducing sugar yield for various enzyme dosages were  
136 generated, model coefficients were obtained using the response surface regression  
137 (RSREG) procedure, and optimum enzyme dosage determined.

#### 138 *2.4. Extrusion pretreatment - high starch*

139 Thermo-mechanical or extrusion pretreatment of soybean hulls, the primary  
140 focus of this study, was conducted using a laboratory-scale twin-screw extruder (Micro-  
141 18, American Leistritz, Somerville, NJ) with a six-head configuration, screw diameter  
142 of 18 mm, L/D ratio of 30:1 and 2.4 mm circular die opening. Prior to extrusion,  
143 soybean hulls were mixed with 10 and 20% corn starch (Archer Daniels Midland  
144 Company, Salina, KS) as a processing aid to facilitate the flow of soybean hulls. The  
145 soybean hull and corn starch blends were hydrated using a Hobart mixer (The Hobart  
146 Mfg. Co., Troy, OH) and equilibrated for 24 h at room temperature before extrusion in  
147 order to obtain 20 and 25% (wet basis or wb) moisture. This is also referred to as the  
148 process moisture or in-barrel moisture. The extruder screw configuration and barrel  
149 temperature profile are shown in Fig. 1. Three temperature profiles with maximum  
150 barrel temperatures of 80, 110 and 140°C (at the discharge end) were used. Extruder  
151 screw speed was fixed at 420 rpm. The feeder screw speed rate was set at 275 rpm,  
152 which resulted in a feed rate ranging from 1.4 to 2.0 kg/h depending on the moisture and  
153 starch content of the feed. Extruded soybean hull pellets were dried in a convection  
154 oven at 45°C for 24 h. The dried pellets, having a moisture content of 9-10% (wb), were  
155 stored at room temperature. Unlike acid or alkali hydrolyzed soybean hulls, the extruded



156 soybean hulls did not require neutralization or washing steps prior to enzymatic  
157 hydrolysis.

158 Specific mechanical energy (SME) input during the extrusion process was  
159 computed as follows:

$$160 \quad SME \text{ (kJ / kg)} = \frac{\frac{(\tau - \tau_0)}{100} \times \left(\frac{N}{N_r}\right) \times P_r}{\dot{m}} \quad (1)$$

161 where,  $\tau$  is the % torque;  $\tau_0$  is the no load % torque;  $N$  is the screw speed;  $N_r$  is the  
162 rated screw speed (500 rpm);  $P_r$  is the rated motor power (2.2 kW); and  $\dot{m}$  is the mass  
163 flow rate or feed rate (kg/s).

#### 164 *2.5. Extrusion pretreatment – low starch*

165 Another set of thermo-mechanical pretreatments was carried out with low levels  
166 of starch addition (0 and 5%). Results from the above experiment and also a previous  
167 study (Lamsal et al., 2010) indicated that in the case of soybean hulls reduced amount or  
168 absence of the processing aid led to surging, burning of the substrate and even blocking  
169 of the die during extrusion. To avoid these problems, higher levels of in-barrel moisture  
170 were required. Soybean hulls with 5% starch were hydrated to moisture levels of 20, 25,  
171 30, 35 and 40% (wb), whereas the moisture levels for soybean hull without starch were  
172 adjusted to 40, 45 and 50% (wb). All other process parameters were same as in the  
173 previous experiment. At the same feeder screw speed of 275 rpm, the resultant feed rate  
174 however was much lower, ranging from 0.48 to 0.84 kg/h, because of the higher  
175 moisture and lower starch level. Only one barrel temperature profile was evaluated  
176 (maximum 80°C). In an additional experiment, thermo-mechanical pretreatment of  
177 soybean hulls without starch was evaluated at 40% (wb) in-barrel moisture content and  
178 three different extruder screw speeds (280, 350 and 420 rpm). All other process  
179 conditions were same as above.

180 2.6. *Extrusion pretreatment - ethylene glycol*

181 Ethylene glycol has been used for fibrillation of lignocellulosic materials such as  
182 wood (Lee et al., 2009). One set of experiments was conducted to study its effectiveness  
183 with soybean hulls in combination with extrusion pretreatment. Solutions were prepared  
184 at ratios of ethylene glycol (Sigma-Aldrich, St. Louis, MO) to water of 1:1, 1:3 and 1:9.  
185 Pure water and ethylene glycol were used as controls. Soybean hulls were blended with  
186 solvent at a ratio of 2:1. The same extrusion processing conditions were used as  
187 described above (Section 2.5)

188 2.7. *Enzymatic hydrolysis and sugar analysis*

189 Pretreated soybean hulls (10% w/w) were hydrolyzed by using the optimized  
190 enzyme cocktail dosage in 0.05M sodium acetate buffer at pH 5, and incubation at 50°C  
191 for 72 h. Soybean hulls without any pretreatment were also subjected to enzymatic  
192 saccharification as a control. Total reducing sugar yield was determined by using the  
193 3,5-dinitrosalicylic acid (DNS) reagent method (Miller, 1959), while glucose  
194 concentration was analyzed by high performance liquid chromatography (Agilent 1100;  
195 Agilent Technologies; Santa Clara, CA) using a Phenomenex Rezex ROA organic acid  
196 column (130x7.8 mm, H+ (8%); Phenomenex, Torrance, CA). The mobile phase was  
197 0.005N sulfuric acid at 60°C with flow rate of 0.6 ml/min.

198 Reducing sugar and glucose yields were expressed in terms of g/g extruded  
199 pellet. The degree of conversion to reducing sugars ( $C_{RS}\%$ ) was calculated based on  
200 total cellulose and hemicellulose content in soybean hulls and added starch, as follows.

201 
$$C_{RS}\% = \frac{Y_{RS}}{(C + H) \times 1.1} \times 100 \quad (2)$$

202 where, C = amount of cellulose in 1 g of pellet (g/g), H = hemicellulose in 1 g of pellet  
203 (g/g), 1.1 = conversion factor for polymer to monomer sugar,  $Y_{RS}$  = reducing sugar

204 yield (g/g). The degree of conversion from cellulose to glucose conversion ( $C_{CG}\%$ ) was  
205 determined as the ratio of the glucose obtained to the theoretical yield based on the  
206 amount of cellulose in soybean hulls.

$$207 \quad C_{CG} \% = \frac{Y_G}{C \times 1.1} \times 100 \quad (3)$$

208 where  $Y_G$  = glucose yield (g/g) and  $C$  = amount of cellulose in 1 g of pellet. Both  
209 reducing sugar and glucose yields were corrected for the contribution of starch,  
210 assuming 100% conversion of the latter.

### 211 *2.8. Structural characteristic by using X-ray diffraction*

212 X-ray diffraction (XRD) patterns of the lignocellulosic substrate before and after  
213 pretreatment were analyzed using an X-ray diffractometer (D-8 Advance, Bruker AXS,  
214 Karlsruhe, Germany) set at 40 kV and 40 mA. Samples were scanned in the range of  
215 diffraction angle  $2\theta = 10-40^\circ$  with a step size of  $0.05^\circ$  and a scan speed of  $4^\circ/\text{min}$ . The  
216 crystallinity index (CrI) was the percentage of crystalline material in the biomass and  
217 was used to compare the effects of various treatments on the structural modification of  
218 lignocellulosic biomass. It was defined as follows.

$$219 \quad CrI = \frac{(I_{002} - I_{am})}{I_{002}} \times 100 \quad (4)$$

220 where  $I_{002}$  is the diffraction intensity at  $2\theta = 22.6^\circ$ , which represents both the crystalline  
221 and amorphous regions, and  $I_{am}$  is the diffraction intensity at  $2\theta = 18.7^\circ$ , representing  
222 the amorphous material (Yoshida et al., 2008; Zhao et al., 2008).

223

## 224 **3. Results and Discussion**

### 225 *3.1. Lignocellulosic structure of soybean hulls*

226 The soybean hulls contained 35.35, 17.21 and 2.33% (dry basis or db) of  
227 cellulose, hemicellulose and lignin, respectively (Table 2), and 0.83% (db) starch. The

228 remaining portion is reported to comprise of protein, fat, pectin, sugars and ash  
229 (Corredor et al., 2009; Schirmer-Michel et al., 2008). Cellulose is mainly comprised of  
230 glucose monomers, while hemicellulose is composed of mannose and xylose (Huisman  
231 et al., 1998; Stombaugh et al., 2000). Typically, cellulose exists as microfibrils, which  
232 are sheathed with hemicellulose and aligned in the direction of cell walls. Lignin forms  
233 the glue that holds the fibers together, and has a complex, cross linked structure whose  
234 building units include monomers such as guaiacyl and syringyl. Pectin performs a  
235 similar function and mainly consists of arabinose, galactose, rhamnose, fructose and  
236 uronic acid. As cellulose is the only source of glucose, the theoretical yield of latter was  
237 calculated as 0.39 g/ g soybean hull based on cellulose composition, and by multiplying  
238 by 1.1 to account for the water of hydrolysis.

### 239 3.2. *Enzyme cocktail optimization*

240 Response surface analysis of enzymatic saccharification data resulted in a  
241 statistical model for describing reducing sugar yield from dilute acid pretreated soybean  
242 hulls. The model ( $R^2 = 0.96$ ) equation is described below.

$$243 \quad Y_{RS} = 5.18 + 699.51X + 75.72Y + 52.02Z - 3623.71X^2 - 365.15Y^2 - 375.27Z^2 \\ 244 \quad - 824.76XY - 1.69XZ + 463.82YZ \quad (5)$$

245 where  $Y_{RS}$  = reducing sugar yield (g/g),  $X$  = cellulase (mL/g cellulose),  $Y$  =  $\beta$ -  
246 glucosidase (mL/g cellulose), and  $Z$  = cell wall degrading enzyme complex (mL/g  
247 cellulose).

248 Based on the above, maximum reducing sugar release was predicted at 0.0873,  
249 0.0806, and 0.1189 ml/g cellulose for cellulase (Celluclast 1.5L),  $\beta$ -glucosidase  
250 (Novozyme 188) and cell-wall degrading enzyme (Viscozyme), respectively. This  
251 combination of enzymes was used for all subsequent enzymatic saccharification  
252 experiments in this study. Cellulose concentration and its second order term, i.e.  $X$  and

253  $X^2$ , had the greatest contribution to the statistical model with  $p=0.0002$  and  $0.0037$ ,  
254 respectively. Cellulase acts directly on cellulose and degrades it to lower molecular  
255 weight glucans and glucose. Previous studies on enzymatic saccharification of  
256 lignocellulosic substrates have used cellulase loadings in the wide range of 7 to 33 FPU  
257 /g substrate (Sun and Cheng, 2002). In this study, the optimum amount of 0.0873 ml/g  
258 cellulose corresponded to 6.74 FPU/g cellulose, which is a lower cellulose dosage than  
259 that used in most previous studies. The addition of  $\beta$ -glucosidase achieved better  
260 saccharification by hydrolyzing cellobiose, which as mentioned before was an  
261 intermediate product of cellulose hydrolysis and inhibited cellulase activity. Cellulase  
262 has been supplemented with  $\beta$ -glucosidase in many other studies, and the application of  
263 Novozyme 188 was recommended by the manufacturer at the ratio of 1:0.2 for  
264 Celluclast 1.5L to Novozyme 188 for initial industrial trials (Note, 1990). Substantial  
265 amount of hemicellulose was presumed to be dissolved and removed by the dilute acid  
266 pretreatment. However, the presence of cell wall degrading enzyme complex did  
267 contribute to the efficiency of enzymatic saccharification. The primary mechanism was  
268 hydrolysis of the residual hemicellulose and increase in cellulase accessibility to the  
269 cellulose (Fan et al., 1987). Cell wall degrading enzymes have been reported to act on  
270 even untreated soybean hulls, resulting in more than 97% increase in sugar yield during  
271 enzymatic hydrolysis with cellulase (Corredor et al., 2009).

272         It was clear from optimization results that enzyme concentration beyond a  
273 certain level led to a decrease in the amount of reducing sugar liberated. Faster sugar  
274 production with an enzyme overdose might lead to inhibition of the hydrolysis process  
275 (Drissen et al., 2009). Enzyme dosage also significantly affects the cost of the overall  
276 ethanol production process, which is another reason for optimizing the dosage.

277 *3.3. Crystallinity of pretreated soybean hull*

278           The crystallinity index (CrI) of soybean hulls prior and subsequent to various  
279 pretreatments is shown in Fig. 2. Increase in cellulose crystallinity after pretreatment of  
280 lignocellulosic substrates using sulfuric acid, sodium chlorite, peracetic acid or other  
281 chemicals has been observed previously (Weimer et al., 1995; Corredor et al., 2008;  
282 Yoshida et al., 2008; Zhao et al., 2008). In the current study also the average  
283 crystallinity of the substrate increased due to chemical pretreatment with acid (AH) or  
284 alkali (ALH). The two processes led to removal of hemicellulose and delignification,  
285 respectively, and thus increased the concentration of cellulose and CrI. This was  
286 confirmed by the lignocellulosic composition data (Table 2). After sulfuric acid and  
287 sodium hydroxide pretreatments, the concentration of cellulose increased by 75 and  
288 89%, respectively, due to removal of other components such as hemicellulose from the  
289 substrate. This would lead to greater exposure of cellulose to enzymatic action and more  
290 efficient saccharification. It was clear that the mechanism by which acid and alkali  
291 pretreatments render the substrate more susceptible to enzymatic hydrolysis does not  
292 rely on decreased crystallinity. Lee et al. (2009) suggested that opening of the cell wall  
293 structure at the microscopic scale due to some pretreatment would be sufficient for  
294 enzymatic saccharification regardless of cellulose crystallinity. Yoshida (2008) also  
295 found that delignification increased the rate of enzymatic hydrolysis of cellulose and  
296 hemicellulose, although delignified biomass showed higher crystallinity. They  
297 concluded that lignin is the most significant factor interfering with the enzymatic  
298 hydrolysis of lignocellulosic biomass. However, lowering of cellulose crystallinity and  
299 increase in substrate surface area due to mechanical pretreatment processes such as ball  
300 milling (Ouajai and Shanks, 2006; Yoshida et al., 2008) can also lead to improved  
301 enzymatic saccharification.

302 In comparison, thermo-mechanical extrusion pretreatment appeared to have a unique  
303 impact on substrate morphology. Cellulose crystallinity increased by 82% (Fig. 2), even  
304 though there was no significant change in lignocellulosic composition (Table 2). This  
305 indicated crystallization of the amorphous phase of cellulose during thermo-mechanical  
306 processing. Other researchers have observed recrystallization of cellulose in the  
307 presence of moisture and heat, during or subsequent to pretreatment processes such as  
308 steam explosion and ball milling (Tanahashi et al., 1983; Bertran and Dale, 1985;  
309 Ouajai and Shanks, 2006). Different degrees of recrystallization of ball milled hemp  
310 cellulose, resulting from various water amounts, drying rates and agitation speed, were  
311 observed by Ouajai and Shanks (2006). Bertran and Dale (1985) also reported that  
312 amorphous cellulose from cotton linter and wood was recrystallized in aqueous media  
313 subsequent to ball milling, and concluded that water promoted recrystallization.  
314 Tanahashi et al. (1983) found that the crystallinity of cellulose in wood increased after  
315 high pressure steam explosion due to the transformation of cellulose from amorphous to  
316 crystalline phase. Similarly, extrusion is a high pressure and temperature process, and it  
317 is reasonable to expect inducement of cellulose crystallization in the presence of  
318 moisture as supported by data.

#### 319 *3.4. Effect of on extrusion pretreatment on saccharification*

320 Extrusion is widely used in in the food and plastics industry and the  
321 corresponding process dynamics have been extensively studied and understood. In  
322 contrast, extrusion of lignocellulosic material is far more complex and challenging,  
323 primarily because the material does not melt even at high temperatures and  
324 consequently shows poor flow properties inside the extruder barrel and die (Prat et al.,  
325 1999). This was observed in the current study also as described below.

326 *3.4.1. Barrel temperature*

327 Extrusion experiments were conducted at maximum barrel temperatures of 80,  
328 110 and 140°C. At higher barrel temperature (110 and 140°C), processing conditions  
329 were unstable. For example, large fluctuations in motor torque and die pressure were  
330 observed, the flow occurred in surges, and burning of soybean hulls inside the barrel  
331 and die led to frequent blockage and cessation of screw rotation. The little amount of  
332 soybean hulls that could be extruded at 110 and 140°C appeared to be darker than those  
333 extruded at 80°C, which again pointed towards burning. Reducing sugar yields are  
334 summarized in Fig. 3. The contribution of starch, assuming complete hydrolysis, was  
335 expressed in darker shade for each treatment.  $Y_{RS}$  ranged from 0.46-0.61 g/g pellet,  
336 which corresponded to conversion percentage ( $C_{RS}\%$ ) of 79-104% after subtracting the  
337 contribution of starch. Greater than 100% conversion was probably because of  
338 naturally present sugars in untreated soybean hull. Significant trends in  $Y_{RS}$  were  
339 however were not observed with respect to the barrel temperature. As soybean hulls  
340 could be processed with relative ease at 80°C without any impact on sugar yield, that  
341 barrel temperature was considered to be optimum and utilized in all subsequent  
342 experiments and analyses.

343 *3.4.2. Starch and in-barrel moisture*

344 As mentioned earlier, starch was added to soybean hulls at levels of 5, 10 and 20%  
345 as a processing aid that facilitated flow during extrusion. Lignocellulosic materials such  
346 as soybean hulls are otherwise tough to extrude especially at low in-barrel moisture  
347 (<35% wb), as determined previously by our research group (Lamsal et al., 2010).  
348 Materials like wheat bran are an exception, as they naturally contain as much as 19%  
349 starch, which allows extrusion at in-barrel moistures of 20-25% wb (Lamsal et al., 2008;



350 Lamsal et al., 2010). The starch is gelatinized during extrusion and forms a viscous melt,  
351 which assists in pushing the lignocellulosic material through the extruder.

352 Glucose yields ( $Y_G$ ) from experiments at barrel temperature of 80°C and  
353 constant screw speed of 420 rpm are summarized in Fig. 4. As before the contribution  
354 of starch, assuming complete hydrolysis, was expressed in darker shade. Without taking  
355 the starch into account, the conversion percentage was as high as 130%. Thus it was  
356 obvious that glucose and higher molecular weight dextrans were produced from  
357 hydrolysis of starch during thermo-mechanical treatment and subsequent enzymatic  
358 saccharification. The latter effect was probably due to side activities of Viscozyme,  
359 Novozyme 188 and Celluclast 1.5L, which is typical of enzymes produced from natural  
360 strains of fungi.

361 Overall glucose yields ( $Y_G$ ) ranged from 0.32-0.38 g/g pellet, and did not exhibit  
362 any trends with respect to starch level. It should be noted that soybean hulls without any  
363 starch could also be extruded, provided in-barrel moisture was high enough (40-50%  
364 wb), resulting in  $Y_G$  comparable to treatments with 5-20% starch addition. After  
365 subtracting the contribution of starch, glucose yields ranged from 0.15-0.35 g/g and the  
366 corresponding cellulose to glucose conversion percentage ( $C_{CG}\%$ ) ranged from 51-90%.  
367  $C_{CG}\%$  had a clear decreasing trend with respect to starch level. For example, at 20% in-  
368 barrel moisture  $C_{CG}\%$  decreased from 78% to 51% as starch increased from 10 to 20%.  
369 The corresponding decline was from 63% to 51% at 25% in-barrel moisture. Such a  
370 trend was also observed for reducing sugar conversion percentage ( $C_{RS}\%$ ) (Fig. 3).  
371 Specific mechanical energy (SME) ranged from 738-1905 kJ/kg, and increased starch  
372 level resulted in higher SME due to greater viscosity development. However, it was  
373 clear that both  $Y_G$  and  $C_{CG}\%$  were not proportional to SME. One reason could be that

374 mechanical energy was distributed unevenly between soybean hulls and starch, in favor  
375 of the latter.

376 The above reasoning also appeared to be valid for the relationship between in-  
377 barrel moisture, SME and glucose yield at high starch levels (10-20%). Higher motor  
378 torque was obtained as in-barrel moisture increased from 20 to 25% due to greater  
379 viscosity development, which in turn led to increase in SME. However,  $Y_G$  and  $C_{CG}\%$   
380 either decreased or remained the same. On the other hand at lower starch levels (0-5%),  
381 torque exhibited the near universal trend of increasing with lower in-barrel moisture. In  
382 the absence of adequate starch, mechanical energy was directed towards soybean hulls  
383 and exposed the interior of the substrate to thermal energy and subsequent enzymatic  
384 saccharification, thus leading to greater  $Y_G$  and  $C_{CG}\%$ . This demonstrated the primary  
385 mechanism involved in thermo-mechanical pretreatment of lignocellulosic substrates,  
386 with higher torque breaking down the cell wall structure more efficiently and liberating  
387 the cellulose microfibrils (Lee et al., 2009; Lamsal et al., 2010). The drop in glucose  
388 yield observed as in-barrel moisture decreased from 45 to 40% was contrary to the  
389 above mentioned trend and might be due to degradation of sugars. Factors other than  
390 those discussed above might also be at work, including interactions between starch and  
391 lignocellulosic components, competition between them for water, and also residence  
392 time of material inside the extruder barrel.

### 393 3.4.3. *Screw speed*

394 Fig. 5 summarizes glucose yields ( $Y_G$ ) from thermo-mechanical pretreatment  
395 experiments at screw speeds ranging from 280 to 420 rpm, while barrel temperature and  
396 in-barrel moisture were constant at 80°C and 40% wb, respectively. Specific mechanical  
397 energy (1478-1528 kJ/kg) increased with screw speed, which is again a universal trend  
398 observed during extrusion.  $Y_G$  ranged from 0.22 g/g pellet to 0.37g/g pellet, with the

399 maximum value obtained at the intermediate screw speed of 350 rpm. As discussed in  
400 section 3.4.2, in the absence of starch mechanical energy was directed at soybean hulls,  
401 thus improving glucose yield as screw speed increased from 280 to 350 rpm. The  
402 excessive mechanical energy input at the highest screw speed of 420 rpm possibly  
403 resulted in degradation of sugars and reduction in glucose yield. The low residence time  
404 associated with high screw speeds might also be a factor, and could have led to reduced  
405 thermal exposure of the substrate during extrusion and subsequently lower  
406 saccharification.

#### 407 *3.4.4. Ethylene glycol*

408 The use of ethylene glycol (EG) as a processing aid was based on a previous  
409 study involving thermo-mechanical pretreatment of wood (Lee et al., 2009). EG was  
410 used because of its cellulose affinity. Its addition was shown to accelerate the  
411 fibrillation process during extrusion of lignocellulosic materials because it intercalates  
412 between cellulose microfibrils, thus preventing reaggregation of liberated microfibrils  
413 due to strong hydrogen bonds. While Lee et al. (2009) observed 62.4% cellulose to  
414 glucose conversion by extruding Douglas fir with 200% EG, this approach was not  
415 successful in the current study. Soybean hulls could not be processed with either 100%  
416 EG or 1:1 solution of EG: water. Decreasing the EG: water ratio to 1:3, 1:9 and 100%  
417 water allowed extrusion of the soybean hulls and led to successively higher glucose  
418 yield at 0.19, 0.24 and 0.27 g/g pellets, respectively, which corresponded to 48-70%  
419 cellulose to glucose conversion percentage. The low amount of solvent used (50%) and  
420 differences in the extrusion process (counter rotating extrusion was used by Lee et al.,  
421 2009) might be some reasons for the poor performance of EG.

#### 422 *3.5. Comparison of thermo-mechanical pretreatment with other methods*

423 Fig. 6 shows data for cellulose to glucose conversion percentage ( $C_{CG}\%$ ) after  
424 enzymatic hydrolysis of soybean hulls subjected to different pretreatments, including  
425 acid (AH), alkali (ALH) and thermo-mechanical extrusion (EX). Enzymatic hydrolysis  
426 of untreated soybean hulls (SH) was the control, which resulted in glucose yield and  
427 conversion percentage of 0.16 g/g and 40.8%, respectively. The grinding process for  
428 soybean hulls exposed enough of the cellulosic structure to allow substantial  
429 saccharification even without any pretreatment. All three pretreatments led to  
430 improvement in glucose yield as compared to the control (69.6, 128.7 and 132.2%  
431 increase for AH, ALH and EX, respectively) although the mechanisms by which it was  
432 brought about differed greatly.

433 After enzymatic hydrolysis of AH and ALH pretreated soybean hulls, glucose  
434 yield was 0.27 and 0.36 g/g, respectively, which corresponded to 69.2 and 93.3%  
435 cellulose conversion. Dilute sulfuric acid pretreatment improved the efficiency of  
436 enzymatic hydrolysis by solubilization and removal of hemicellulose and increasing  
437 cellulose accessibility (Torget et al., 1990). Alkaline pretreatment promotes cellulose  
438 conversion by delignification due to the degradation of ester bonds and cleavage of  
439 glycosidic linkages in the cell wall matrix, which lead to the alteration of the structure  
440 of lignin and reduction of the lignin-hemicellulose complex (Cheng et al., 2010). This  
441 might also be accompanied by swelling of cellulose and its partial decrystallization. The  
442 cellulose conversion from AH and ALH treatments in this study were comparable with  
443 results reported previously. Corredor et al. (2009) observed a maximum cellulose to  
444 glucose conversion of 73% for soybean hulls by applying the combination of 2%  $H_2SO_4$   
445 at 140°C, followed by steam explosion and enzymatic hydrolysis for 36 h. Martin et al.  
446 (2007) pretreated rice hulls with 2% of  $H_2SO_4$  solution at a solid to liquid ratio of 1:10  
447 at 121°C, resulting in 61.4% cellulose conversion after saccharification. Saha et al. (2005)

448 showed that fermentable sugar yields from rice hulls varied, depending on the acid  
449 solution concentration, reaction time and temperature. Less fermentation inhibitors were  
450 observed at lower reaction temperature, and the maximum monomeric sugar yield of 60%  
451 was obtained by treatment with 1% H<sub>2</sub>SO<sub>4</sub> at 121°C. Glucose yield from coastal  
452 Bermuda grass after alkali hydrolysis using 0.75% NaOH at 121°C reached up to 90%  
453 with 86% of lignin removal (Wang et al., 2010).

454 Thermo-mechanical pretreatment (EX) was comparable or better than the more  
455 established chemical methods (AH and ALH), with regard to saccharification efficiency.  
456 Enzymatic hydrolysis of EX pretreated soybean hulls (no starch, 40% moisture, 350  
457 rpm) led to glucose yield of 0.37 g/g and conversion of 94.8%. The primary mechanism  
458 was disruption of cell wall structure due to a combination of mechanical and thermal  
459 energy. The conversion percentage for EX was even better than that reported by some  
460 recent studies involving extrusion. Our lab group previously reported reducing sugar  
461 yields of 60-73% and 25-36% for wheat bran and soybean hulls, respectively, after  
462 pretreatment with twin screw extrusion followed by saccharification (Lamsal et al.,  
463 2010). Karuppuchamy and Muthukumarappan (2009) and Karunanithy and  
464 Muthukumarappan (2010a; 2010b) used a single screw extruder for pretreatment of  
465 soybean hulls, corn stover, switchgrass and prairie cord grass followed by enzymatic  
466 hydrolysis, resulting in 62, 61, 45 and 66% sugar recovery, respectively. These authors  
467 also reported absence of fermentation inhibitors such as hydroxymethylfurfural, which are  
468 often produced during chemical pretreatment (Torget et al., 1990; Saha et al., 2005).

469

#### 470 **4. Conclusions**

471 Thermo-mechanical extrusion was shown to be a feasible pretreatment method  
472 for lignocellulosic ethanol production. The challenge of processing lignocellulosic

473 substrates with poor flow properties can be overcome by utilizing high in-barrel  
474 moistures and/ or processing aids such as starch. Cellulose conversion from extrusion  
475 pretreatment of soybean hulls was comparable or better than that obtained from  
476 traditional chemical pretreatments utilizing acid and alkali. The enzyme loading used in  
477 this study was much lower than in most previous studies, yet higher glucose yields were  
478 obtained. This continuous pretreatment technology shows great promise, especially  
479 since it can be scaled up easily to obtain high throughputs. Absence of fermentation  
480 inhibitors is another advantage over traditional pretreatment methods.

481

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490

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620  
621  
622

623 **Figure captions**

624 Fig. 1. Schematic of lab-scale twin extruder screw profile and barrel temperature

625 settings.

626 Fig. 2. Crystallinity of soybean hulls before and after pretreatment<sup>a</sup>.

627 Fig. 3. Reducing sugar yield (g/g pellet) as determined by DNS assay<sup>a,b</sup>.

628 Fig. 4. Effect of in-barrel moisture content and starch addition on glucose concentration

629 (g/g pellet)<sup>a,b</sup>.

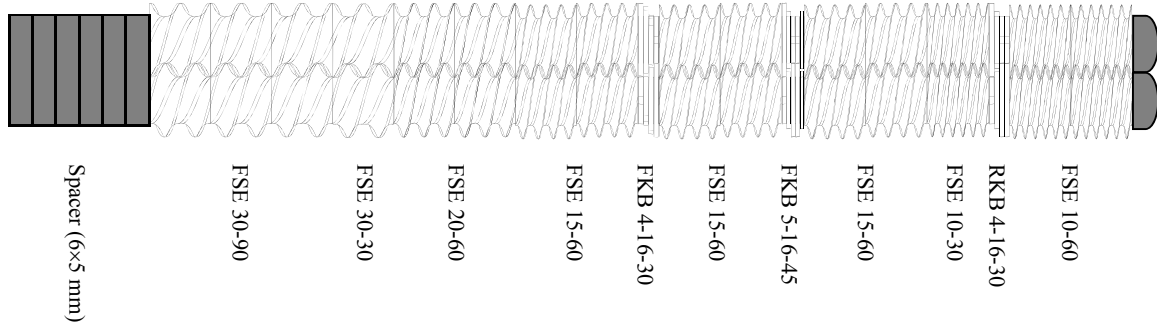
630 Fig. 5. The effect of screw speed on glucose concentration (g/g pellet)<sup>a,b</sup>.

631 Fig. 6. Cellulose to glucose conversion (%) after saccharification of soybean hulls

632 subjected to different pretreatments<sup>a</sup>.

Head No.	1	2	3	4	5	6
Barrel temperature						
Temp.(°C)	40	50	60	70	75	80
	40	50	65	80	90	110
	40	60	80	100	120	140

633



FSE: forward conveying screw element (all double flight, intermeshing)

FKB: forward kneading block; RKB: reverse kneading block

Numbers on screw elements: pitch (mm)-element length (mm)

Numbers on kneading blocks: number of disks-total block length (mm)-staggering angle of disks

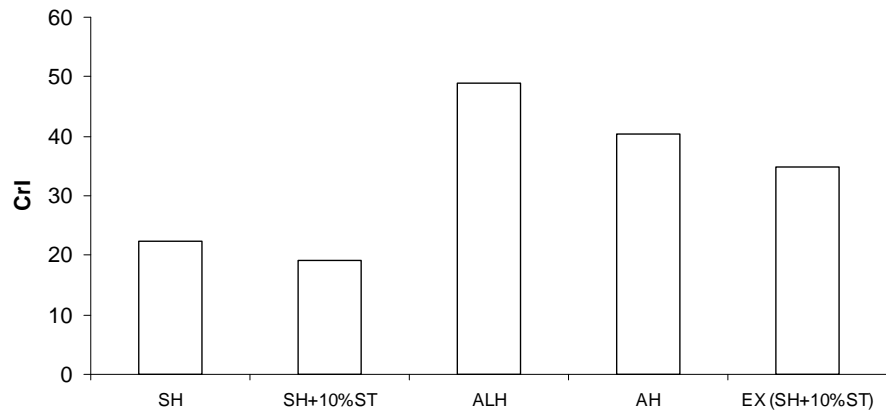
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<sup>a</sup>SH: untreated soybean hull; SH+10%ST:untreated soybean hull + 10% starch; ALH: alkali

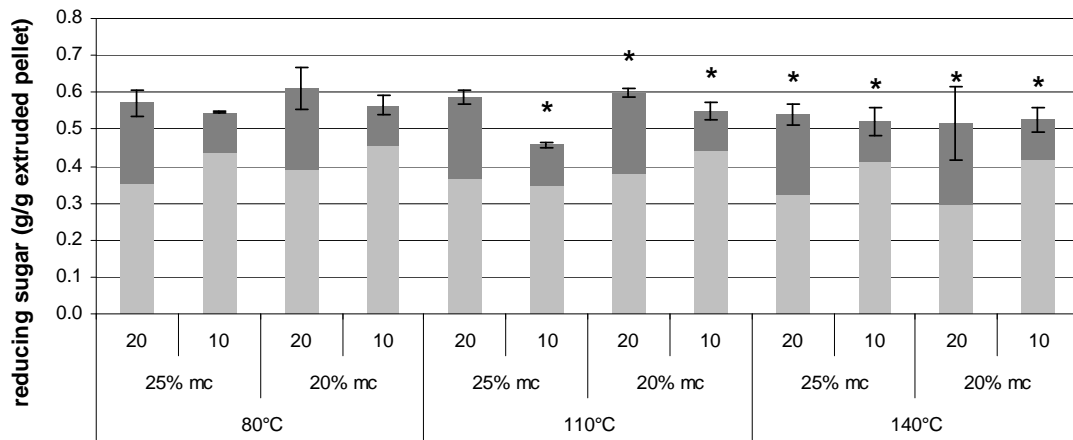
642

pretreated soybean hull; AH: dilute acid pretreated soybean hull; EX: extruded pellet from

643

soybean hull + 10% starch.

644



645

646

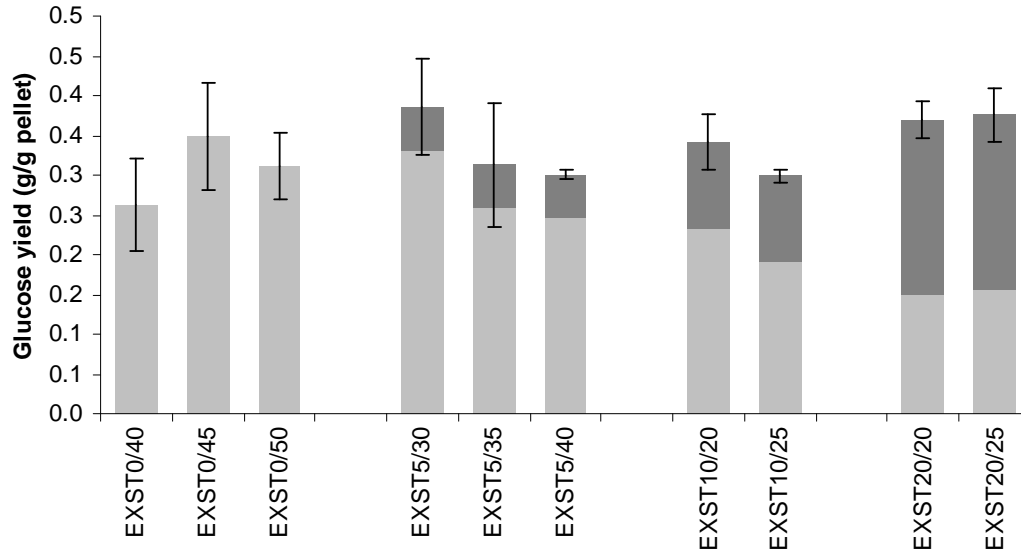
647 <sup>a</sup>Dark shaded portion of each bar indicates starch contribution assuming 100% hydrolysis. Error bars

648 represent standard deviation.

649 <sup>b</sup>X-axis from top to bottom: starch addition %, in-barrel moisture content %, maximum barrel

650 temperature.

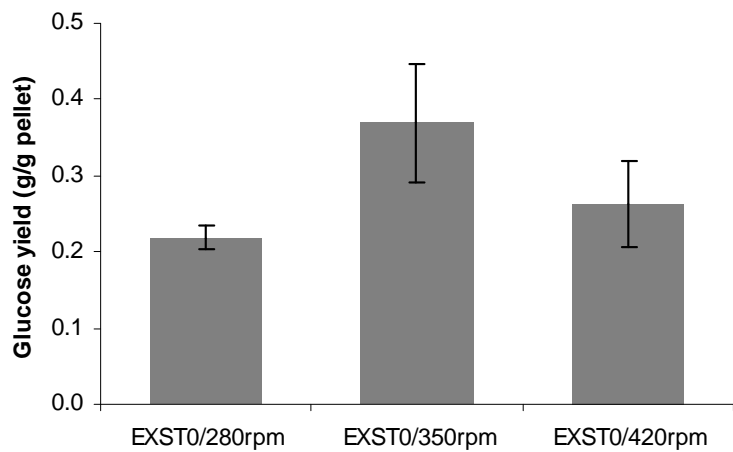
651 \* Surging and/ or burning observed.



652  
653

654 <sup>a</sup>Screw speed 420 rpm and barrel temperature 80°C. Dark shaded portion of each bar indicates  
655 starch contribution assuming 100% hydrolysis. Error bars represent standard deviation.

656 <sup>b</sup>X-axis: EXSTx/y; x = % starch addition, y = in-barrel moisture (%wb).

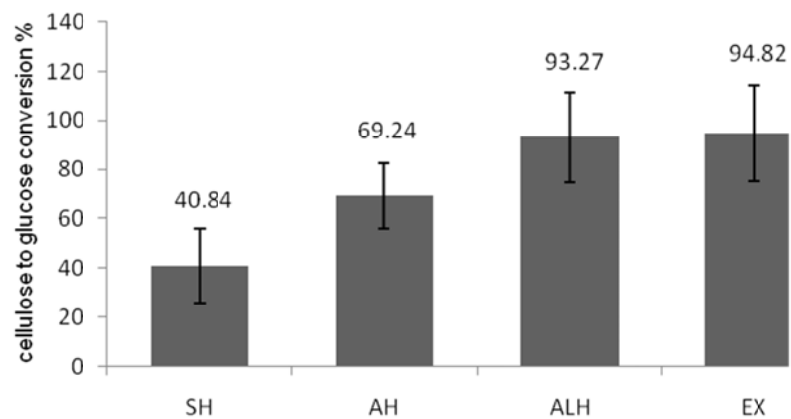


657

658 <sup>a</sup>No starch, barrel temperature 80°C and in-barrel moisture 40% wb. Error bars represent standard  
659 deviation.

660 <sup>b</sup>X-axis: EXST0/screw speed.

661



663

666

667

668

667

<sup>a</sup>SH: untreated soybean hull; AH: dilute acid pretreated soybean hull; ALH: alkali pretreated soybean hull; EX: extruded soybean hull (no starch, screw speed 350 rpm and in-barrel moisture 40% wb). Error bars represent standard deviation.



667 **Table 1. Experimental design for enzyme cocktail optimization (response**  
 668 **surface methodology).**

669

		mL/g cellulose		
		-1	0	1
X	Cellulase (Celluclast 1.5L)	0.0254	0.0609	0.1016
Y	$\beta$ -glucosidase (Novozyme 188)	0.0254	0.0609	0.1016
Z	Cell wall degrading enzyme (Viscozyme® L)	0.0254	0.0609	0.1016

670

Test no.	X	Y	Z
1	-1	-1	0
2	1	-1	0
3	-1	1	0
4	1	1	0
5	-1	0	-1
6	1	0	-1
7	-1	0	1
8	1	0	1
9	0	-1	-1
10	0	1	-1
11	0	-1	1
12	0	1	1
13	0	0	0
14	0	0	0
15	0	0	0

671

672

673 **Table 2 Lignocellulosic composition (% db) of soybean hull (SH) before and after**  
674 **pretreatment.**

675

	Lignin	Hemicellulose	Cellulose
Untreated SH	2.33±0.05	17.21±0.06	35.35±0.20
Acid hydrolyzed SH	5.22±0.11	7.80±0.14	61.79±0.43
Akali treated SH	11.66±0.21	14.59±0.12	66.72±0.48
(SH + 0% starch) extrudate	1.52±0.01	20.42±0.22	36.88±0.05

676

677